

# The Global Race for A Better Battery

*Y. Shirley Meng*

The University of Chicago

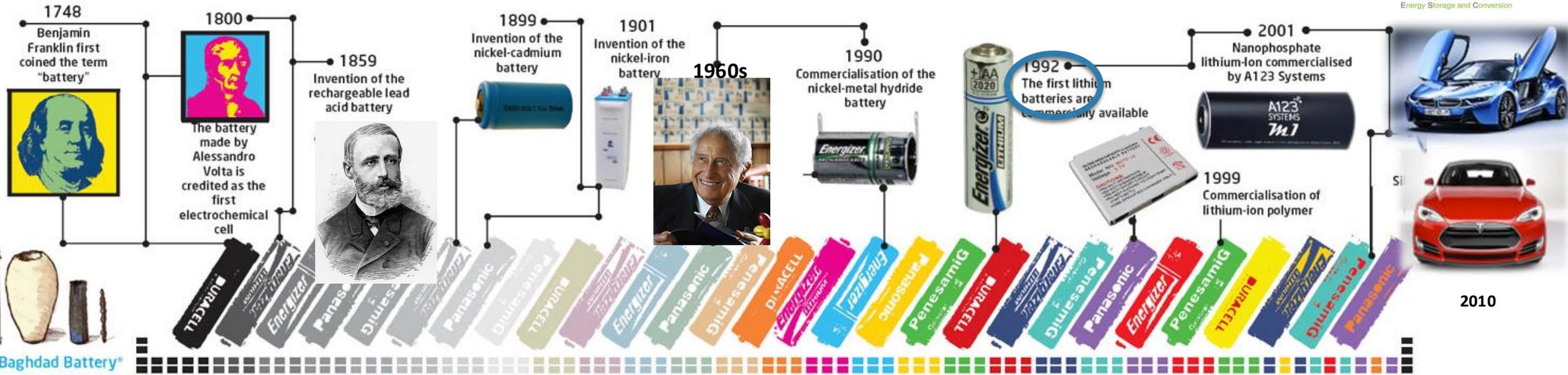
Argonne Collaborative Center for Energy Storage Science

Energy Storage Research Alliance

University of Maryland, College Park

Oct. 23, 2025

# A Brief History of Battery and Industrialization



**Baghdad Battery\***  
 \*There is a possibility that the battery was invented twice. Discovered by German archaeologist Wilhelm König on the outskirts of Baghdad, terracotta jars with a copper sheet inlay and an iron rod. These two combine to form an electrochemical couple in an electrolyte, the building blocks of a battery. The jars are believed to be 2000 years old.



**Intercalation Chemistry – Nobel Chemistry 2019 !**

IoT  
mWh

Robots  
Wh

Drones  
kWh

Aviation / SemiTruck  
MWh



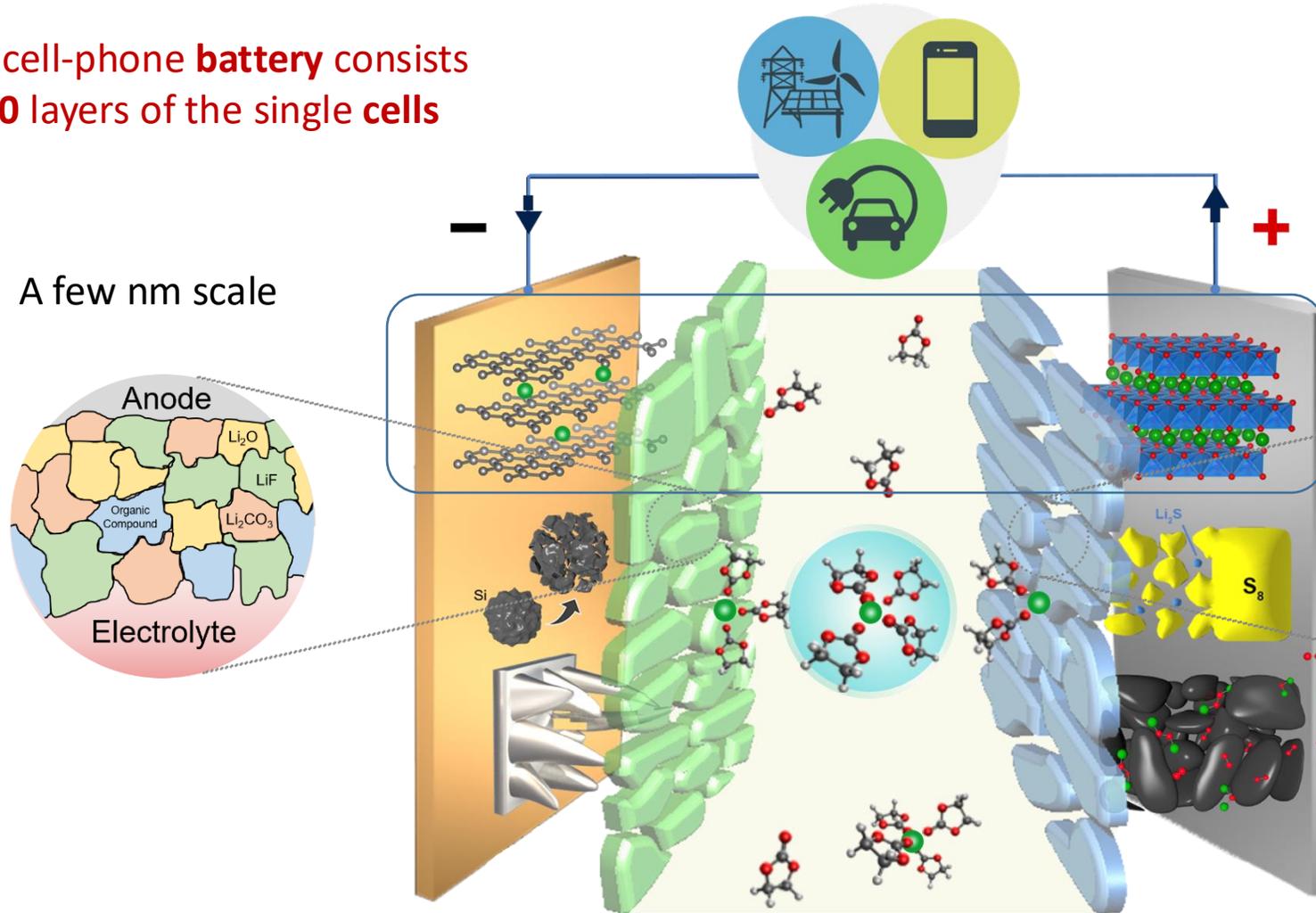
**1Whr is a large energy unit = 3600 W.s (J)**

**Energy Storage for Renewables  
GWh**

# Rechargeable Battery - A Complex Engineering System

Kang Xu, Venkat Srinivasan and Y. Shirley Meng, Science 2023

One cell-phone **battery** consists **10-20** layers of the single **cells**



Moving Ions  
(Chemical Bond)

Dynamic Phenomena

Strain - Fatigue

Thermodynamically Closed  
System 99.9% efficiency  
needed

**SEI** – Life and Safety  
Differentiators

# World Production of LIB >1TWh/Year 2024 (Today)

The Top 10

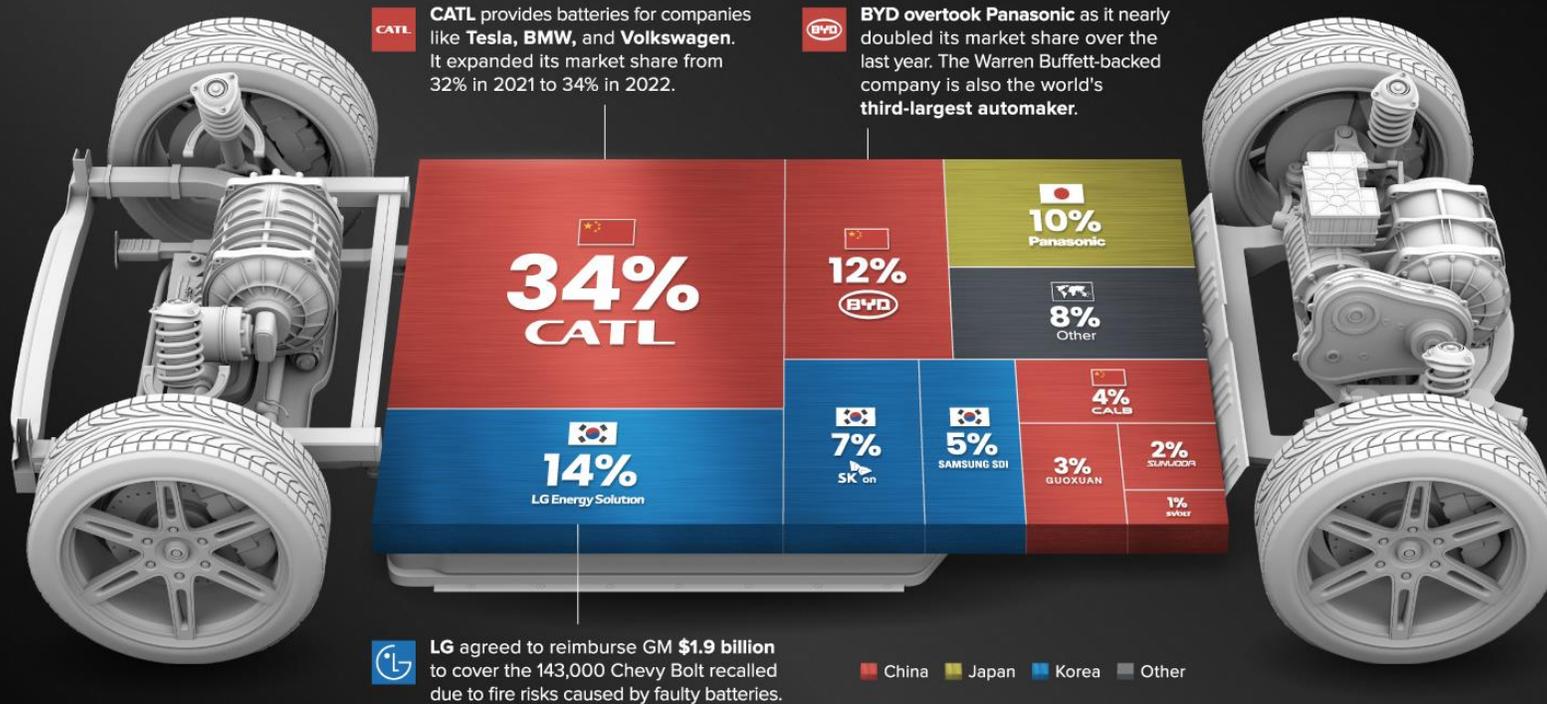
## EV BATTERY MANUFACTURERS

in 2022

E

The EV battery market is expected to grow from \$17 billion in 2019 to \$95 billion by 2028.

Here are the world's biggest battery manufacturers in 2022.



Source: SNE Research via Bloomberg

**In 1998**, Academician Chen Liquan, built the first production line using a complete set of Chinese equipment.

**In 1999**, South Korea entered the lithium-ion battery market, and LG Chem completed South Korea's first battery product.

**In 2000**, BYD won an order from Motorola

**In 2001**, China entered WTO

**In 2004**, CATL became an iPod supplier. China's lithium battery industry emerged. China's annual output of lithium-ion batteries is 800 million units, accounting for 38% of the global share, second only to Japan.

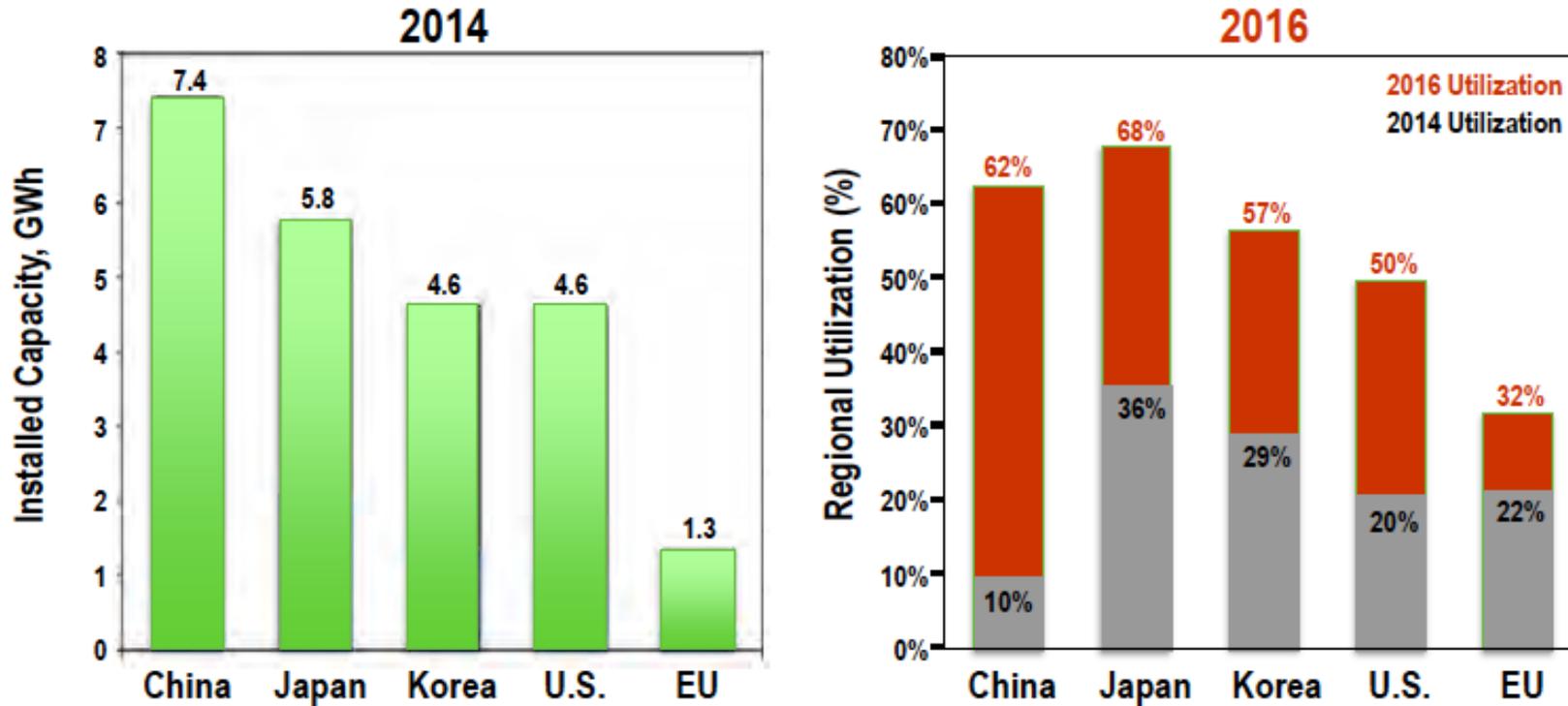
**Start in 2002** "Electric Vehicle Key Project"  
 - 863 National High-Tech Development Plan  
 - Spent **880 million yuan** for EV research in five years

**Start in 2013** China Central government subsidies

- ▶ 60,000 yuan for passenger electric cars (>250 km)
- ▶ 300,000 yuan for electric buses
- ▶ Exemption from vehicle purchase tax
- ▶ Local subsidies: up to 60% of EV retail price
- ▶ Non-subsidy measures: license-plate restrictions

# 2014 We All Started from The Same Spot! (Past)

## Regional Automotive LIB Cell Capacity and Utilization



- Automotive lithium-ion battery demand growing but short of global manufacturing capacity.
- Utilization of U.S. plants increased from 20% in 2014 to ~50% in 2016.
- Forecasted compound annual growth rates in lithium-ion demand: 22%–41% (through 2020).

# A Vision for the Future (World View)

ONLY 2% done with what we need , that is 200-300TWh batteries !

- 8.5 Billion People by 2030 – More than half in developing countries  
each person has about 10kWh battery – we will need 85TWh battery

- 6 Billion Smartphones/Smart Tablets – All connected via 5G

This is a small market now – but people with real 5G need fast charging/fast discharging

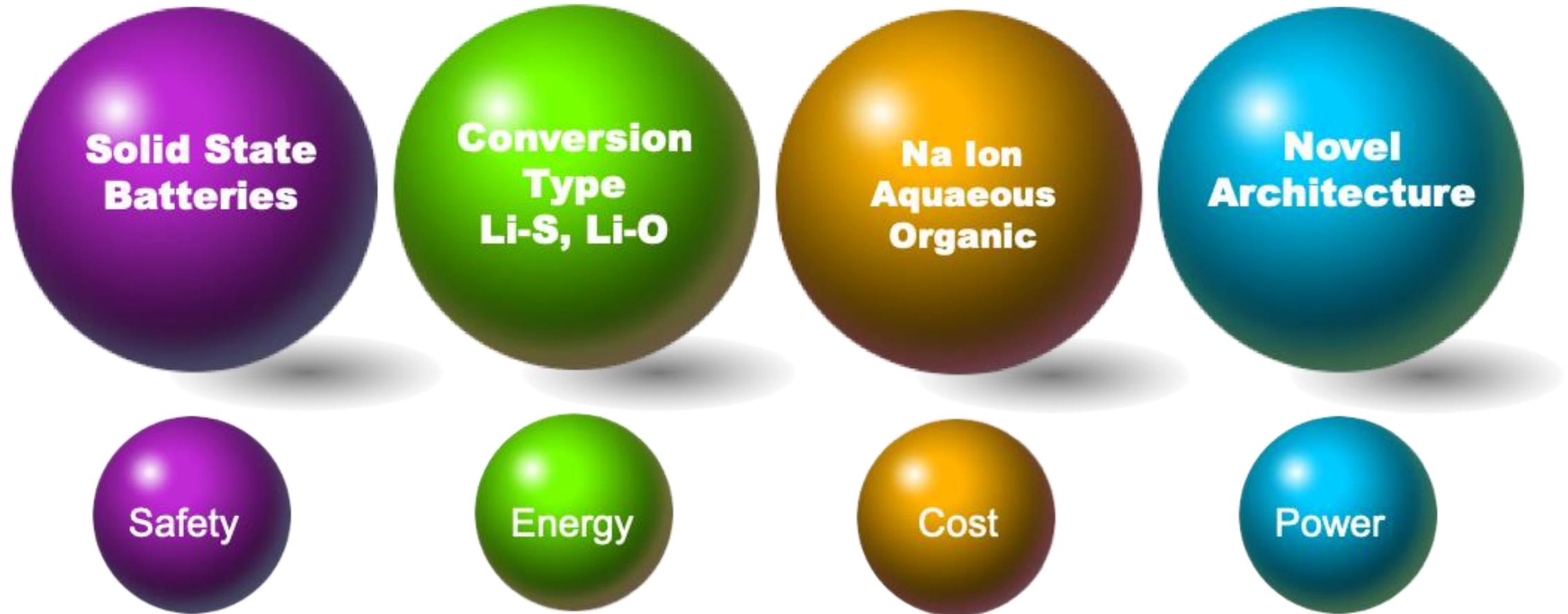
- 1 Billion Cars to be Electrified – Possibly Electric Planes

We predict that total number of cars will reduce, cars will be better utilized and we will have electric heavy duty trucks and possible V2G enabled – 100TWh

- Grid of the Future

Long duration needs can be fulfilled by technologies other than batteries (H<sub>2</sub>+Hydro+Thermal) – 100TWh

# Next Decade of Energy Storage and Battery Technology



**Solid State Batteries**

**Conversion Type  
Li-S, Li-O**

**Na Ion  
Aqueous  
Organic**

**Novel  
Architecture**

**Safety**

**Energy**

**Cost**

**Power**

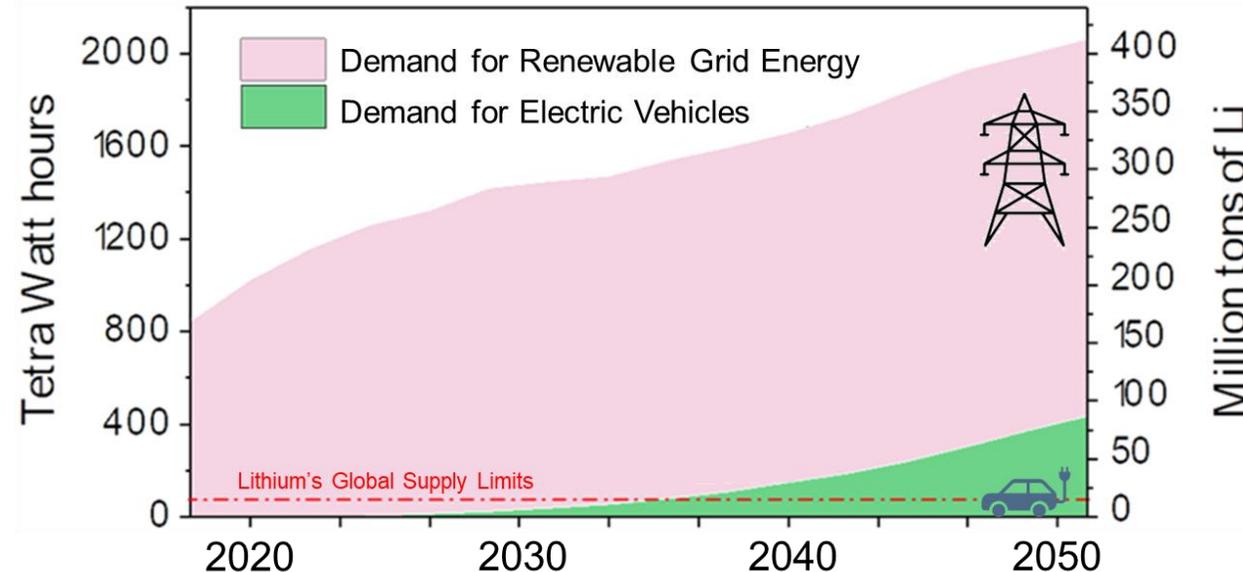
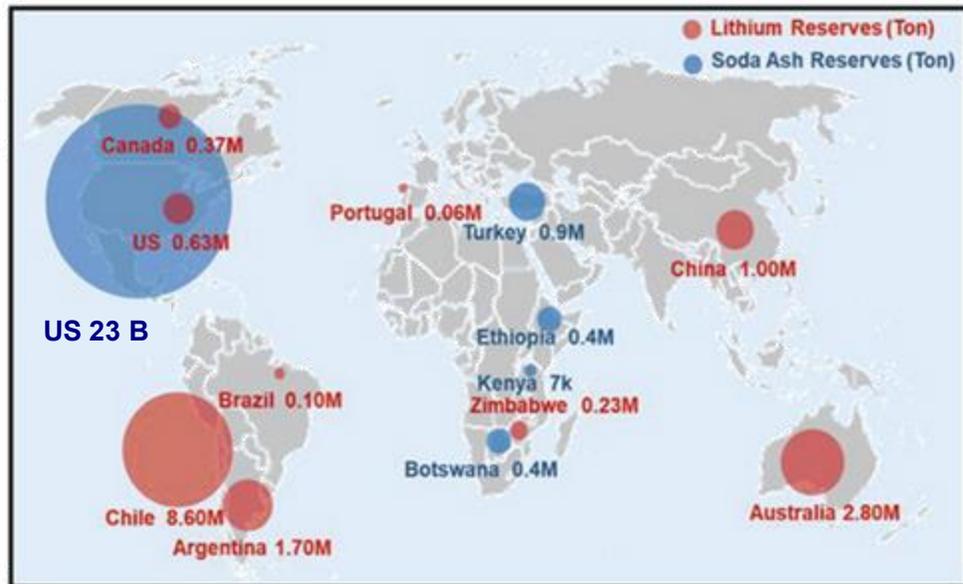
High energy batteries that never catch fire

Batteries last more than 30 years

Batteries can be 100% recycled (supply chain)

Batteries that can be charged full in 5 minutes

# Na vs Li Materials Sustainability



## US Energy Storage : Scale

### 1) Electrical Grid

- Per capita - 12000 kWh / year
- 400 tera-watt hours if just 10% storage  
→ 60 million tons of Li Needed

### 2) Vehicles

- 17 million / year, if all EVs
- 0.85 tera-watt hours  
→ 0.13 million tons of Li Needed

**Abundant Sodium in the U.S. is needed to accelerate energy transition**

**Na battery will reduce the demand for Li, Cu, Co and possibly Ni**

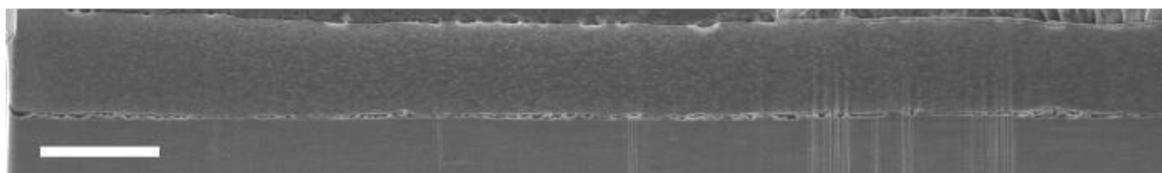
**Na battery can be shipped more safely**

**Na battery can allow more customer choices**



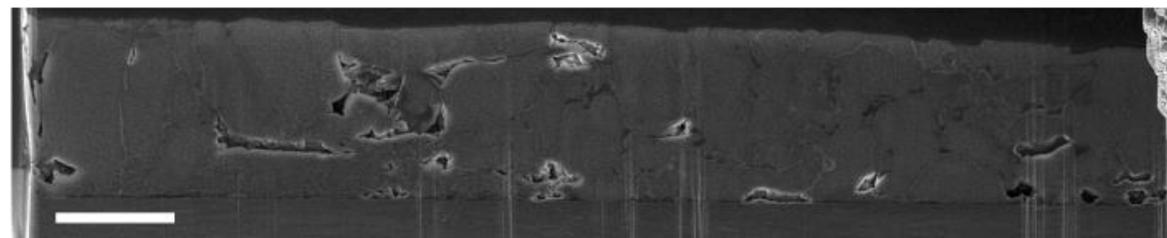
# Analysis of Na Metal Deposition and Stripping

(a) **1 M NaPF<sub>6</sub> in DME**



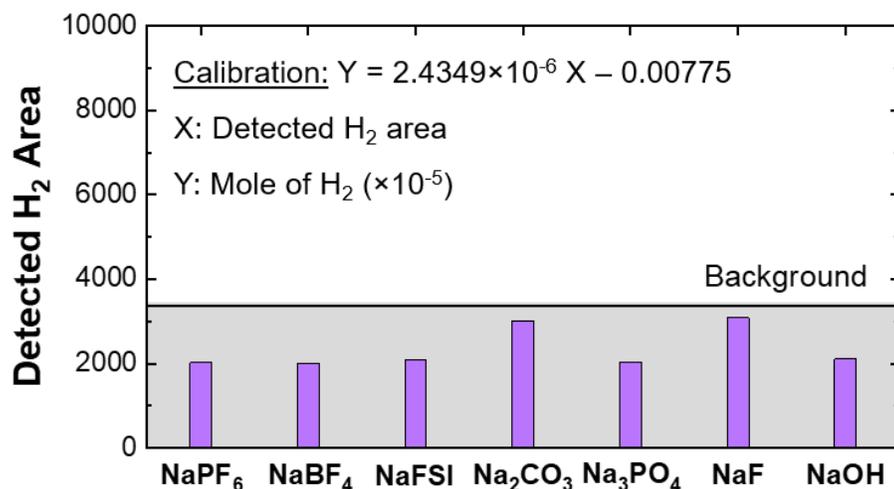
**180 kPa**

(b) **1 M NaPF<sub>6</sub> in EC:DMC (1:1)**



**250 kPa**

The sodium was plated at 0.5 mA/cm<sup>2</sup> for 1 mAh/cm<sup>2</sup> on Al foil. The images are acquired under 5kV voltage and 0.2 nA current using a TLD detector. The scale bars are 10 μm.



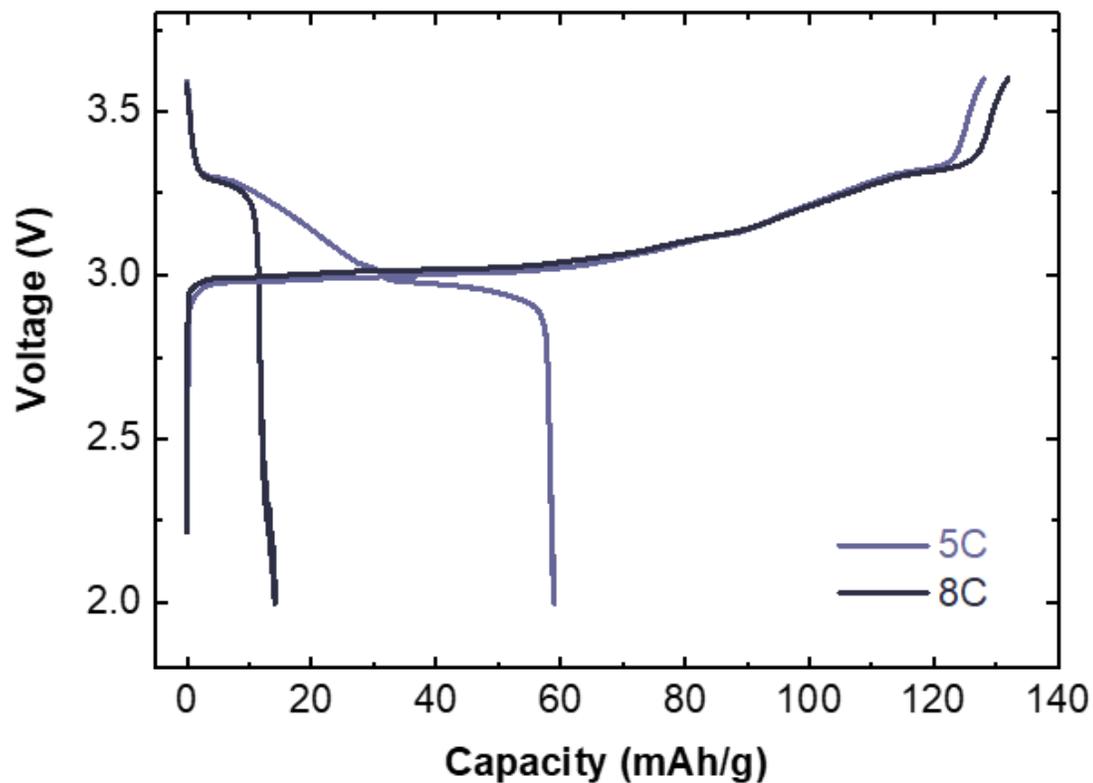
Standard	Detected H <sub>2</sub> Area
NaPF <sub>6</sub>	2025
NaBF <sub>4</sub>	2004
NaFSI	2100
Na <sub>2</sub> CO <sub>3</sub>	3013
Na <sub>3</sub> PO <sub>4</sub>	2037
NaF	3127
NaOH	2115

The controlled TGC experiment on standard commercial powders showed no hydrogen generation. This test was performed using ethanol as the solvent.

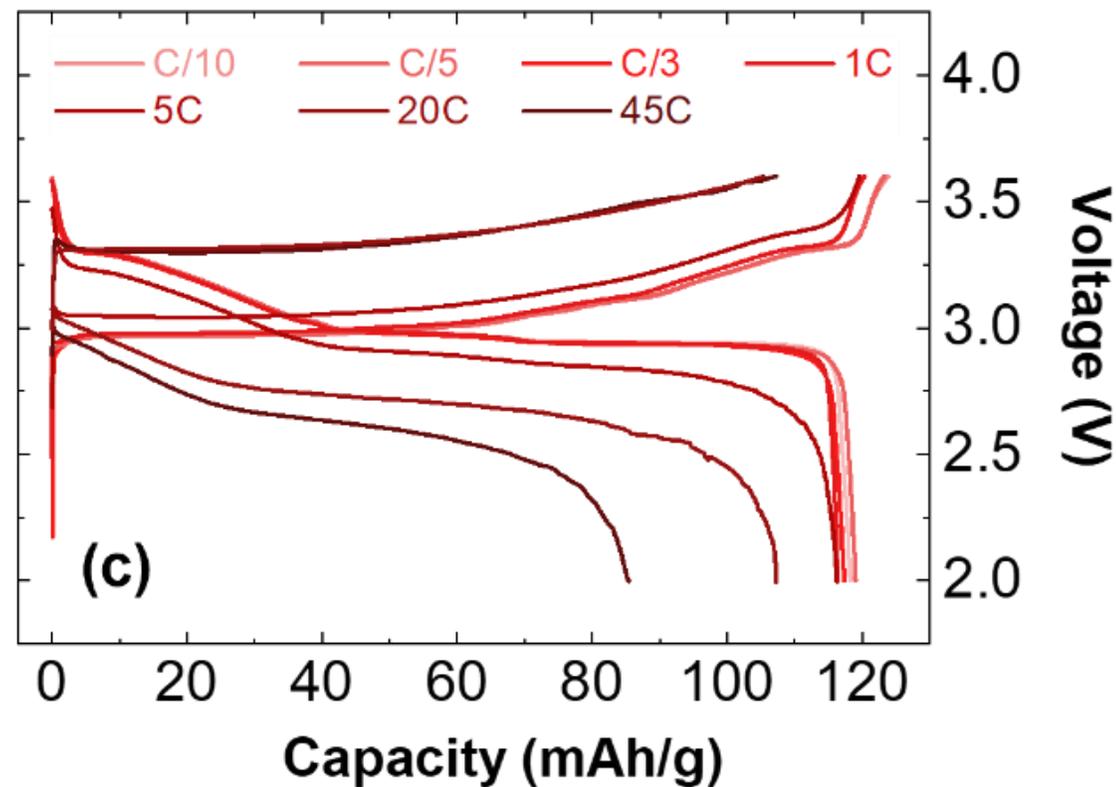
# Power of Sodium Battery

1C = 1 hour charging/ 20C = 3 min charging!

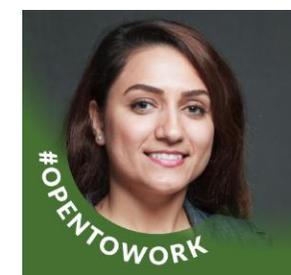
Na Metal as anode, NaCrO<sub>2</sub> as the cathode. The cells have controlled 100% excess of sodium inventory.



1M NaPF<sub>6</sub> in EC:DMC (1:1)



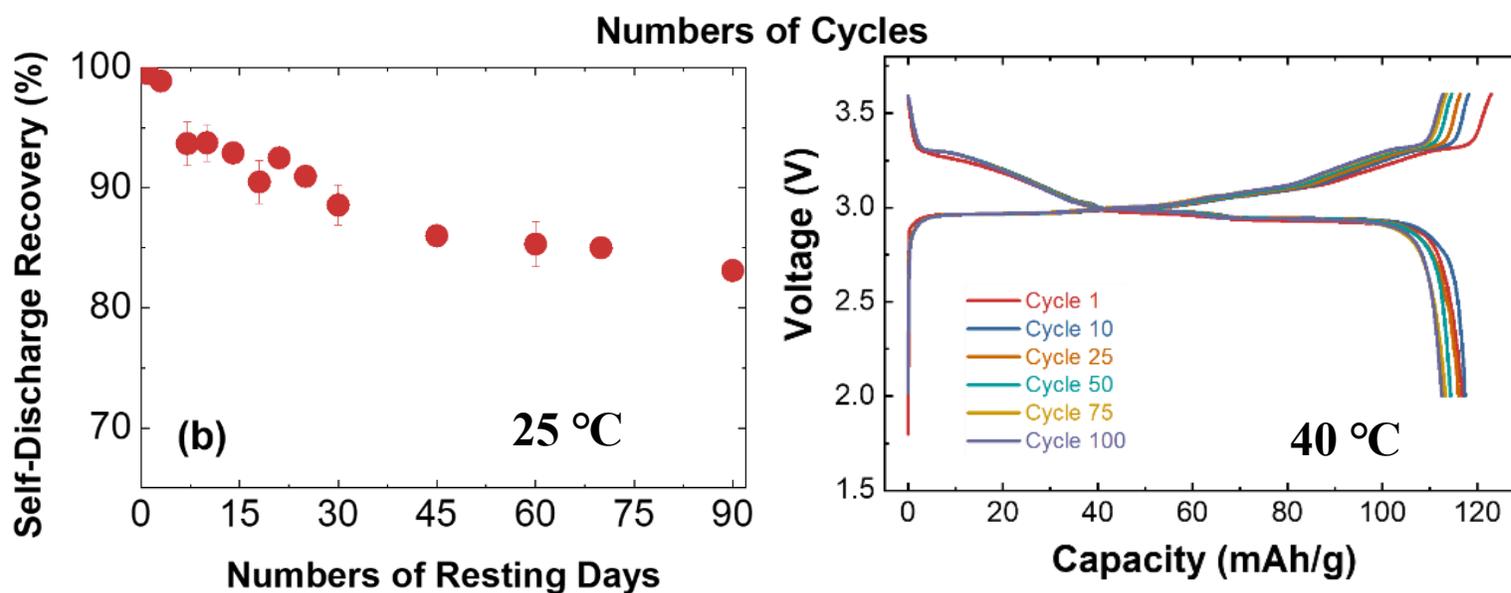
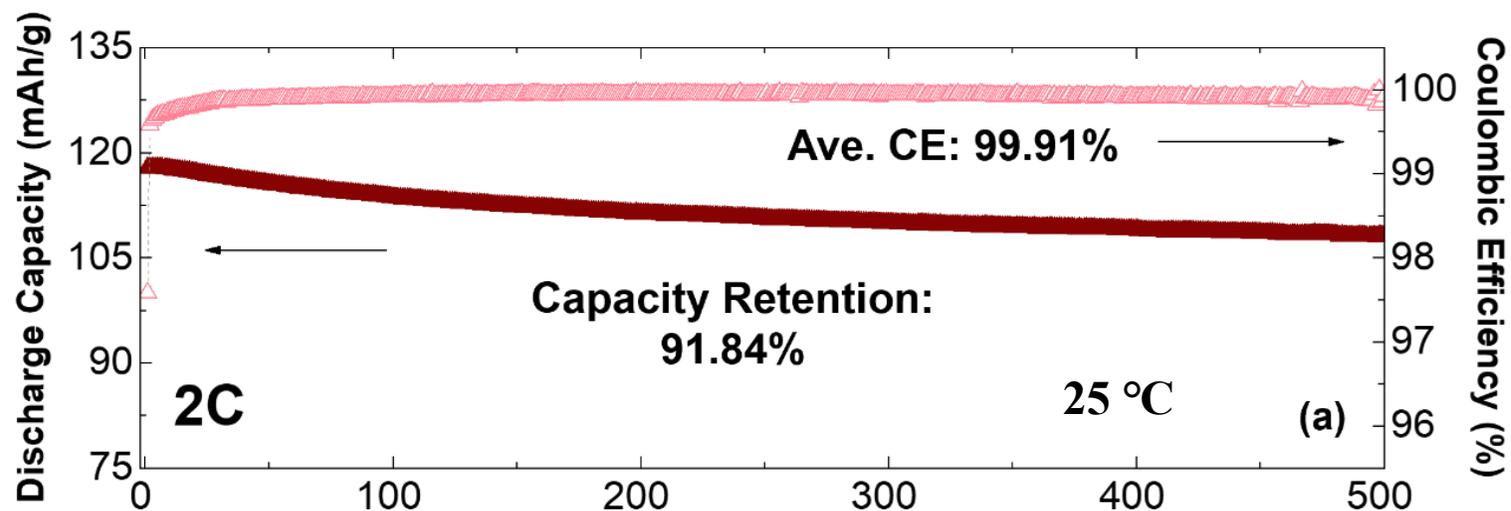
1M NaPF<sub>6</sub> in DME



Dr. Baharak Sayahpour  
(ASML)

We made the  
anode, electrolyte,  
cathode

Active work since  
my CAREER award  
in 2010



National  
Science  
Foundation

# UNIGRID's Value: Energy & Safety

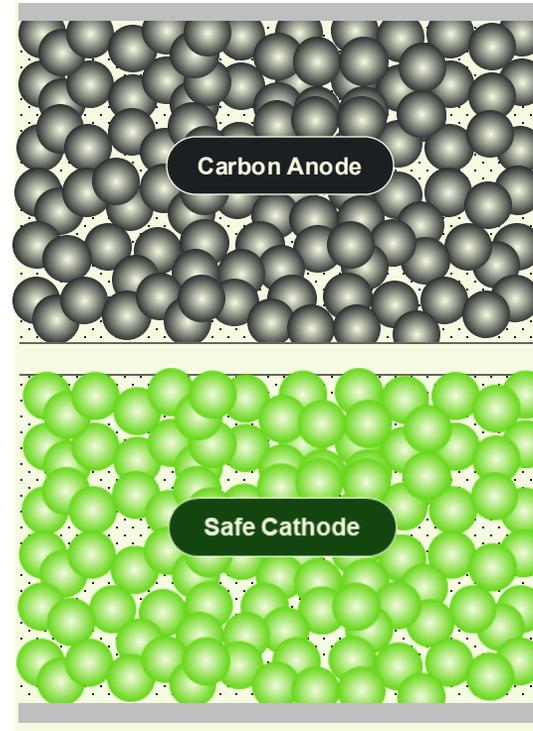


Dr. Darren Tan

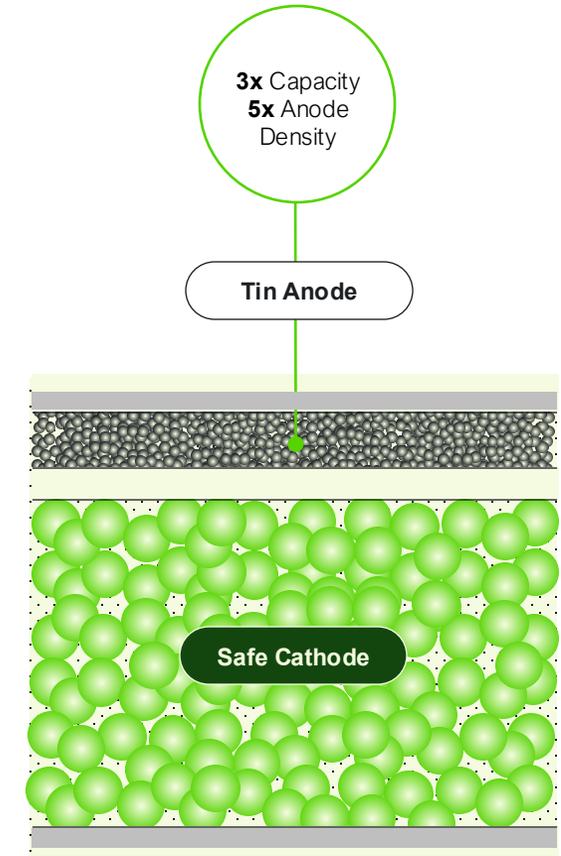
Sodium Ion Battery Landscape	Key IP	Cathode	Anode	Wh/kg	Wh/L	Cycles	Market	
<b>UNIGRID</b>		Anode	Any	Alloy	180 Wh/kg	450 Wh/L	>3000	EV / ESS
		Cathode	Layered Oxide	Carbon	160 Wh/kg	200 to 300 Wh/L	>3000	EV / ESS
		Cathode	Layered Oxide / PBAs	Carbon	160 Wh/kg	200 to 300 Wh/L	—	EV / ESS
		Cathode	Symmetric Prussian Blue Analogues	—	20 Wh/kg	18 Wh/L	>50000	Backup Power
		Cathode	Layered Oxide	Carbon	145 Wh/kg	180 to 300 Wh/L	>4500	EV
		Cathode	Prussian Blue Analogues	Carbon	—	180 to 300 Wh/L	—	EV / ESS
		Cathode	Poly-anionic	Carbon	—	180 to 300 Wh/L	—	EV / ESS
		Cathode	Layered Oxide & Poly-anionic	Carbon	145 Wh/kg	180 to 300 Wh/L	>5000	EV / ESS
		Cathode	Layered Oxide	Carbon	145 Wh/kg	180 to 300 Wh/L	>5000	EV / ESS
		Cathode	Poly-anionic	Carbon	110 Wh/kg	180 to 300 Wh/L	>5000	EV / ESS
		Cathode	Layered Oxide / PBAs	Carbon	140 Wh/kg	180 to 300 Wh/L	>4000	EV / ESS
		Separator	Layered Oxide	Carbon	—	—	—	ESS
		Cathode	—	Carbon	140 Wh/kg	180 to 300 Wh/L	>1000	ESS
		Cathode	—	Carbon	160 Wh/kg	180 to 300 Wh/L	>2000	EV / ESS

UNIGRID

# Anode - Energy Innovation



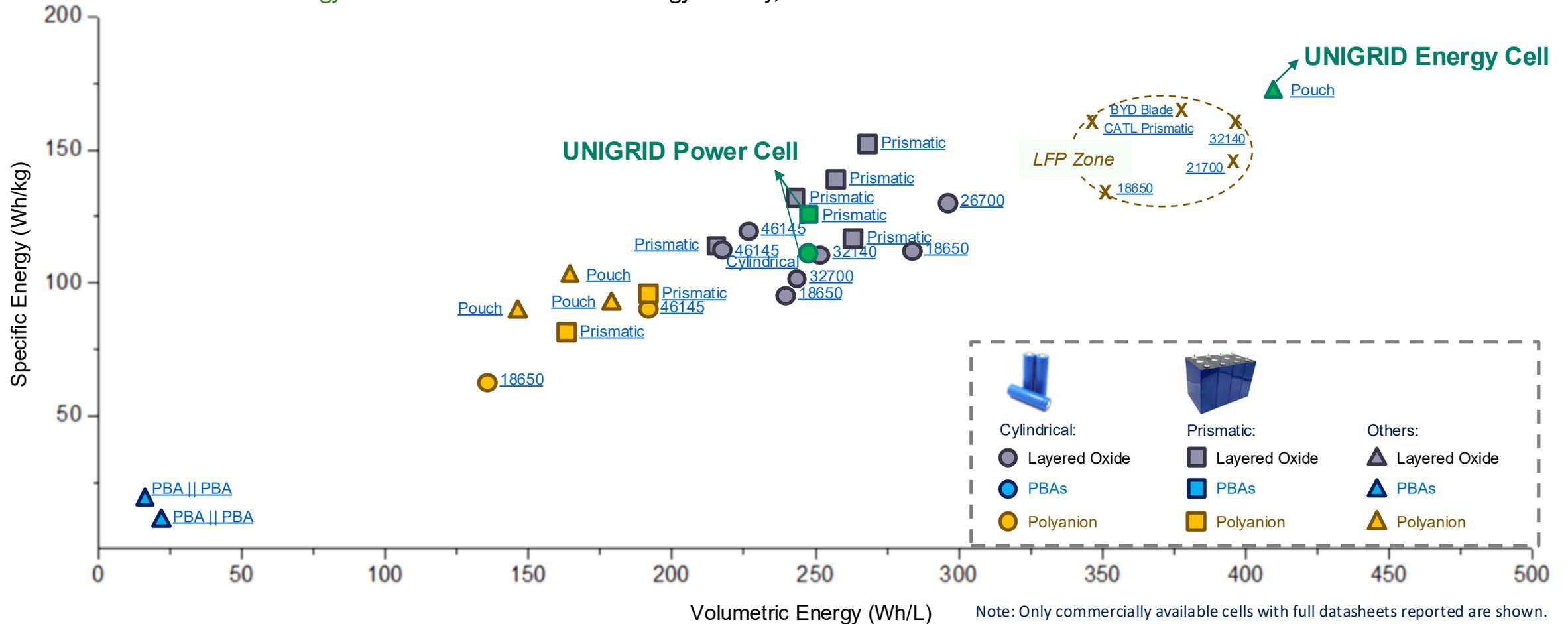
UNIGRID  
Power Cell



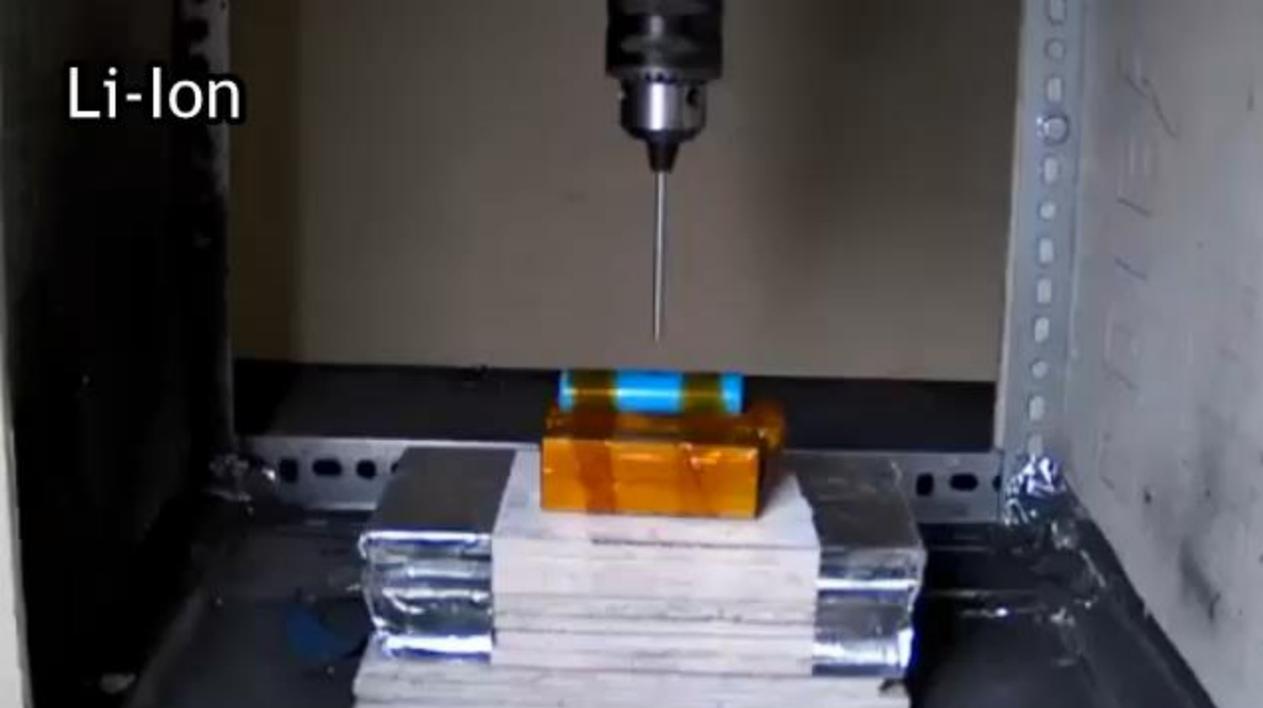
UNIGRID  
Energy Cell

## Sodium-ion - Reported Energy Densities

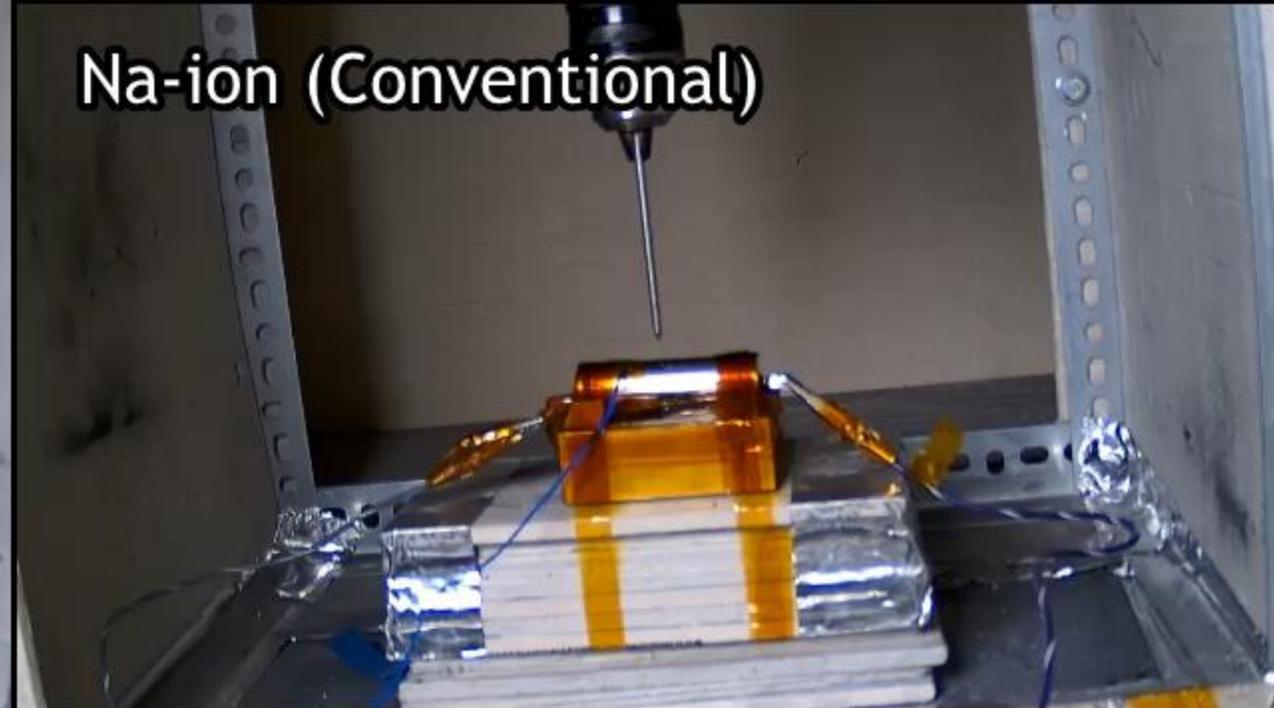
- UNIGRID Energy Cell sets record in Na-ion energy density, as the **Na-cell that exceeds LFP**



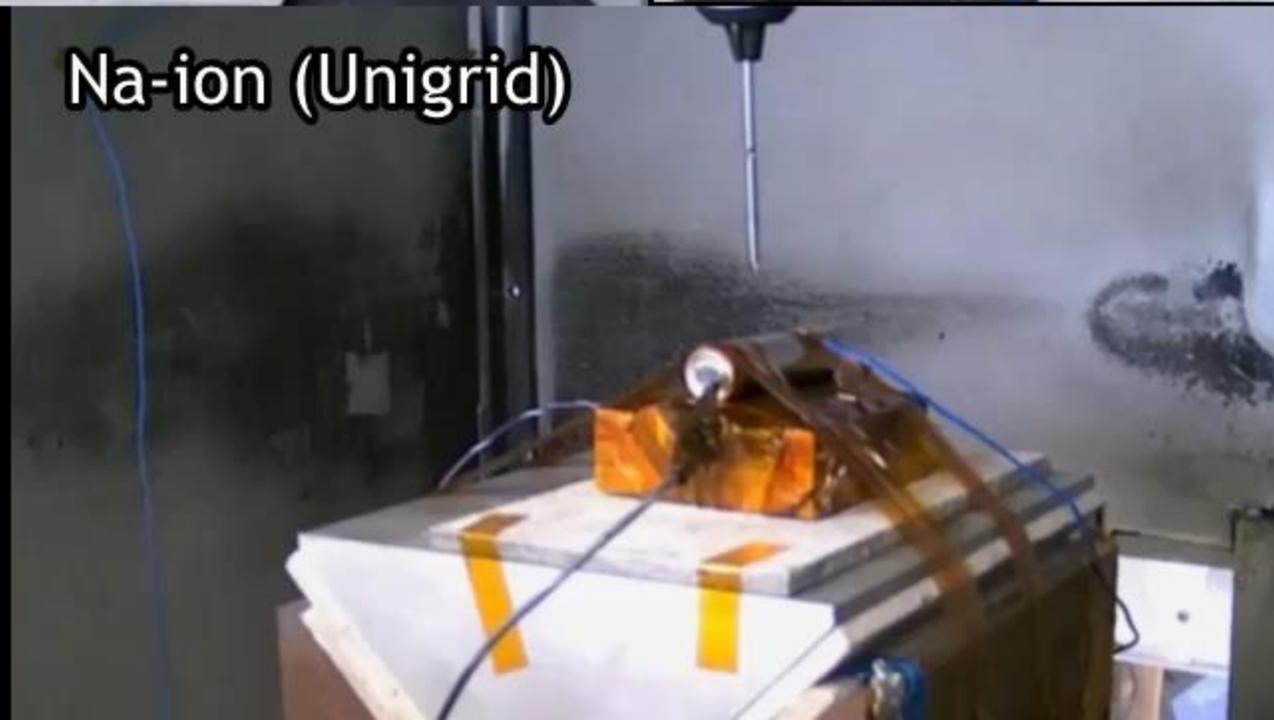
Li-Ion



Na-ion (Conventional)



Na-ion (Unigrid)



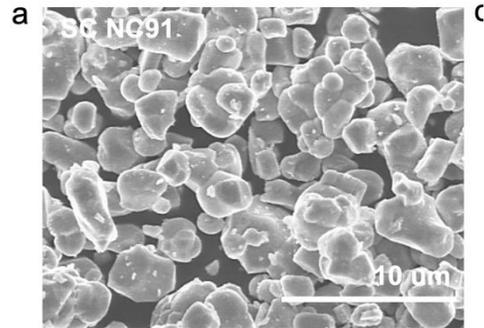
# The Race is NOT Over Yet for Lithium Batteries



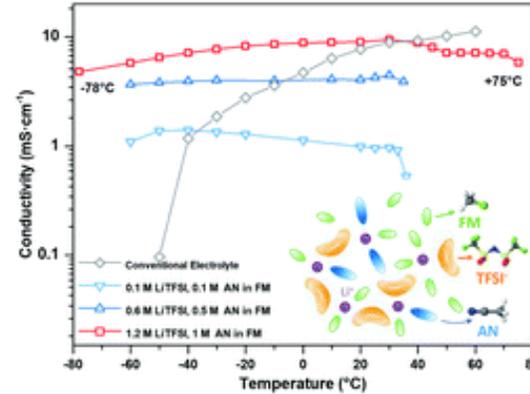
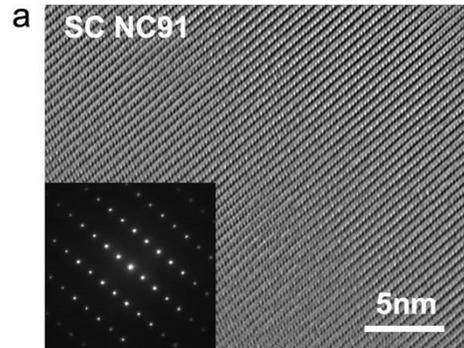
Maxwell Technologies 2017

Tesla Acquisition 2019

All Dry 4680 in Cybertruck 2024



Better Cost Performance Ratio



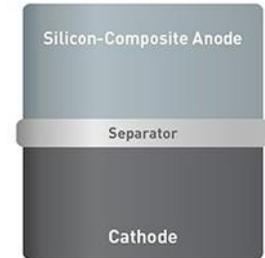
-60 to +60 C  
Wider operation temperatures



**Gen 0**  
**Li-Metal**  
100-200 Wh/kg  
200-300 Wh/L  
Dangerous



**Gen 1**  
**Li-ion**  
200-250 Wh/kg  
600 Wh/L  
Safe



**Gen 2**  
**Li-ion**  
250-300 Wh/kg  
700 Wh/L  
Safe



**Gen 3**  
**Li-Metal**  
400-500 Wh/kg  
1200 Wh/L  
Safest

Dry Battery Electrode  
(DBE) Processing

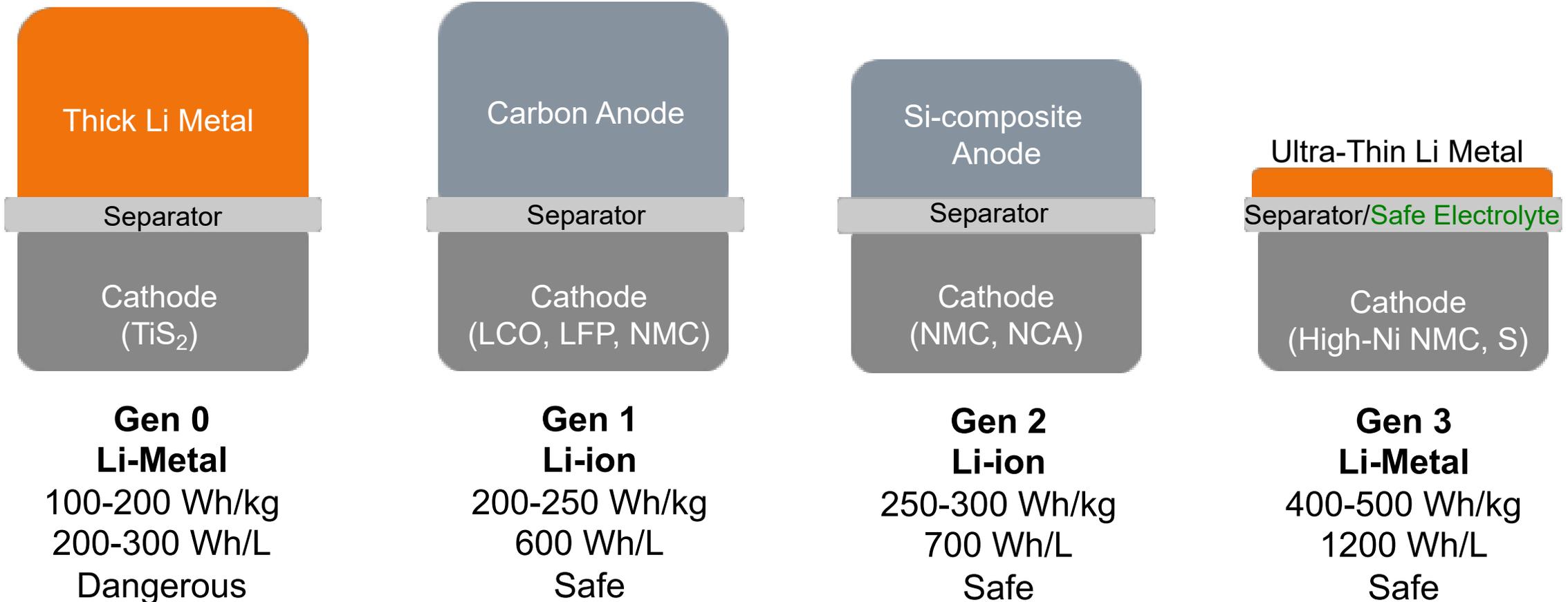
Single Crystal NMC/NCA

Electrolyte Genome

Anode-Free

# Li Metal Batteries and Battery500 Consortium

This will ENABLE 500MILES per charge of EVs



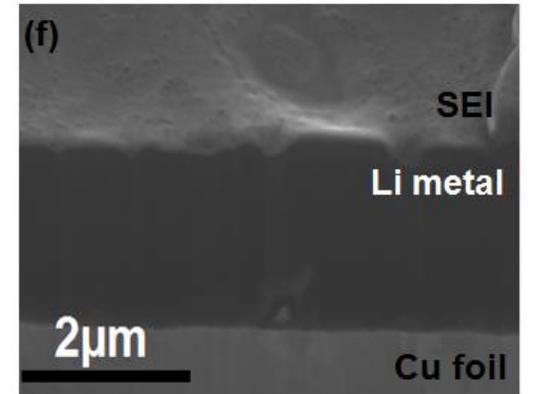
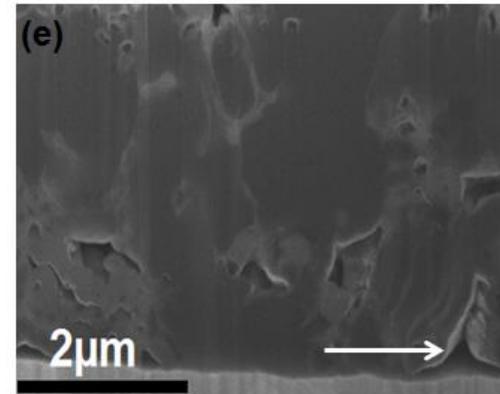
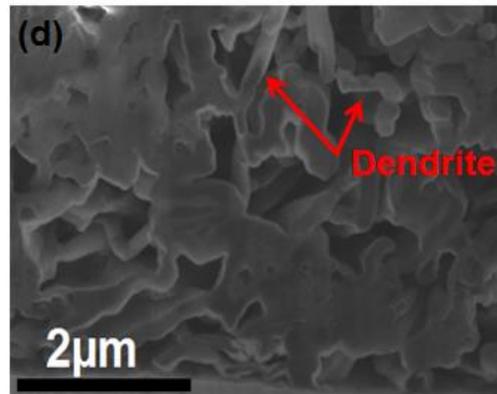
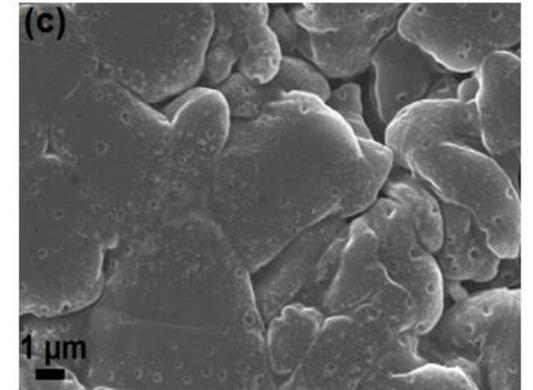
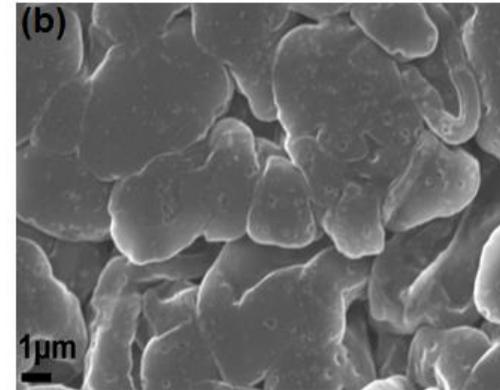
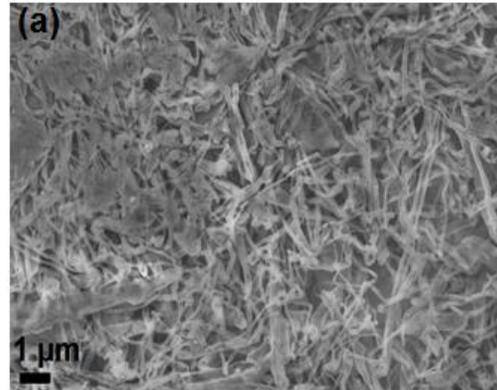
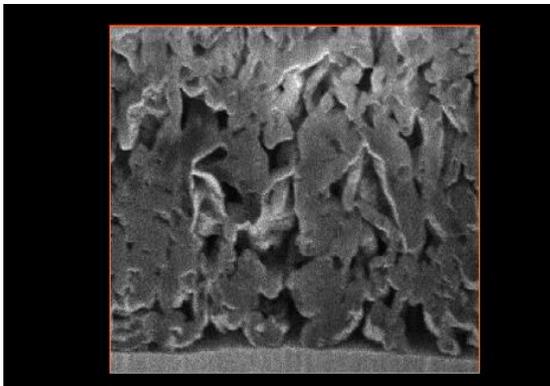
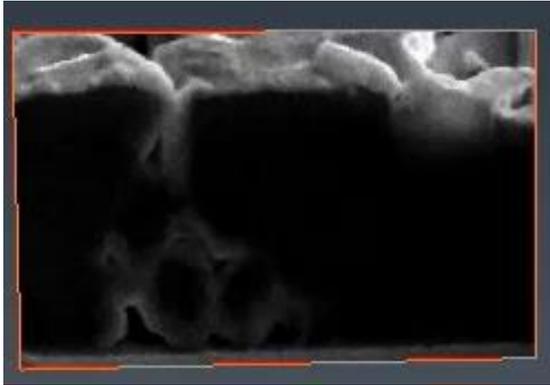
# Electrolyte! Electrolyte! Electrolyte!

J. Alvarado , ...Y.S. Meng, et al, Energy & Env. Sciences, 2019

**Gen II**

**New Salts**

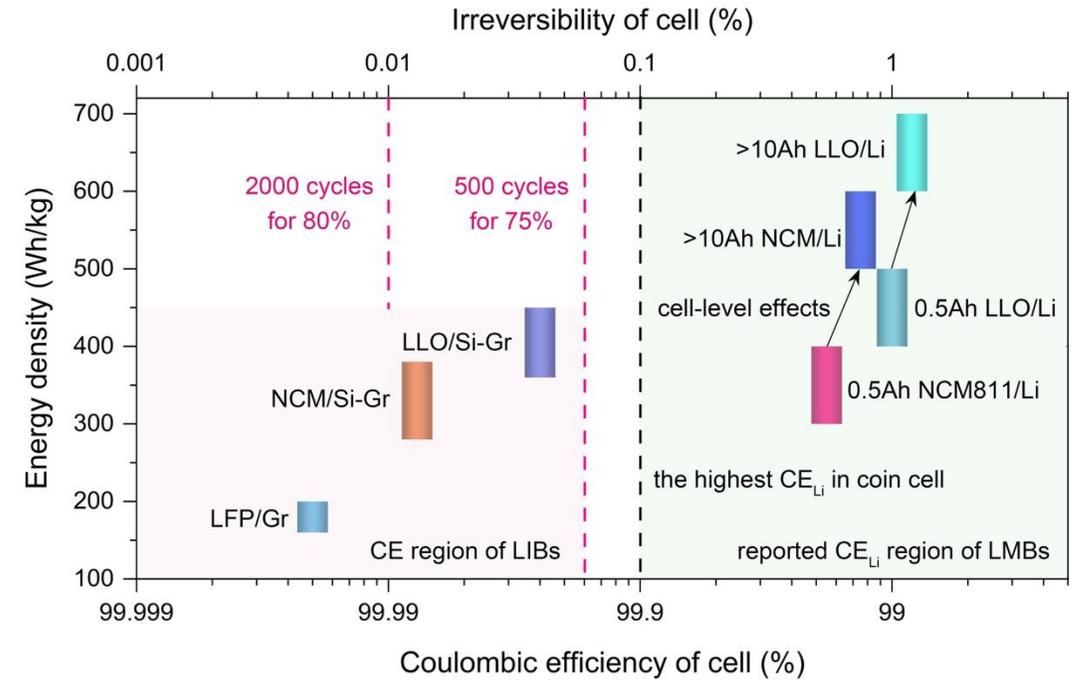
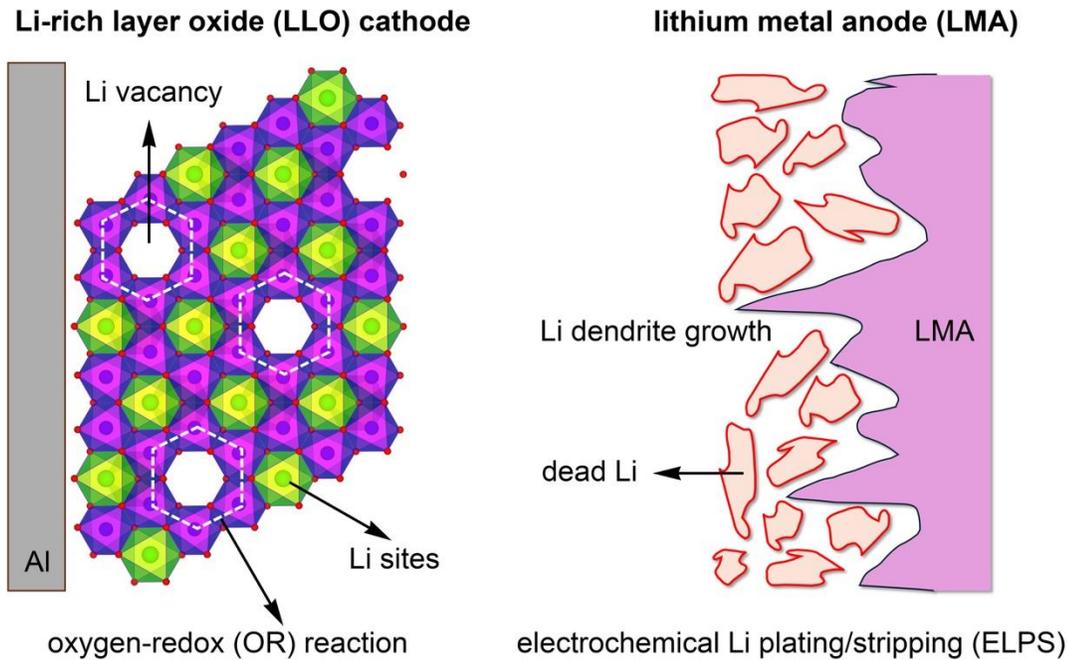
**New Electrolyte**



Breakthroughs in Electrolytes - Concentrated, Locally Coordinated, Non-carbonate  
**Columbic Efficiency increases from 98% to 99.6%**

- ◆ Li-rich layer oxide cathode is the promising layer oxide cathode for lithium metal battery beyond 600Wh kg<sup>-1</sup>
- ◆ the huge Coulombic efficiency gap between current lithium metal anode and the target

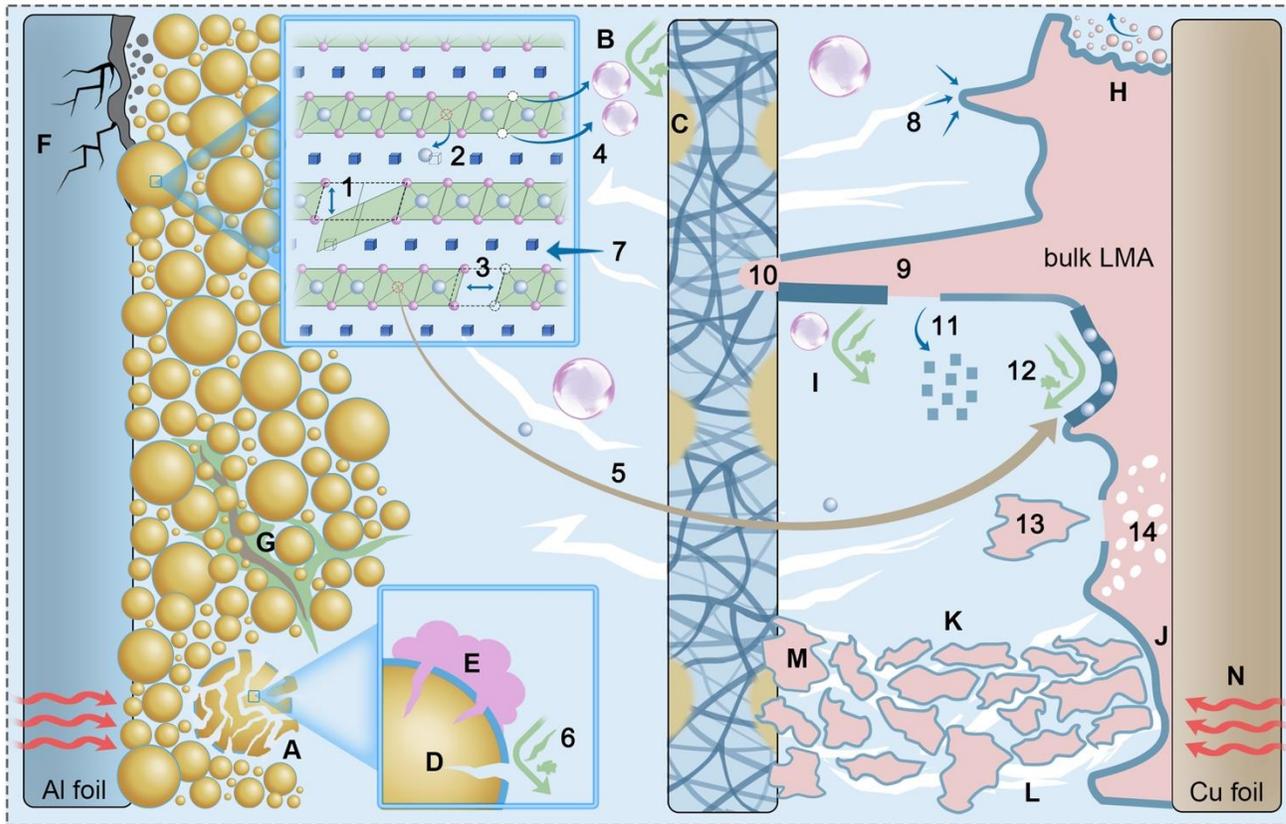
### lithium metal battery beyond 600 Wh/kg



**How can we achieve the Coulombic efficiency of Li > 99.90%?**

**What happens to the ~0.5% loss for Li during electrochemical plating/stripping?**

◆ illustrations of material-level degradation and corresponding cell-level impacts in lithium metal battery



- | cathode   | anode | electrolyte                                       | separator |
|---|-------|---|-----------|
| material-level degradation (1-14)                 |       | cell-level degradation (A-N)                      |           |
| 1: collapse of TM-oxygen polyhedra                |       | A: cathode particle pulverization                 |           |
| 2: surface TM ions migration to Li sites          |       | B: oxygen-induced electrolyte decomposition       |           |
| 3: distortion of TM-oxygen polyhedra              |       | C: byproducts blocking separator porosity         |           |
| 4: instability of lattice oxygen                  |       | D: surface-localized cracking of cathode          |           |
| 5: TM ions dissolution and migration to anode     |       | E: cathode interfacial byproducts                 |           |
| 6: electrolyte oxidation at cathode surface       |       | F: localized foil corrosion and cracking          |           |
| 7: increasing Li <sup>+</sup> transfer resistance |       | G: electrode without conductive contact           |           |
| 8: local ion depletion and charge accumulation    |       | H: galvanic corrosion of LMA                      |           |
| 9: Li dendrite growth                             |       | I: chemical corrosion of LMA                      |           |
| 10: penetration and rupture of SEI                |       | J: exhaustion of active Li in bulk LMA            |           |
| 11: SEI dissolution and decomposition             |       | K: accumulation of porous dead Li layer           |           |
| 12: electrolyte reduction at metallic Li          |       | L: localized insufficient electrolyte wettability |           |
| 13: root-preferred stripping to form dead Li      |       | M: dead Li blocking separator surface             |           |
| 14: multi-sites cavitation on bulk LMA            |       | N: uneven Joule heat distribution                 |           |

**the active Li loss in lithium metal anode plays key role in determining the lifespan of battery**

**BUT the active Li loss is masked by typical battery capacity fade**

# Possible to Invent More Electrolytes to Enable Metal

Aradoc

DispatchDate: 22.09.2021 · ProofNo: 910, p.1

nature  
energy

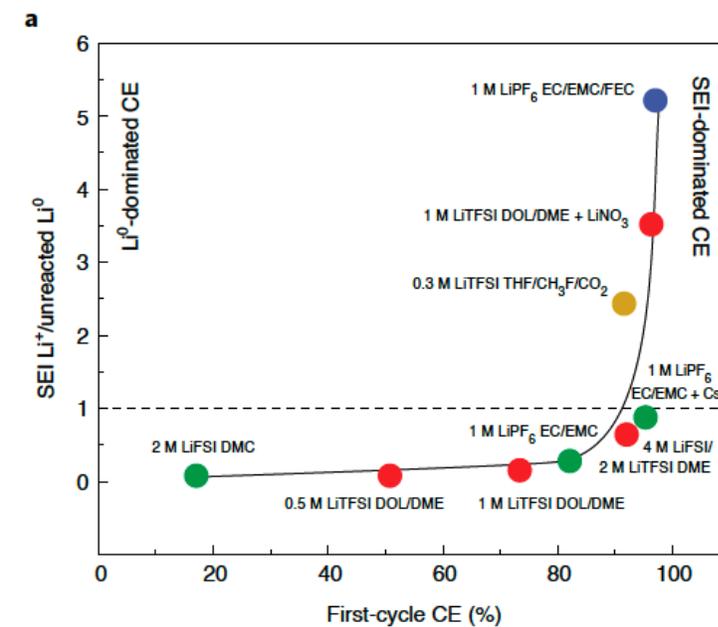
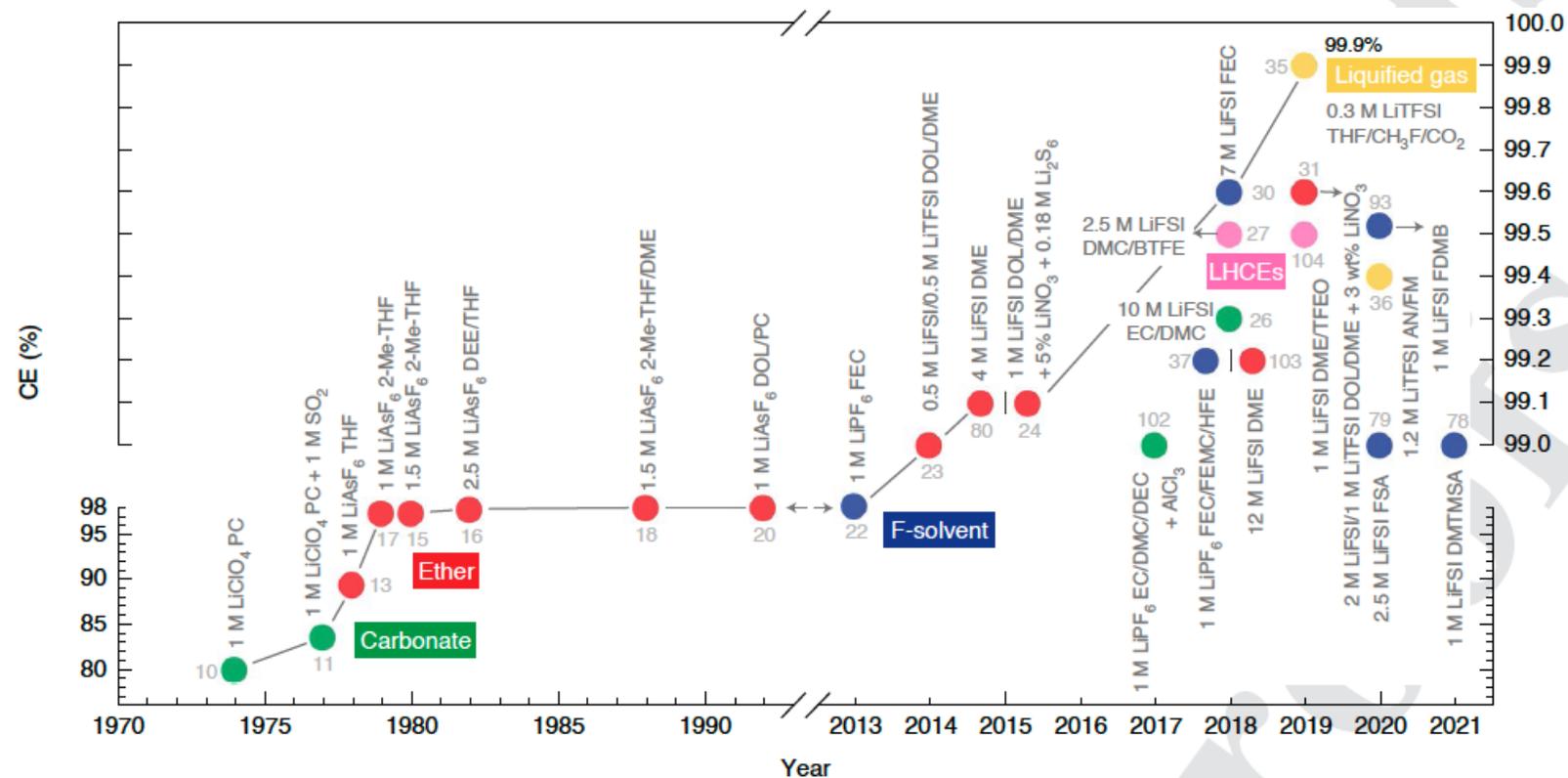
REVIEW ARTICLE

<https://doi.org/10.1038/s41560-021-00910-w>

Check for updates

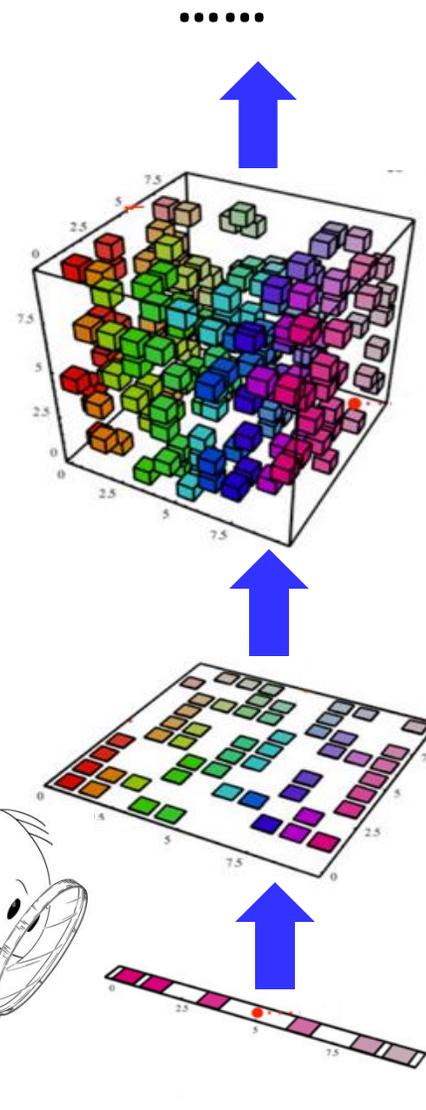
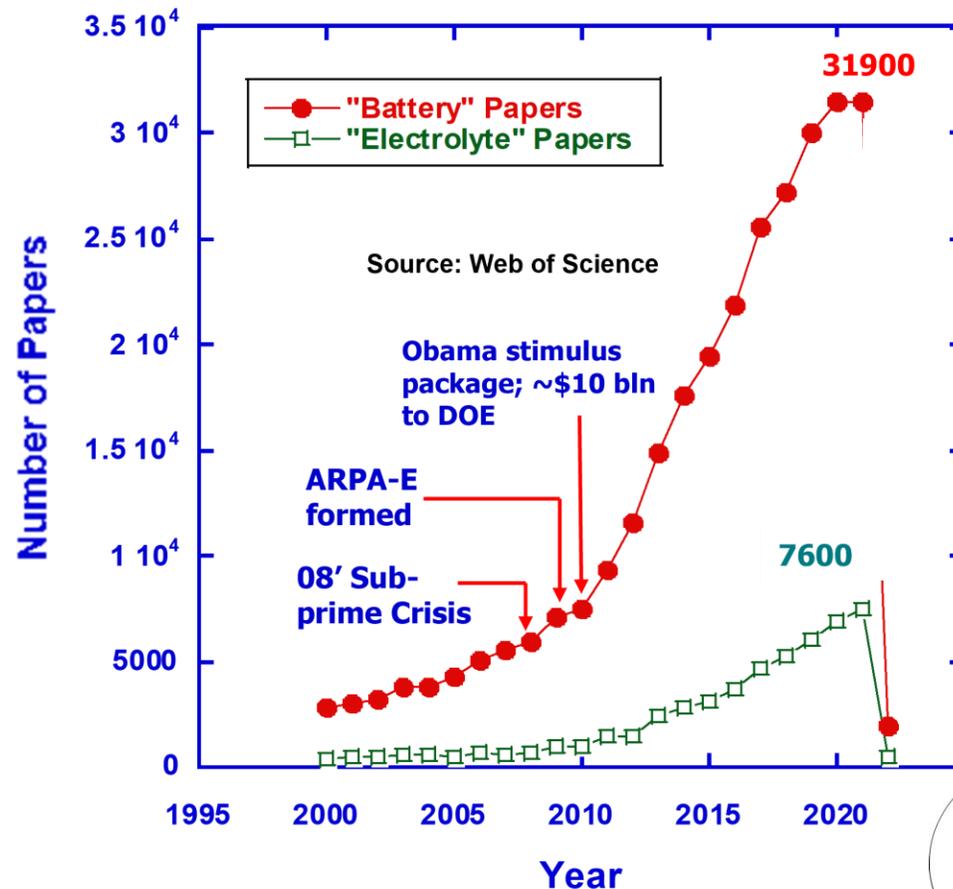
Collaboration with  
Prof. Betar Gallant and Yang Shao-Horn

## Moving beyond 99.9% Coulombic efficiency for lithium anodes in liquid electrolytes



# Electrolyte Literature – A New Barrier for New Discovery?

- Battery research has been “hot” since 2010
- In **2021**, around **650** battery papers published per week
  - ~150 papers on “battery electrolyte”
- Not a single individual can follow the literature
- Let alone understanding in-depth, or identifying patterns/relations underneath the text



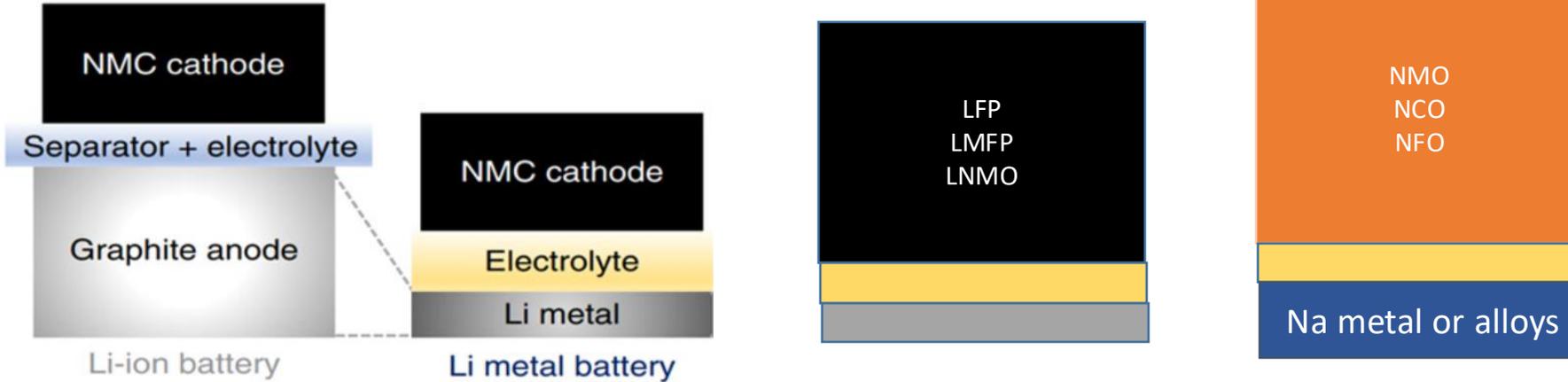
Make Contact  
with the Molecular  
Universe.



# All Solid-State Batteries – Platform Technology

## High-Energy-Density and Safe Batteries

### with Solid-State Electrolyte



**Energy Density > 500Wh/kg**

Conversion type Cathodes

Metal Anodes

Ultra Thin Separator

### Safety

Particularly for Oxides Sulfides ?

Polymers X

**Fluorine - Free Chemistry**

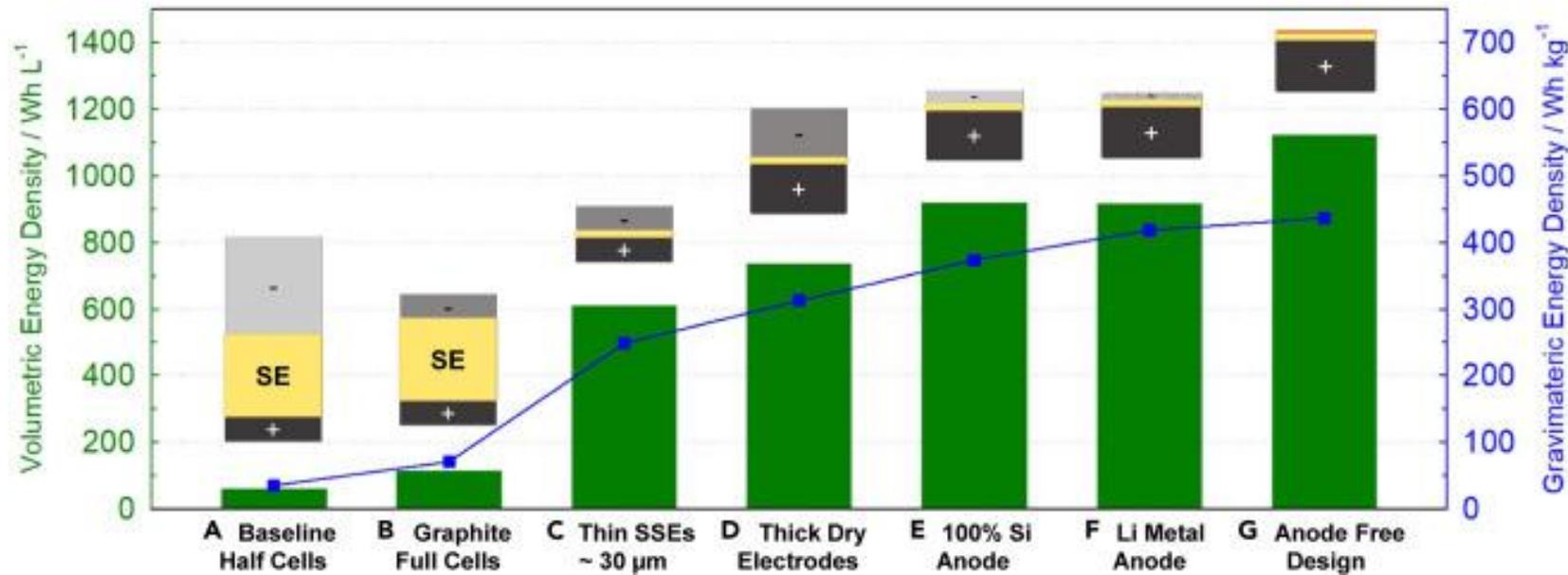
**Stackable Design**

**Dry Processing**

**Ultralong Cycle Life**

**Wider Operation Temperature**

**Enable Conversion Chemistry**



# Sodium Anode-Free Solid-State Batteries

Can achieve 3 goals simultaneously...

## 1. Maximize energy density

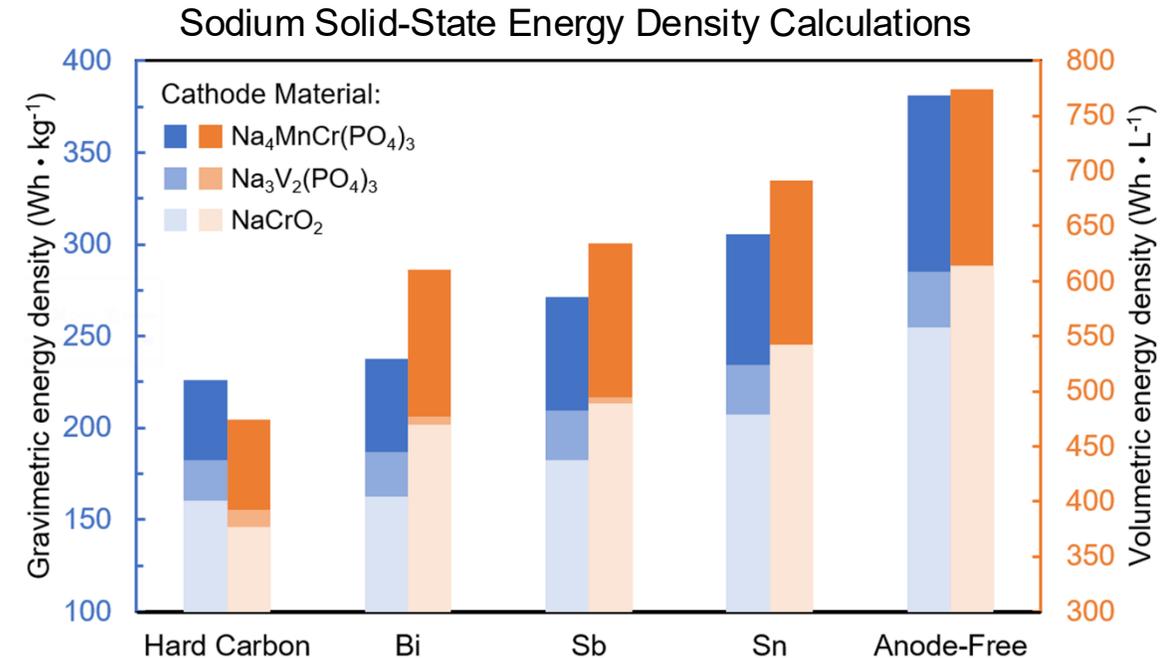
- Lowest reduction potential → highest cell voltage
- Smaller and lighter cells

## 2. Minimize cost

- No anode material cost, lower processing cost
- Sodium cheaper than Lithium

## 3. Improved safety

- No flammable organic liquid electrolytes
- No large amounts of sodium metal foils



# Critical Design Factors

G. Deysher, J.A.S. Oh, ... Y.S. Meng, "[An Anode-Free Sodium All-Solid-State Battery](#)", 2024, Nature Energy

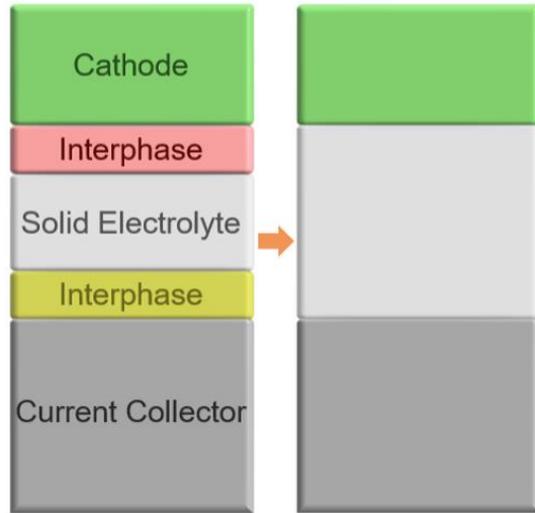


Dr. Grayson Deysher

- 4 fundamental criteria for enabling anode-free solid-state batteries

## i. Electrochemically Stable Electrolyte

- Mitigate Na Inventory Loss -

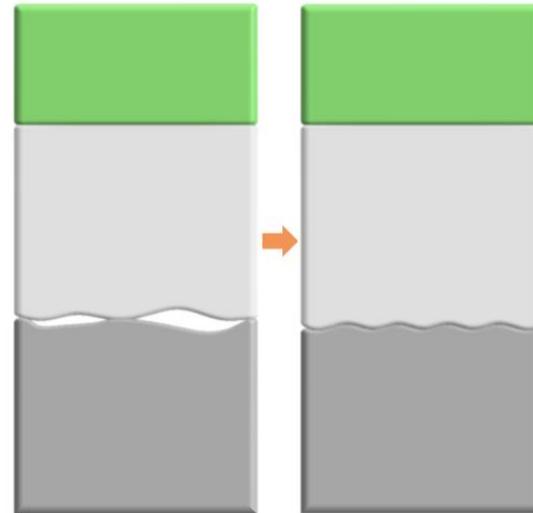


X

✓

## ii. Intimate Interface Contact

- Uniform Plating/Stripping -

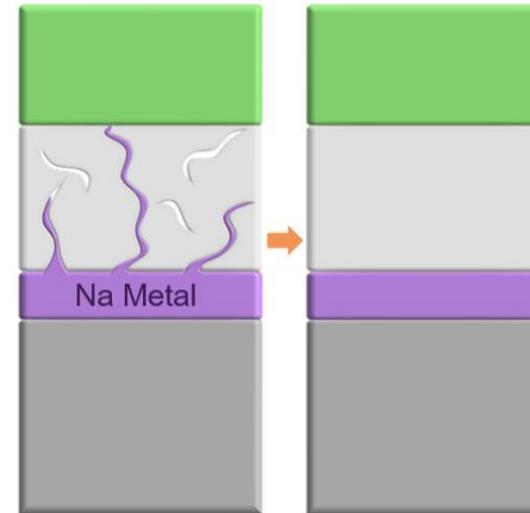


X

✓

## iii. Dense Solid Electrolyte

- Prevent Dendrite Growth -

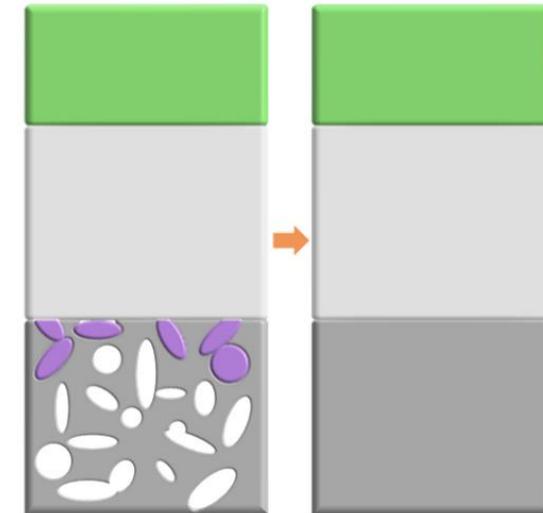


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✓

## iv. Dense Current Collector

- Avoid Na<sup>0</sup> Trapping -



X

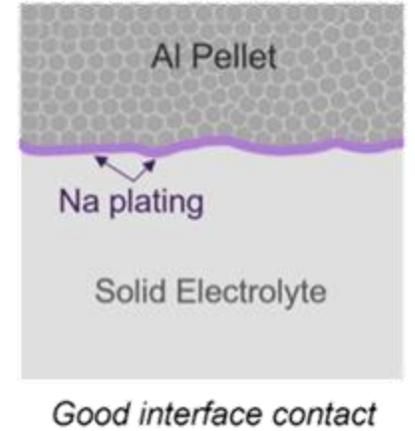
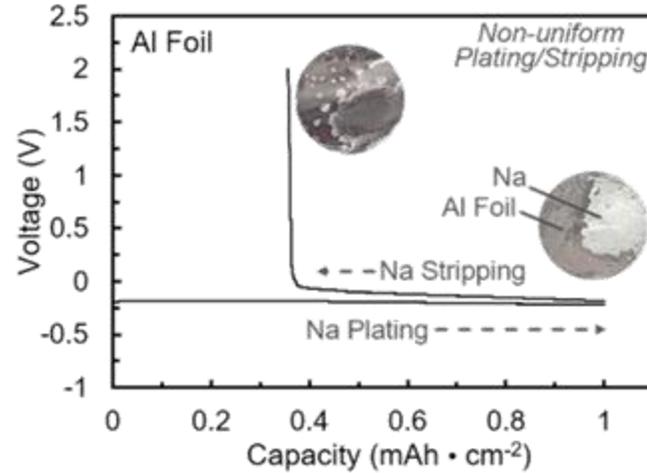
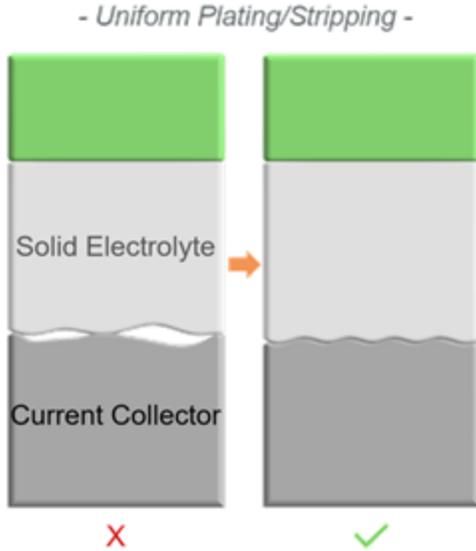
✓

- Must design a cell from the ground up
  - Material selection and cell architecture

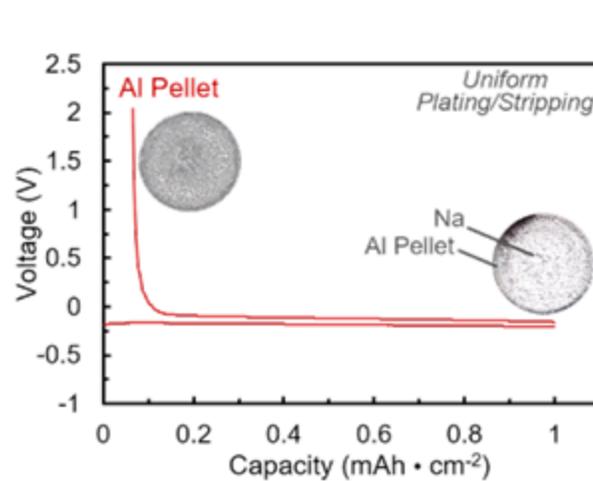
# Intimate Interface Contact

ii. G. Deysher, J.A.S. Oh, ... Y.S. Meng, "[An Anode-Free Sodium All-Solid-State Battery](#)", 2024, Nature Energy

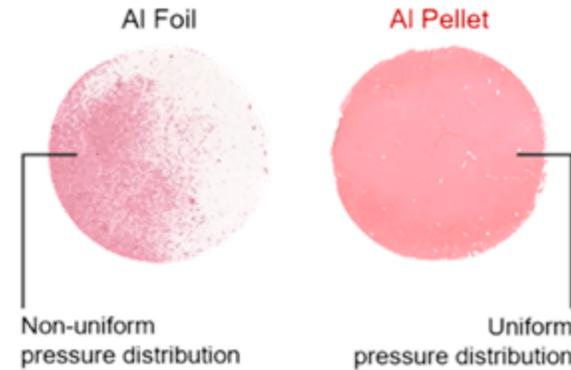
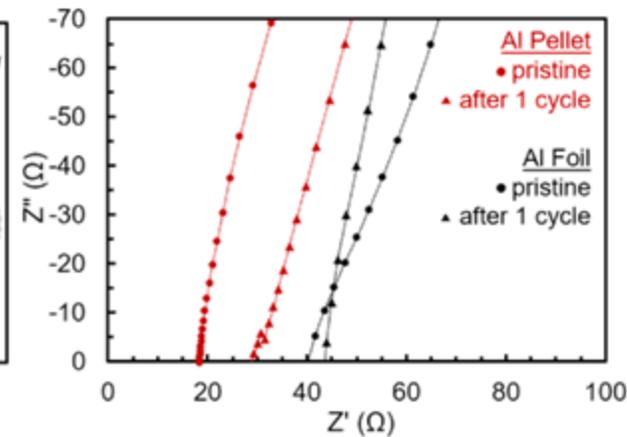
Patent Pending



- Pelletized Al can achieve intimate contact with the NBH electrolyte



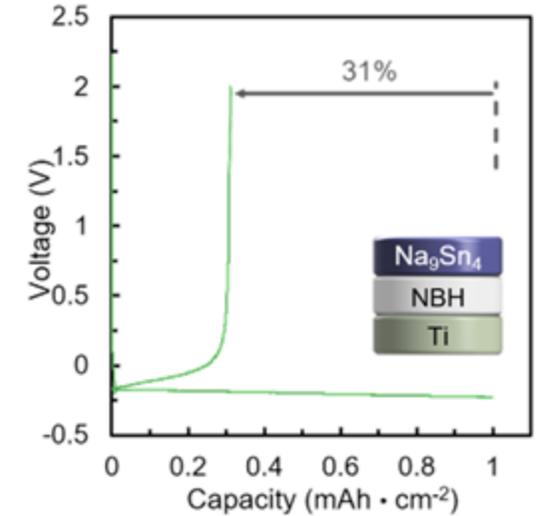
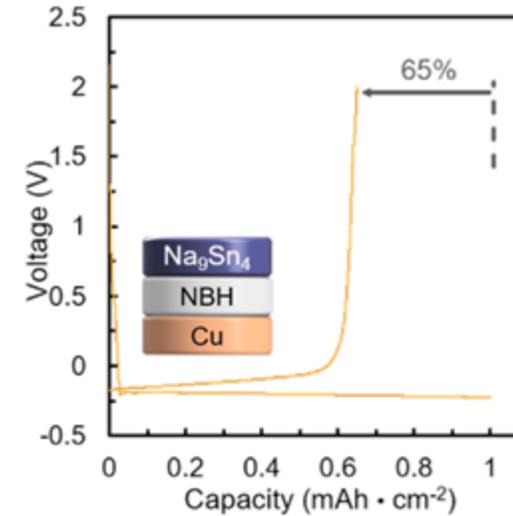
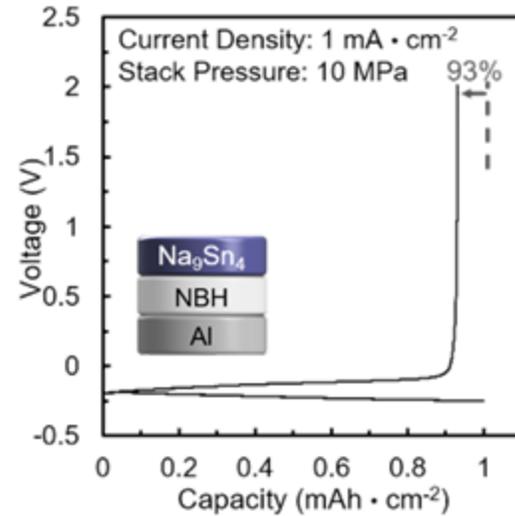
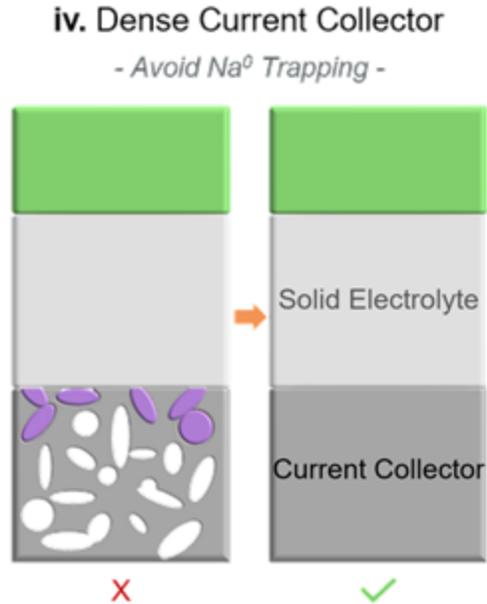
Al Pellet = Lower interface resistance



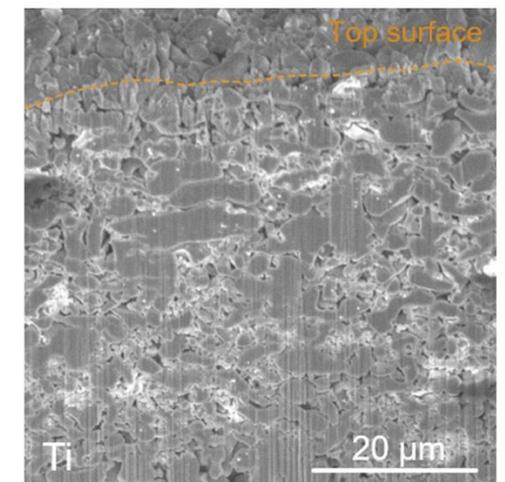
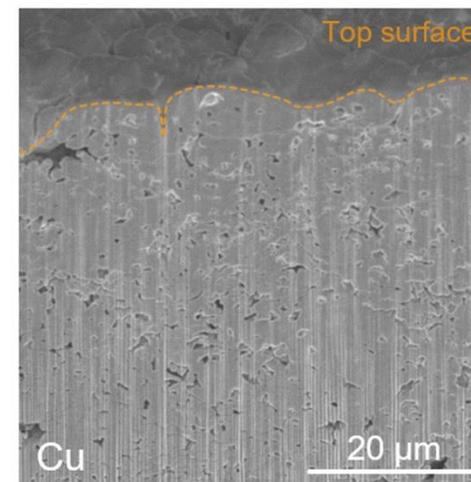
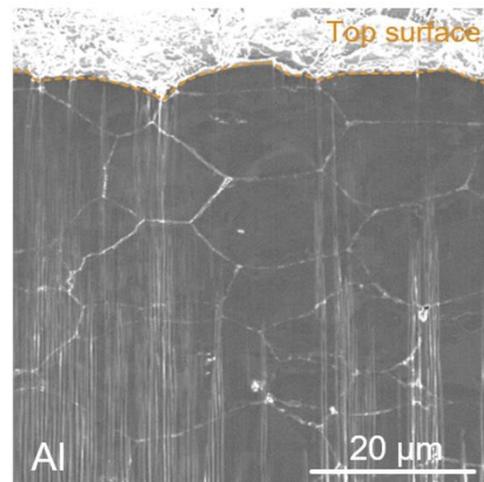
# Designed Dense Current Collector

G. Deysher, J.A.S. Oh, ... Y.S. Meng, "[An Anode-Free Sodium All-Solid-State Battery](#)", 2024, Nature Energy

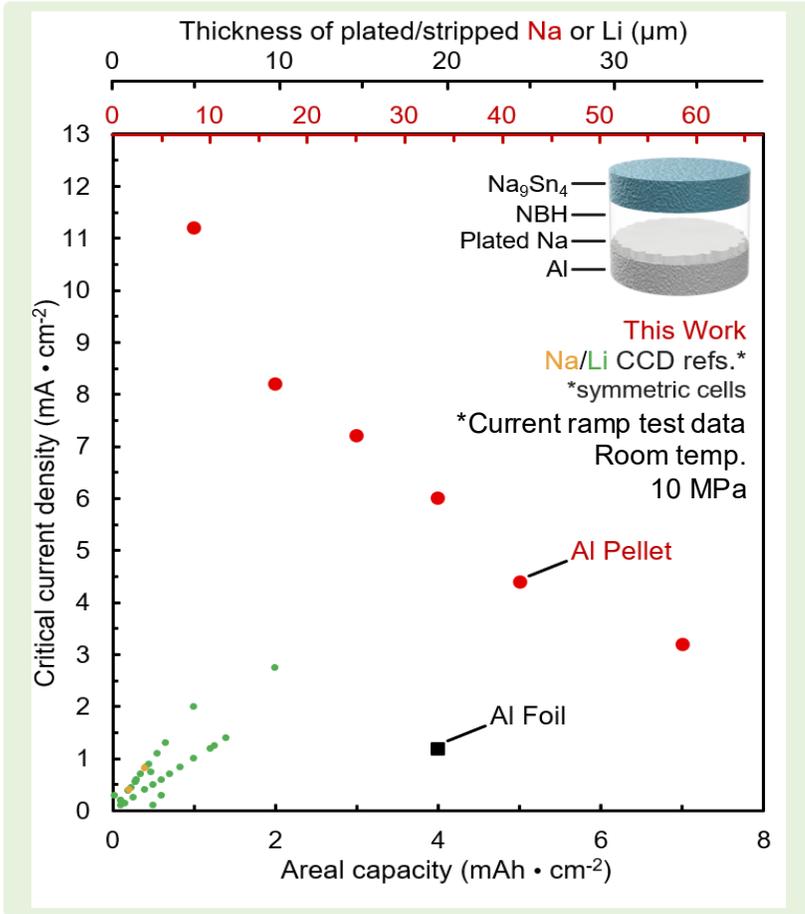
Patent Pending



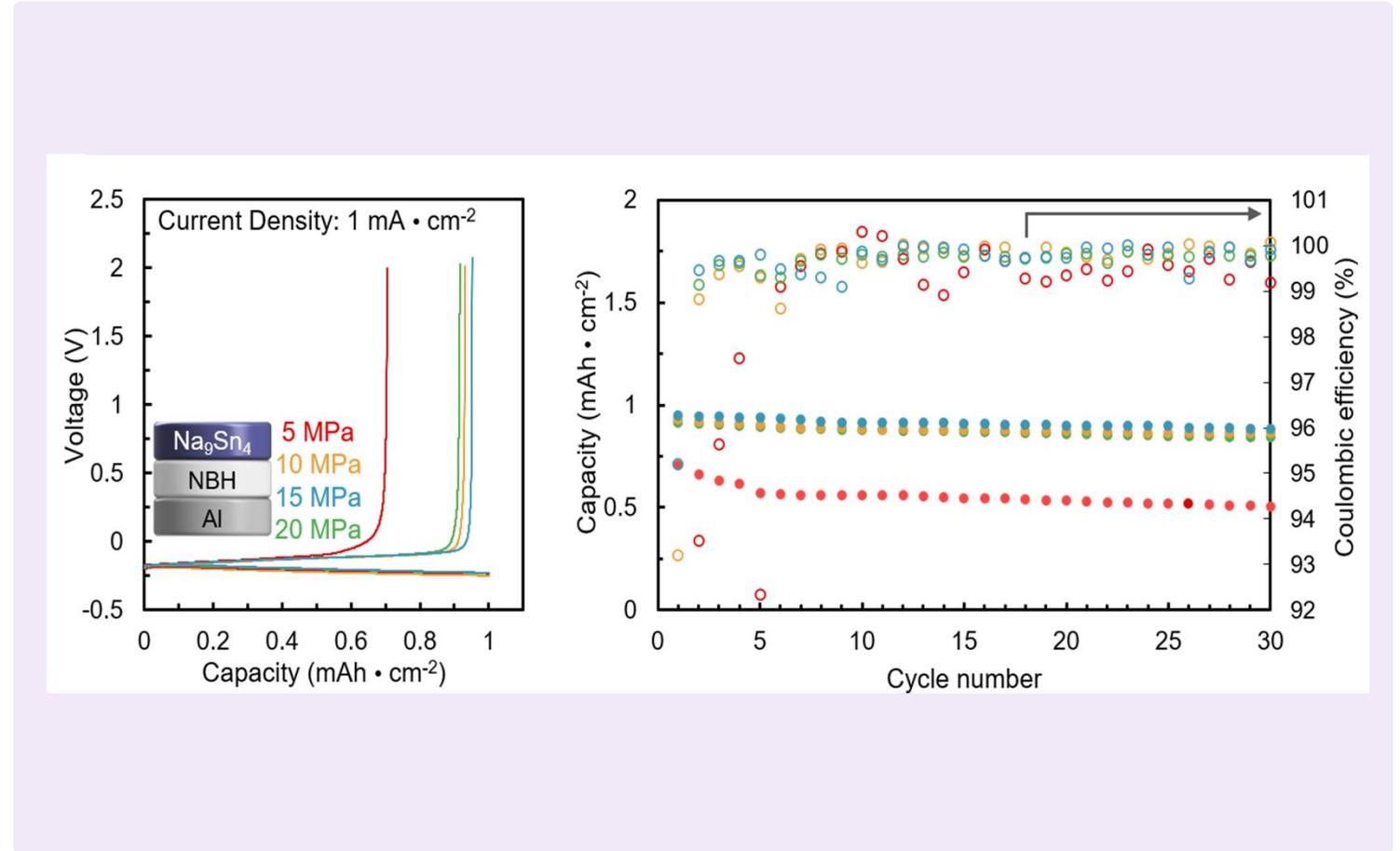
- Soft Al can be easily densified during cell fabrication
  - Unlike Cu and Ti



## Current Density Patent Pending



## Cell Stack Pressure



- High critical current density
  - Pathway to fast charging

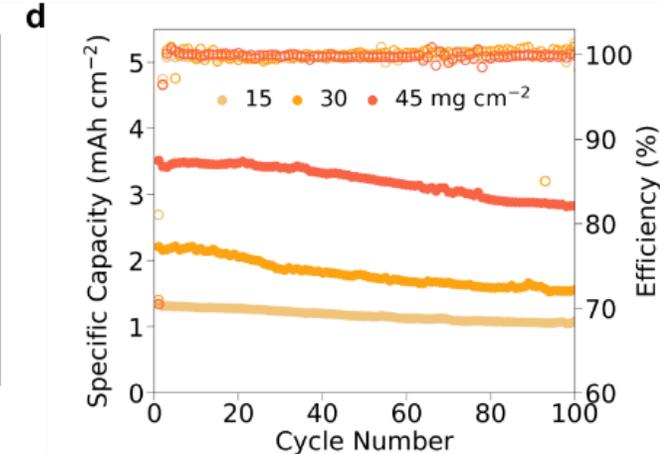
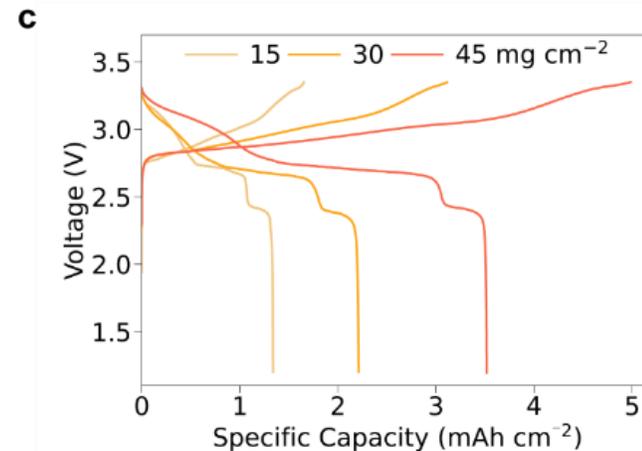
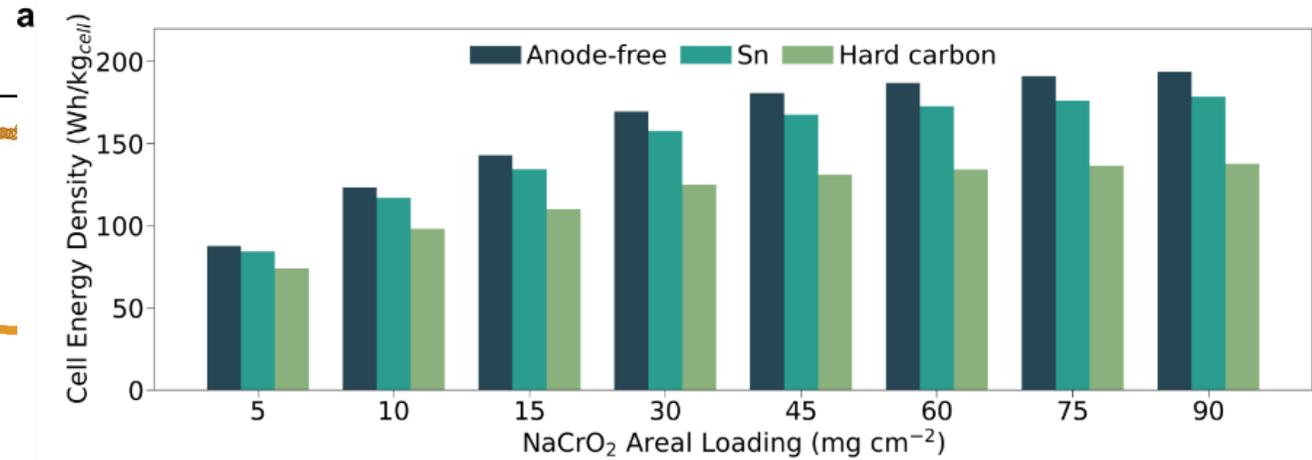
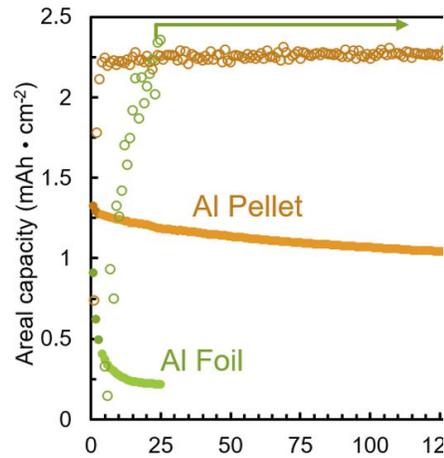
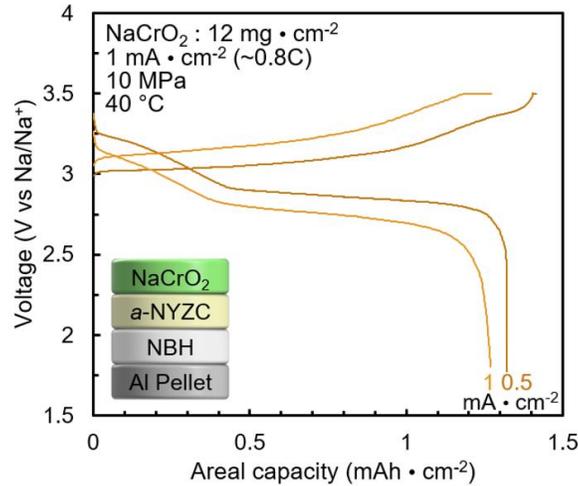
- Low pressure cyclability
  - Pathway to practical commercial cells

# Sodium All Solid-State Full Cell

G. Deysher, J.A.S. Oh, ... Y.S. Meng, "[An Anode-Free Sodium All-Solid-State Battery](#)", 2024, Nature Energy

Patent Pending

- Enable full-cell cycling (NaCrO<sub>2</sub> cathode)



**Thick Battery Cathode enabled by novel superionic Na conductor**

Dr. J.A.S. Oh [Patent Pending and To be Submitted](#)



e

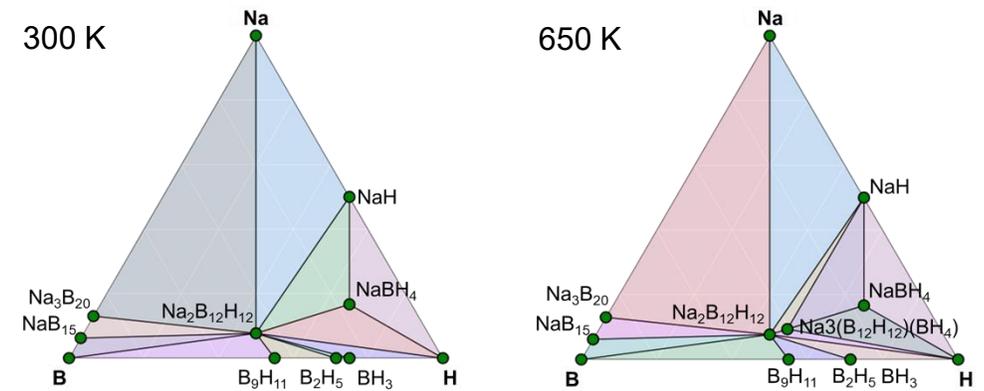
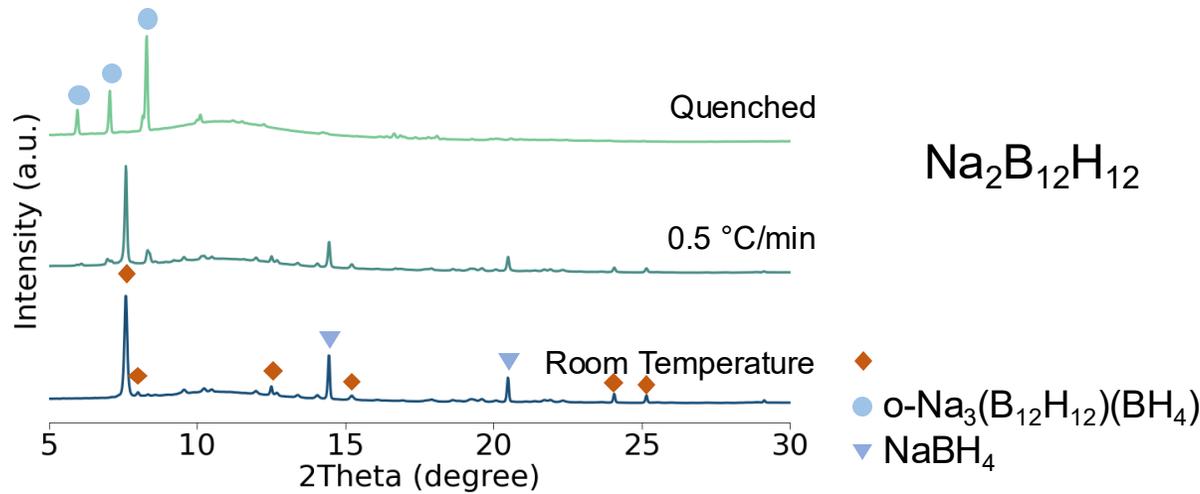
# Materials Discovery – Metastable o-NBH

Patent Pending



Dr. Sam Oh

- $\text{Na}_2\text{B}_{12}\text{H}_{12}$  reacted with  $\text{NaBH}_4$
- Both are poor  $\text{Na}^+$  conductors on their own



- Different cooling rate exhibits different crystal structure at room temperature
  - o-NBH stabilized with quenching
  - $\text{Na}_2\text{B}_{12}\text{H}_{12}$  and  $\text{NaBH}_4$  patterns when cooled slowly
  - Metastable nature of o-NBH stabilized by rapid cooling

# Activation Energy

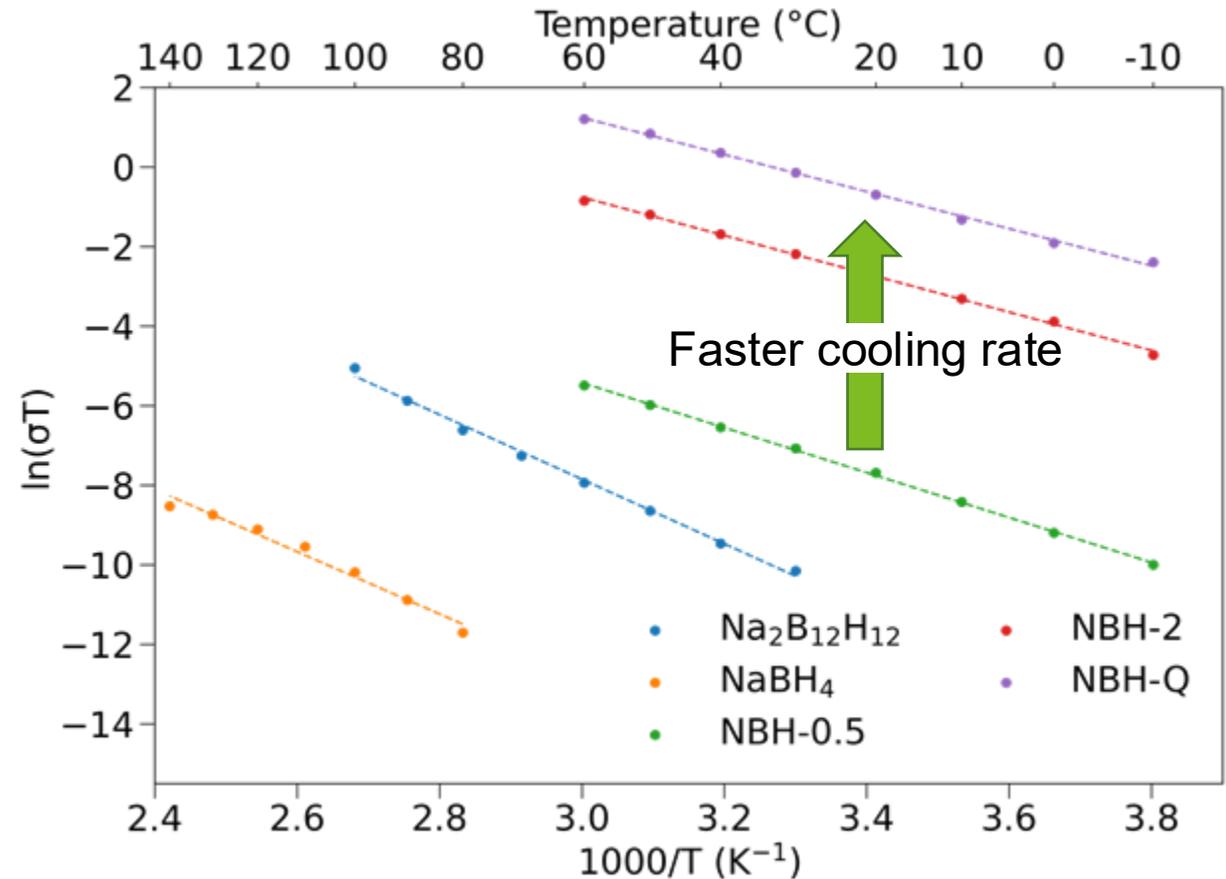
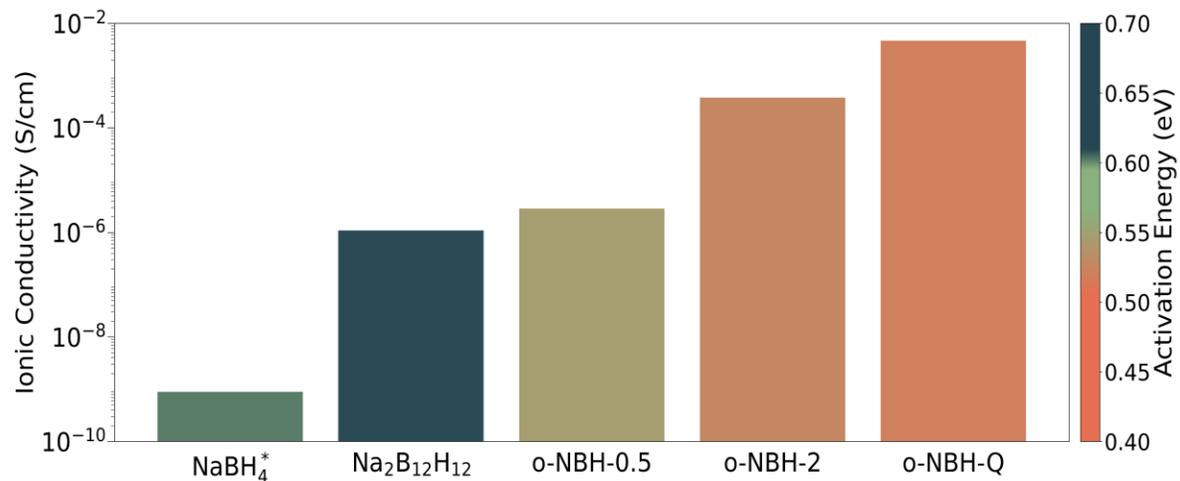
Activation energy estimated by EIS measure at different temperature

➤ Ionic conductivity – temperature follows the Arrhenius relationship

$$\sigma_T = \sigma_T \exp^{-E_a/kT}$$

➤ Faster cooling rate maximizes ionic conductivity

➤ Quenched



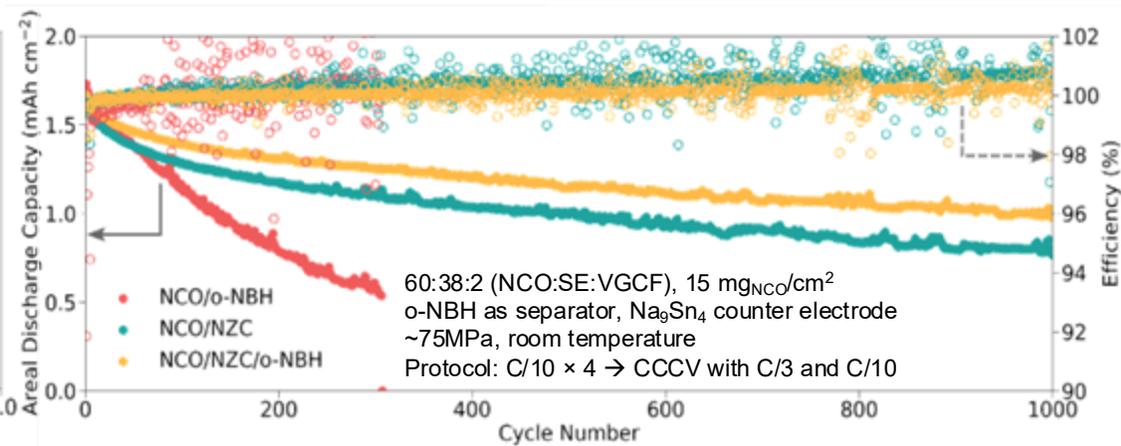
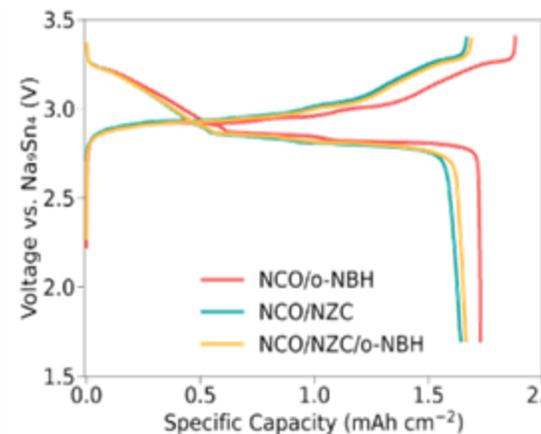
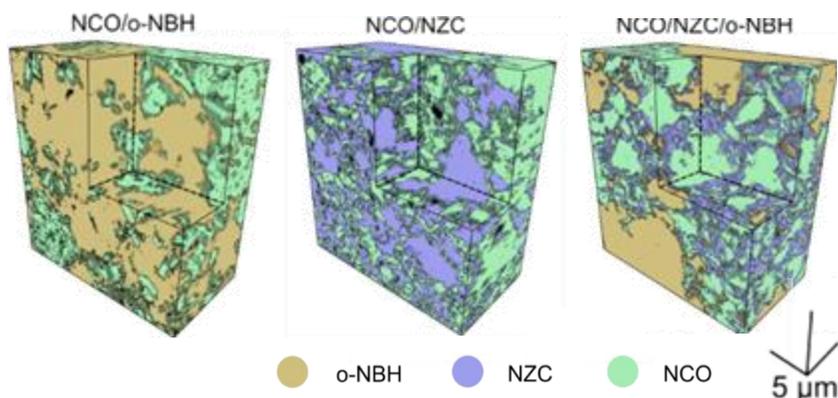
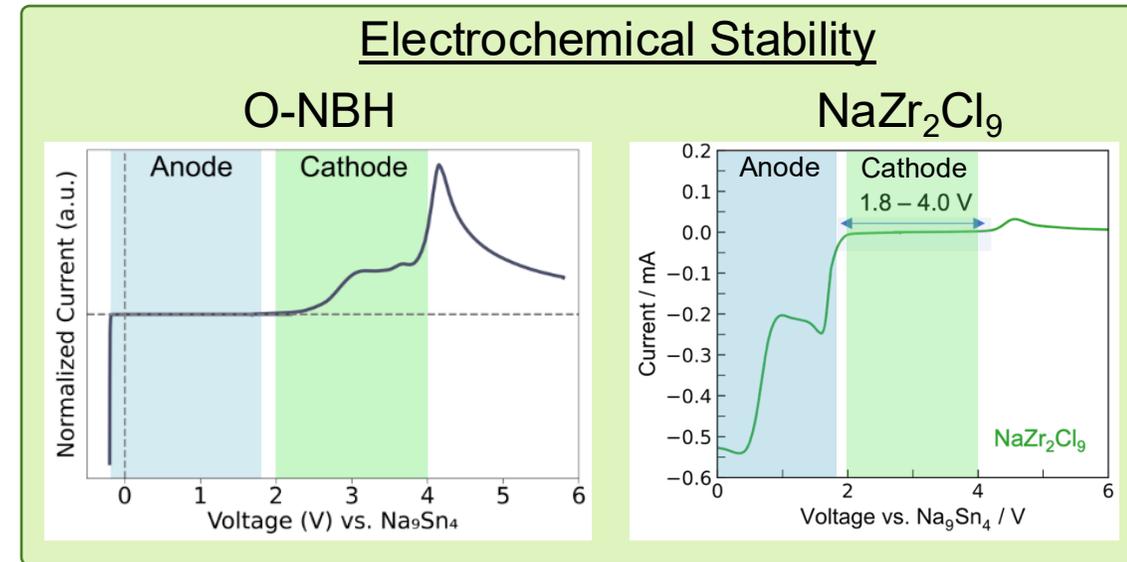
# Cathode Composite Electrochemical Performance



- o-NBH is not stable at cathode potentials
- Oxidizes around 2.5 V

New cathode design strategy →

- Coat  $\text{NaCrO}_2$  cathode with stable  $\text{NaZr}_2\text{Cl}_9$
- Stable cathode-electrolyte interface
- Mix in o-NBH for fast  $\text{Na}^+$  conduction



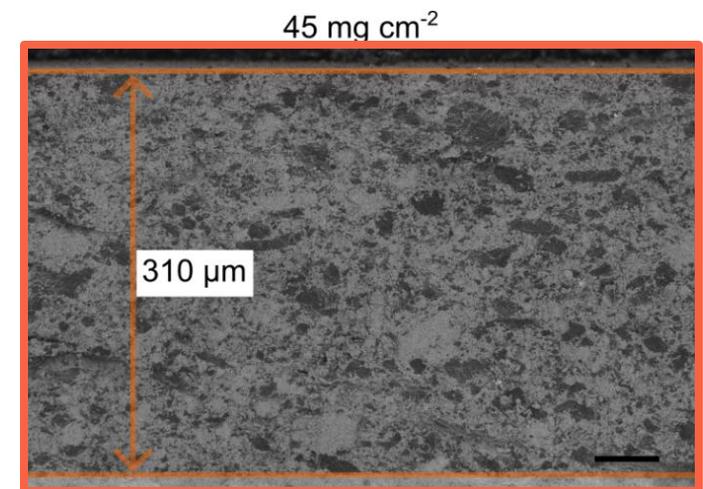
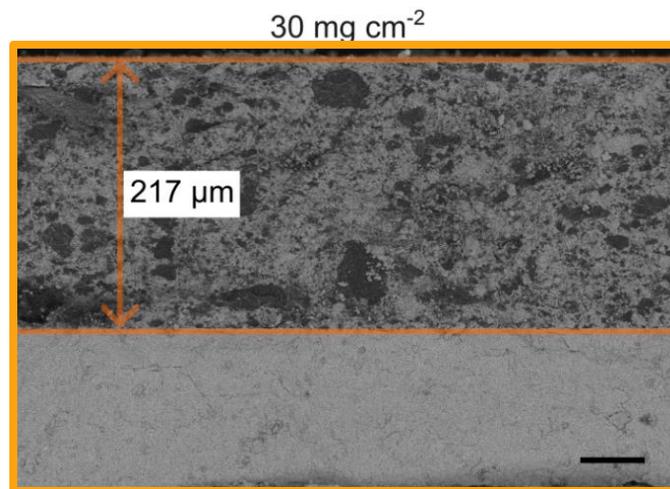
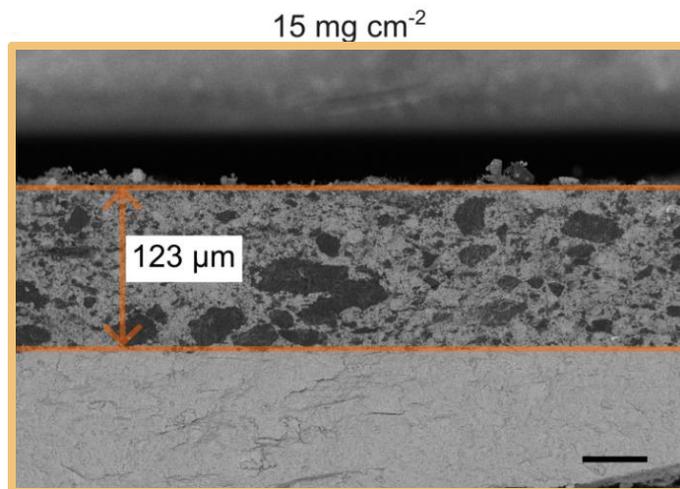
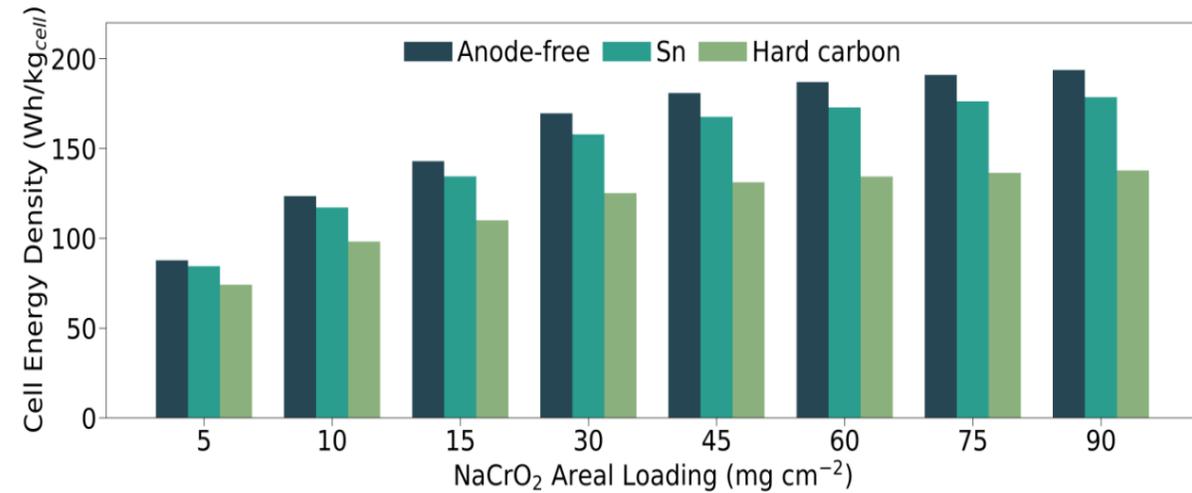
# High Loading Full Cell Cycling

Energy density increases with higher cathode loading

- Plateaus after  $\sim 45 \text{ mg/cm}^2 \text{ NaCrO}_2$

Cathode composites fabricated up to  $\sim 300 \mu\text{m}$  thick

- Theoretical capacity =  $5.4 \text{ mAh/cm}^2$



# Conclusions and Outlook

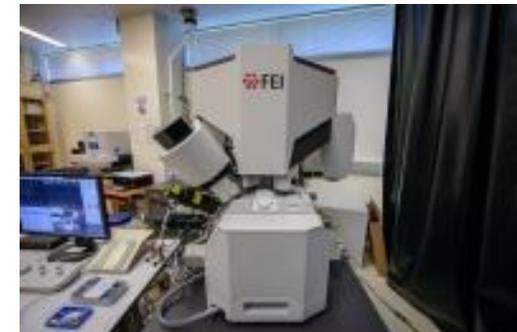
All-solid-state batteries are in good progression:

- Realize high-capacity anode (Si) with excellent capacity retention
- Prelithiation works best with minimum structural loss cathode
- Morphological control is critical
- Proof-of-concept pouch cell operating at **5 MPa** and exposed to dry-room condition

Sodium solid-state batteries are promising complementary to lithium counterpart:

- Research efforts in sodium batteries still lagging
- $\text{Li}_6\text{PS}_5\text{Cl}$  analogue is found in sodium
- *Amorphous* SSE path new research direction to super-ionic conductivity!

# Collaborators and Funding



(John Vetrano, Jane Zhu and  
Craig Henderson)



Vehicle Technology Office  
(Tien Duong and David Howell)

**LGES Frontier Research Laboratory (2021 – present)**  
Various Industrial Partners including  
(**LG Energy Solution / ThermoFisher Scientific / Shell / UL /Nissan /Cummins/ SES/ Supernal /GM / Applied Materials/Tesla/BMW**)

Alumni  
of  
LESC



**Na Batteries Since 2010**



**National Science Foundation  
DMR program (Na Batteries)**

# Global Race for Solid State Battery – West vs. East

