

Enabling ALL-Solid State Battery Via Interfacial Science & Engineering

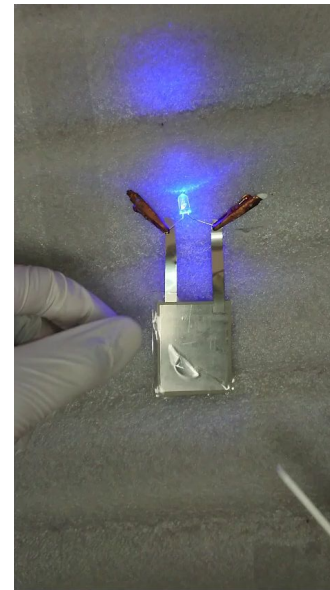
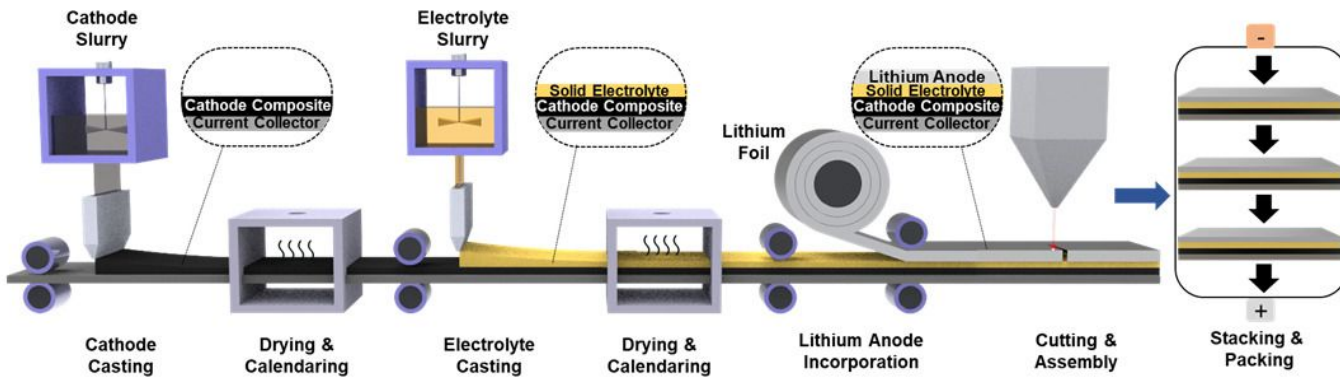
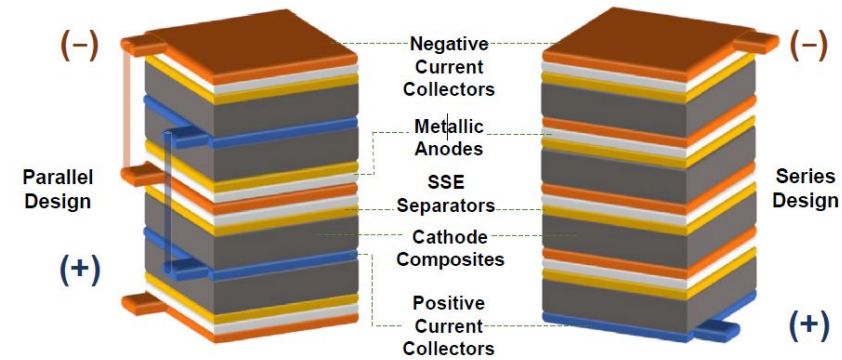
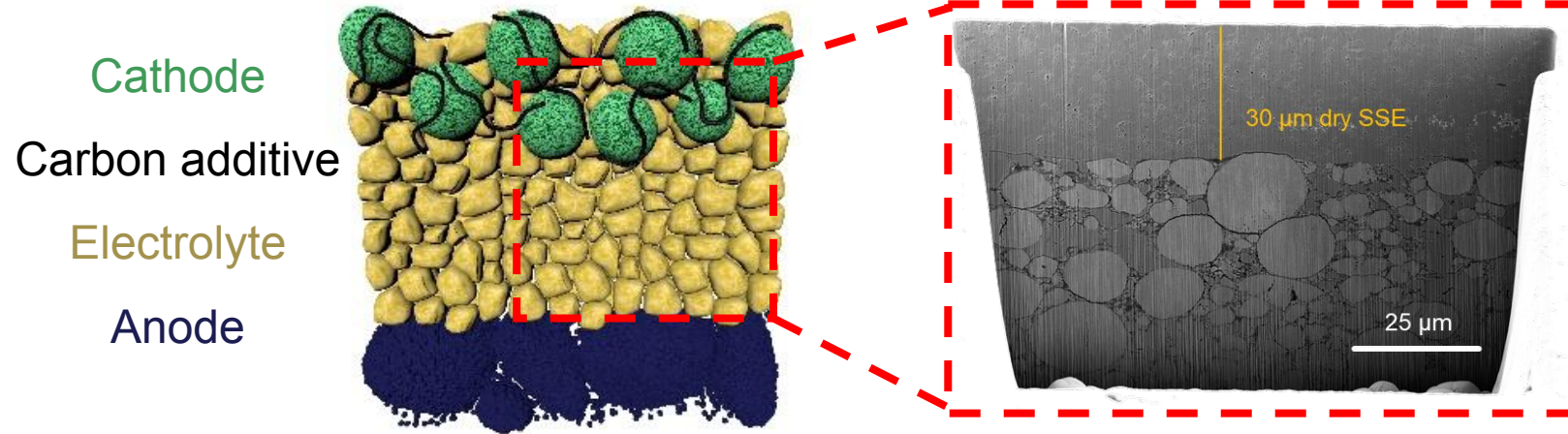
Y. Shirley Meng

University of Chicago

University of California San Diego

Argonne Collaborative Center for Energy Storage Science

A Platform Technology Enabled by Green Manufacturing



Series (bi-polar) stacking:

- Reduces inactive materials components → increase energy density
- Higher overall voltage per cell

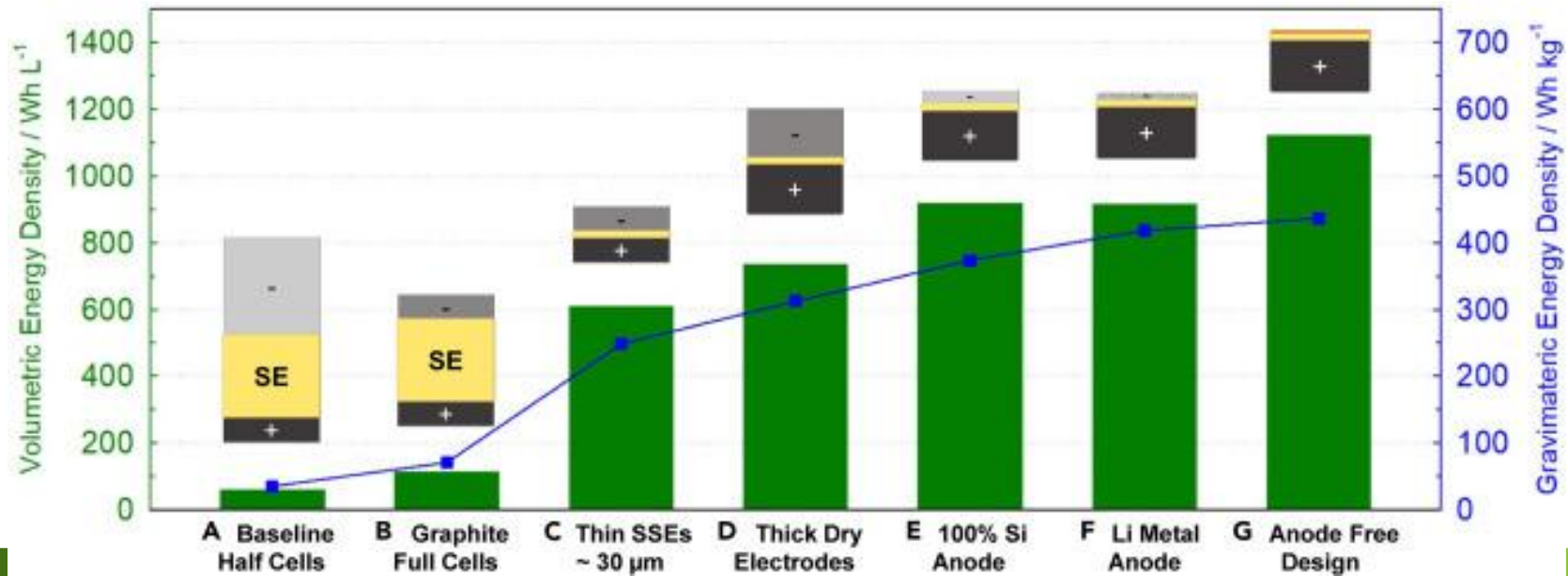
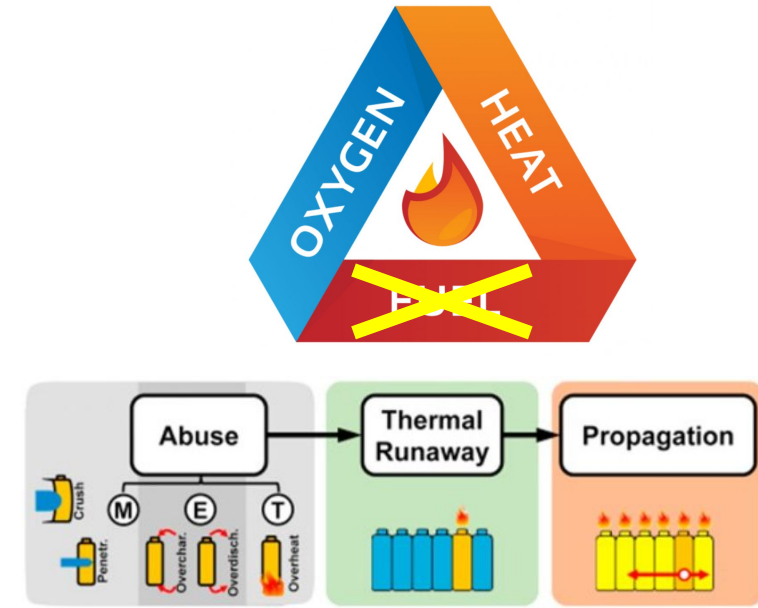
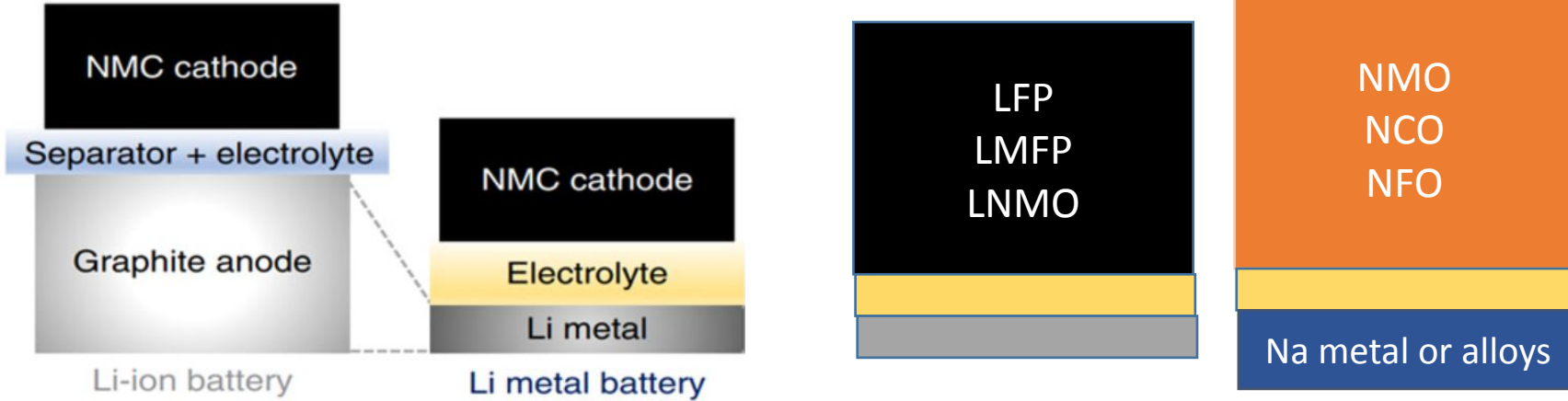
- Enhanced safety and abuse tolerance

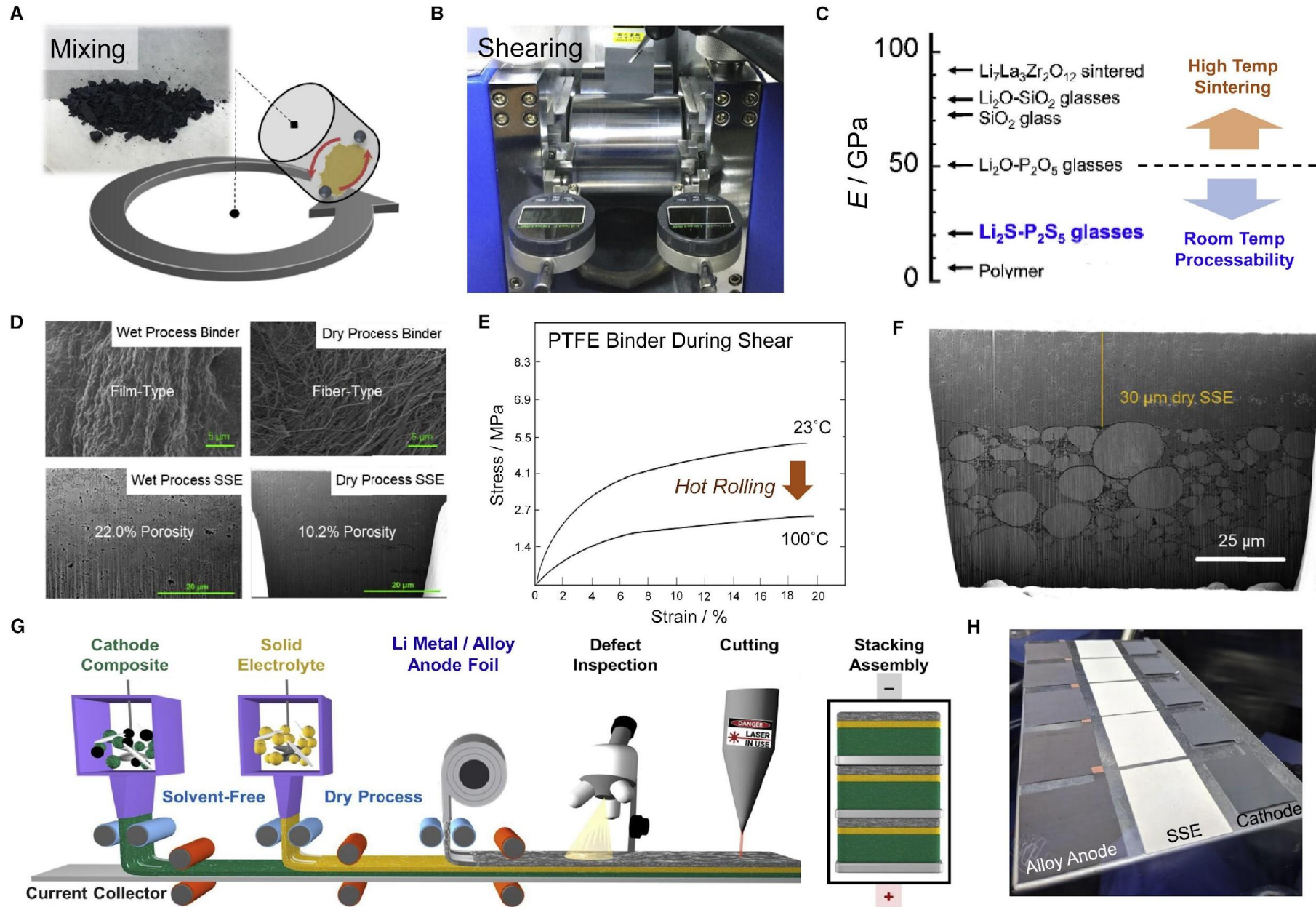
- Thin electrolyte film <math><30\mu\text{m}</math>
- High loading cathode >math>5\text{mAh}/\text{cm}^2</math>
- Stackable design – bipolar design
- Dry processing – green manufacturing

All solid-state batteries – Platform Technology

High-Energy-Density and Safe Batteries

with Solid-State Electrolyte





Dr. Darren Tan



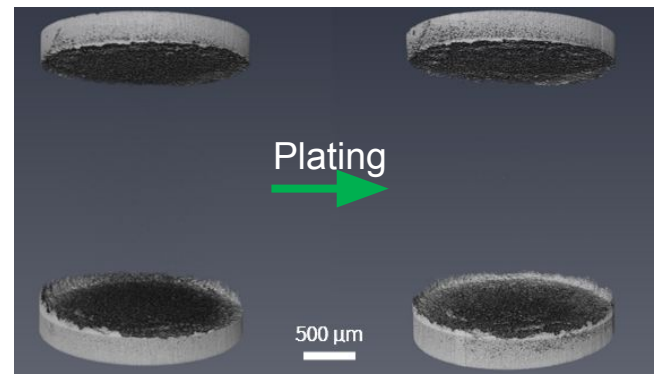
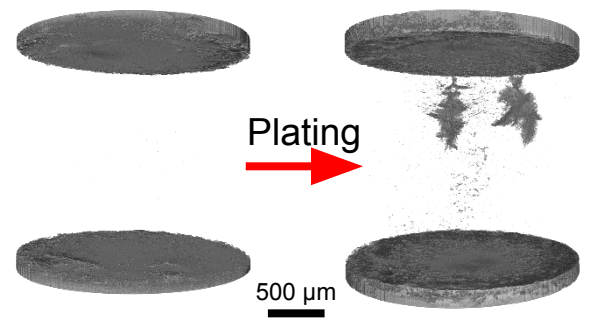
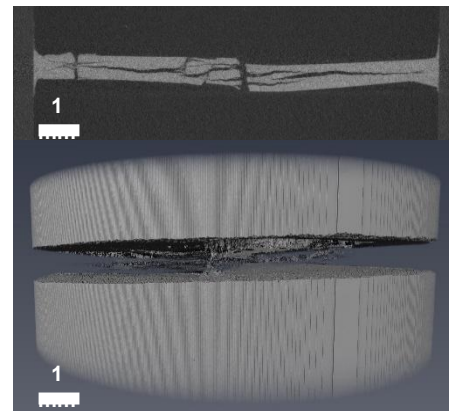
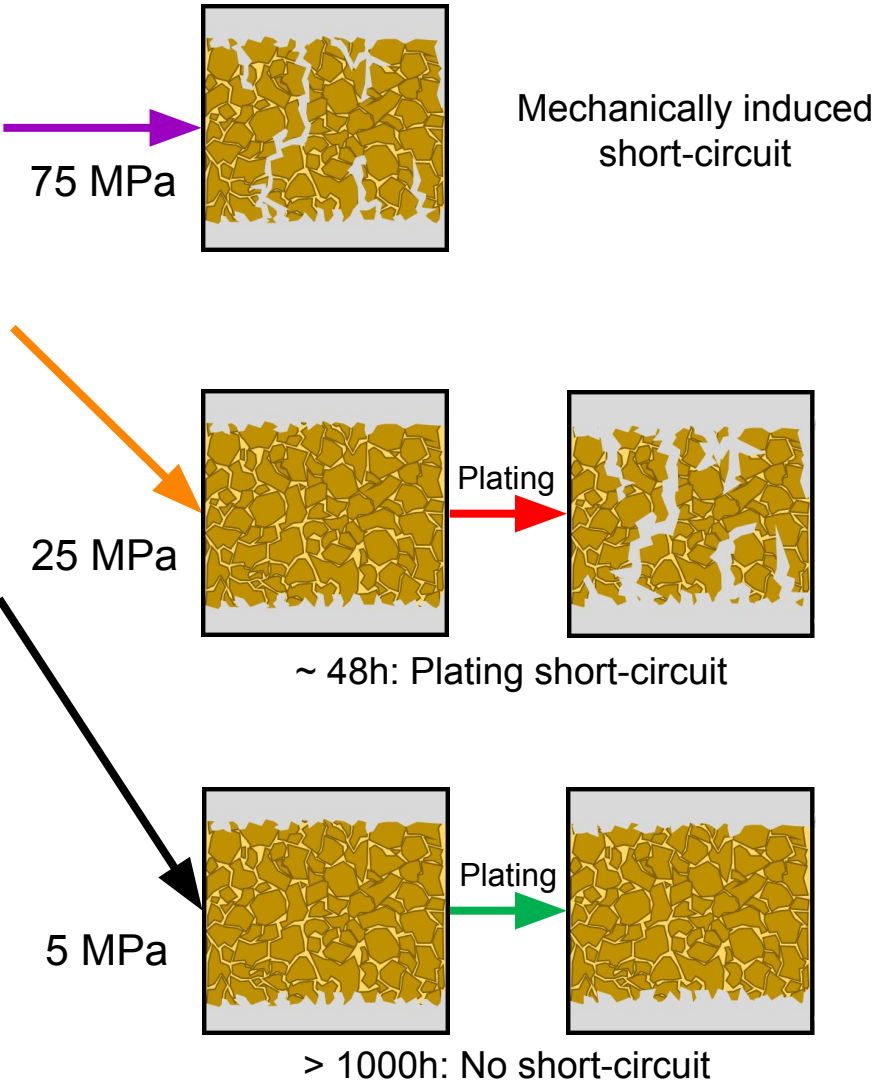
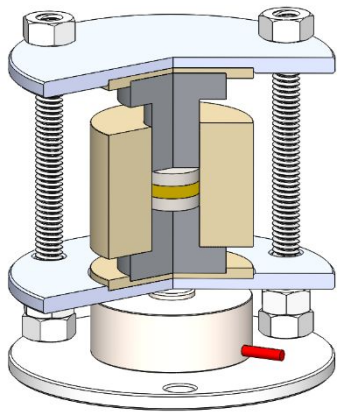
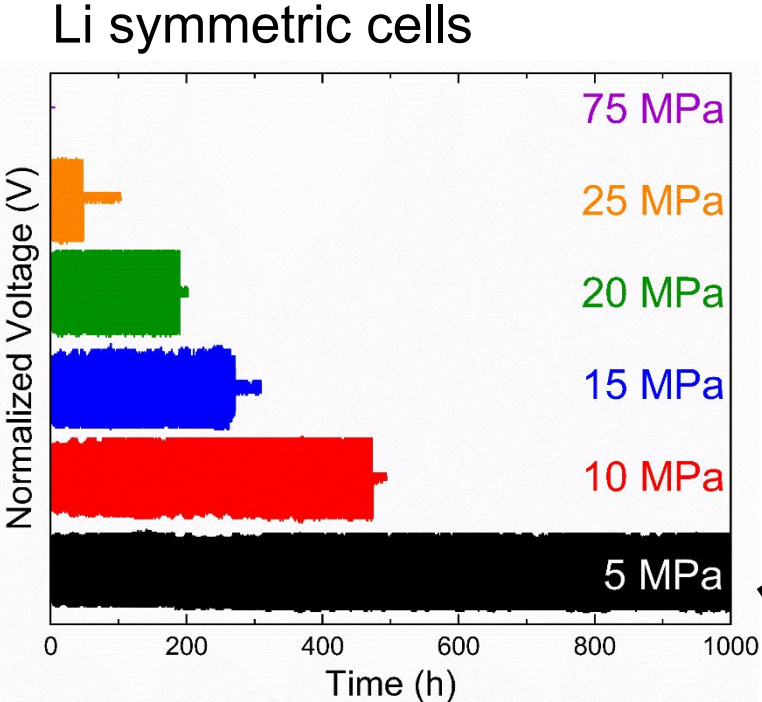
Dr. Jihyun Jang

When will All Solid-State Battery Will be Commercialized ?

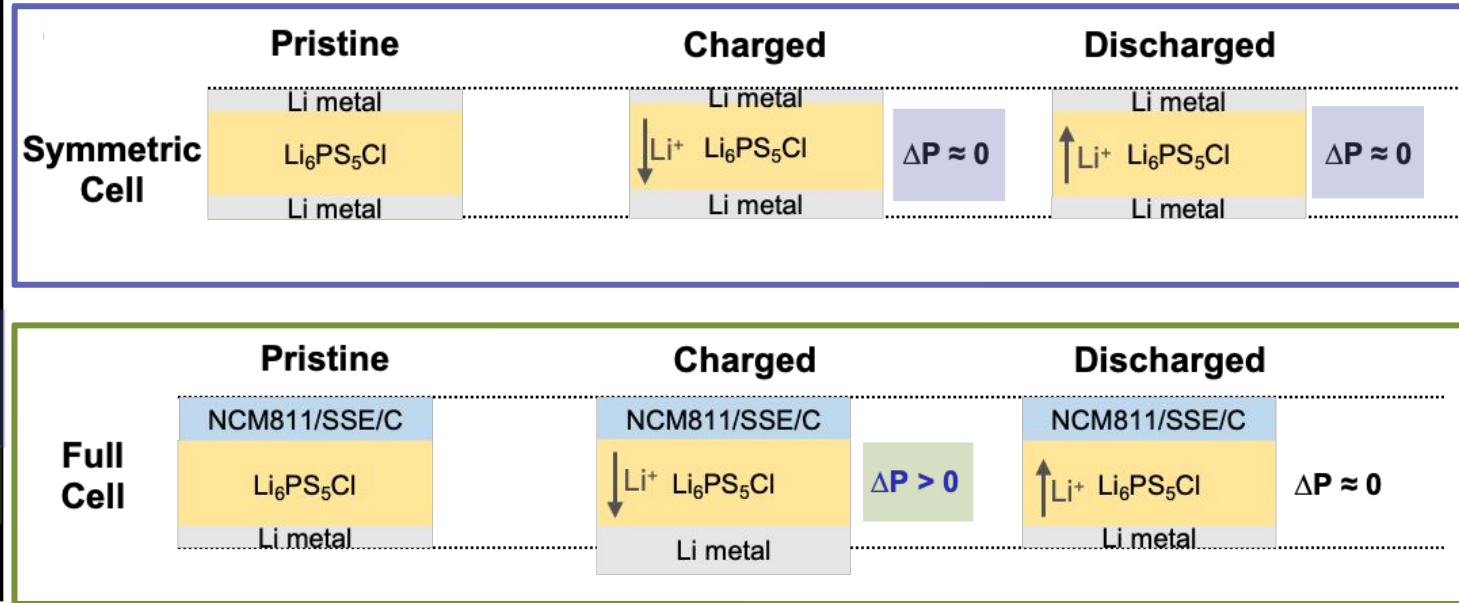
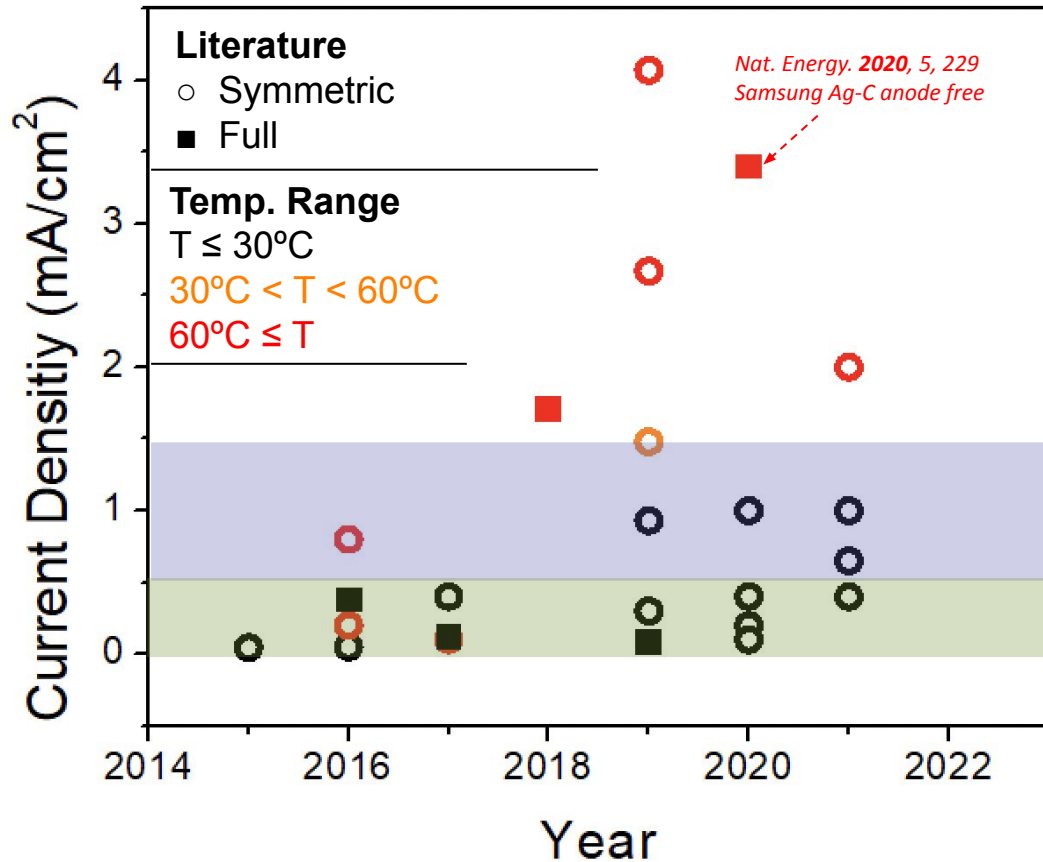
Polymer based ones are already there!

We are going for inorganic solids 😊

Stack pressure effect on Li metal anode



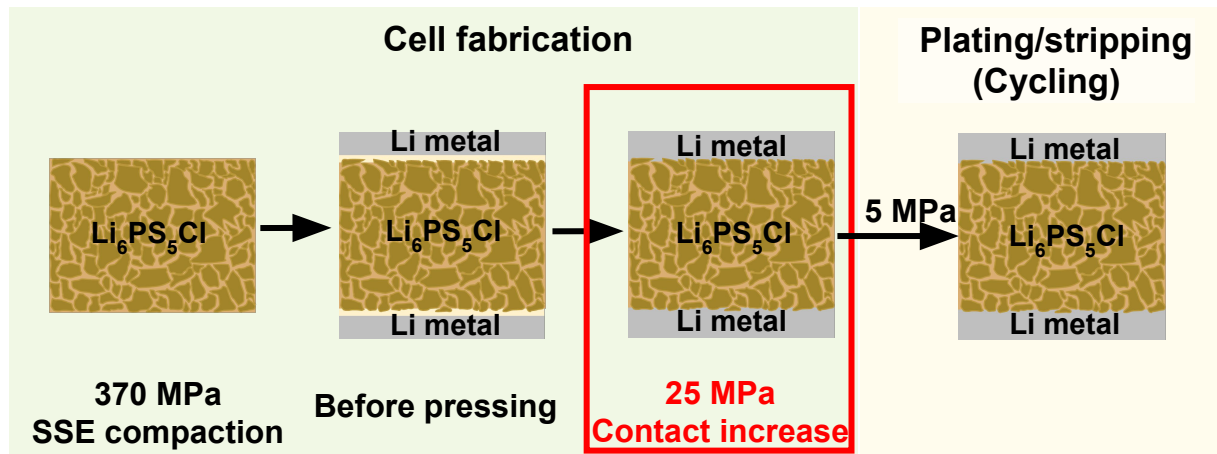
Reported Critical Current Densities of Li Metal ASSB



- Critical current density: Symmetric > Full
- Near room temperature full cell: < 1 mA/cm²
- Pressure change: Symmetric < Full

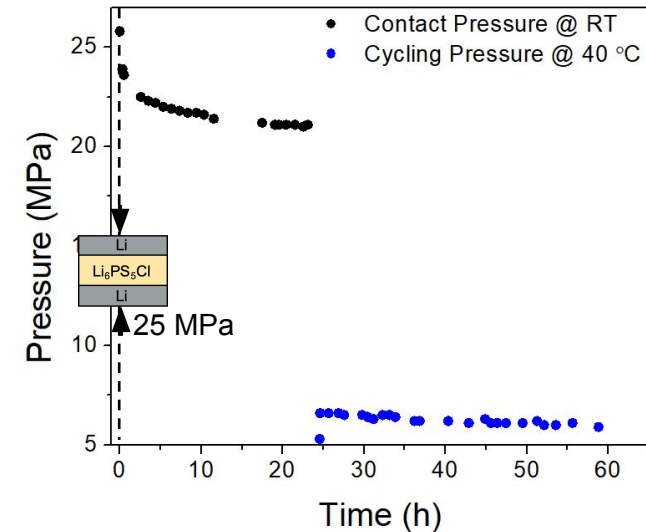
Li Metal Symmetric Cell: Cell Fabrication/Contact

Li Metal Symmetric Cell Fabrication & Cycling Process



- Three different pressures applied during the fabrication/cycling process
 1. SSE compaction pressure = 370 MPa
 2. Contact pressure = 25 MPa
 3. Cycling pressure = 5 MPa

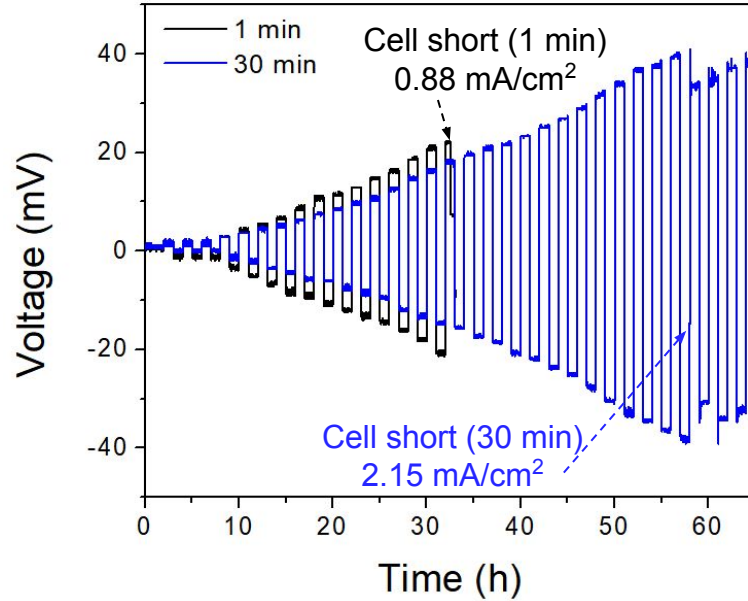
Pressure Monitoring of Li Metal Symmetric Cell



- Contact Pressure:
 - Rapid drop during initial 30 min
 - Gradual decrease afterward
- Cycling (plating/stripping):
 - No significant pressure change

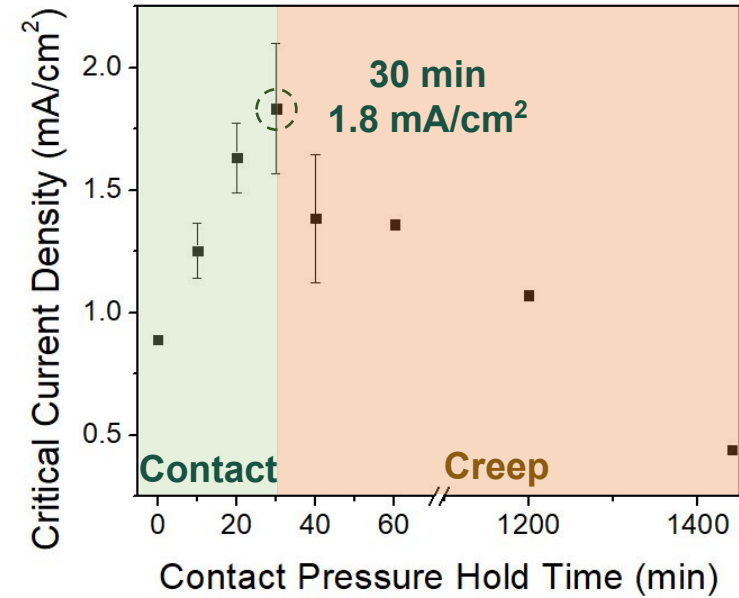
Li Metal Symmetric cell: CCD/Failure

Ramping Test of Different Contact Time Cells



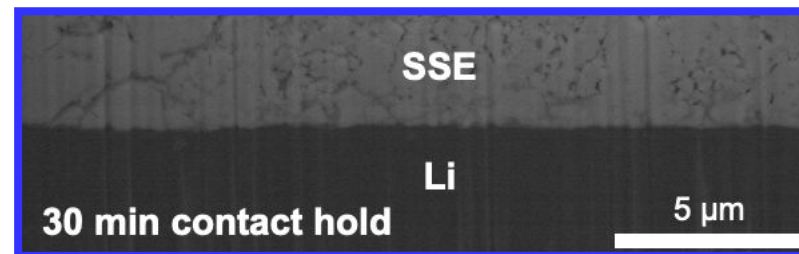
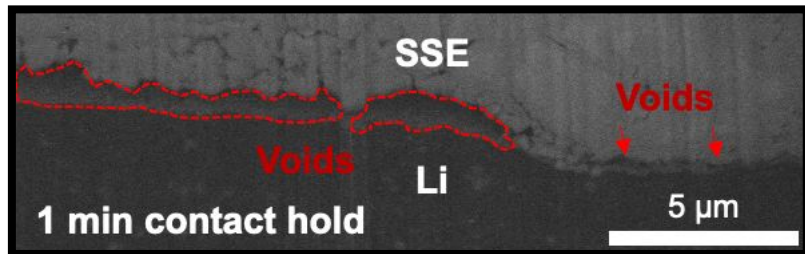
- Ramping test for CCDs of symmetric cells
- Higher CCD in 30 min contact hold sample

Ramping Test of Different Contact Time Cells

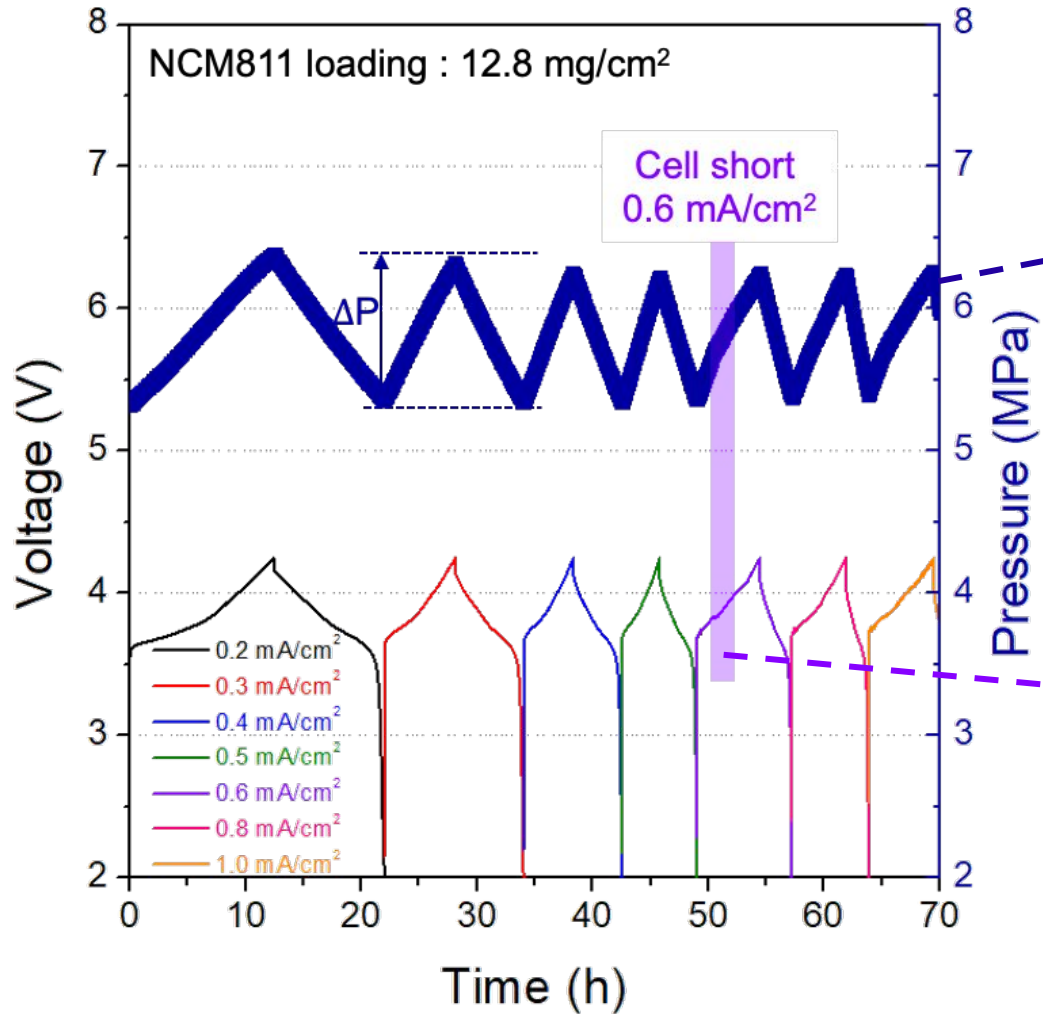


- CCD trends depending on contact hold time
- CCD increase until 30 min contact, decrease afterwards

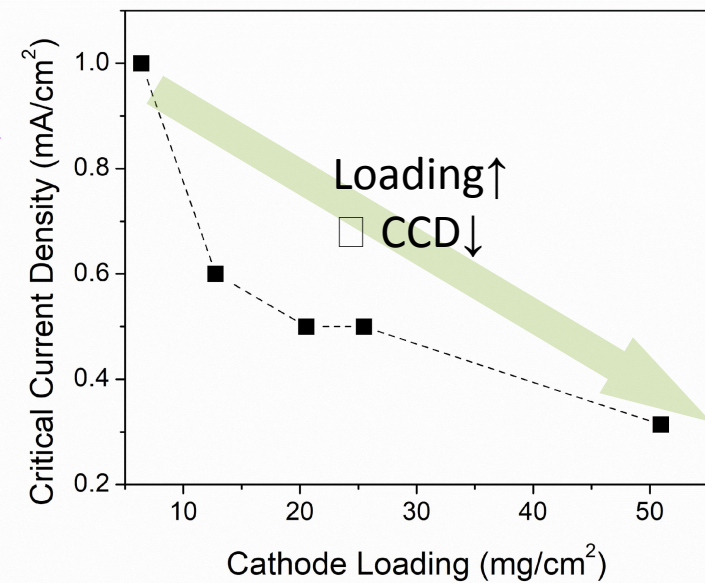
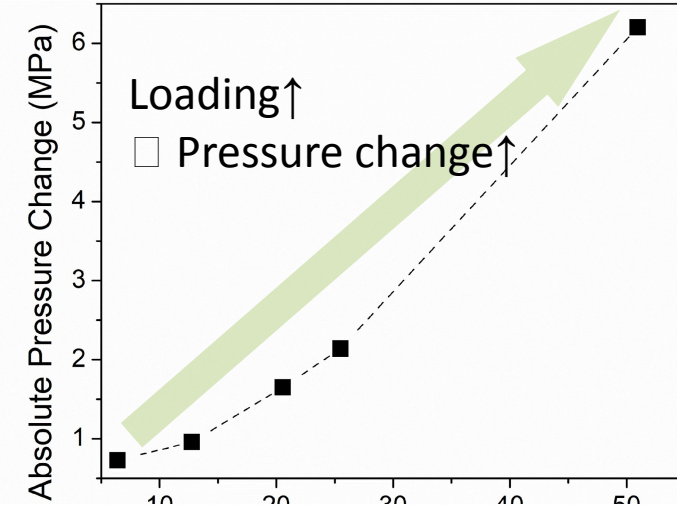
Cryo-FIB/SEM : Direct observation of Li/SSE interface



Li Metal Full Cell: Pressure Dependence



- Ramping test conducted to evaluate CCDs of full cell
- Full cell: Pressure change during cycling with lower CCD



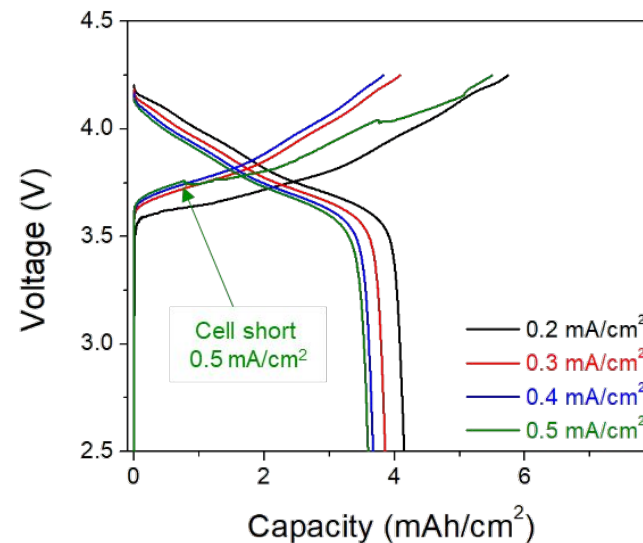
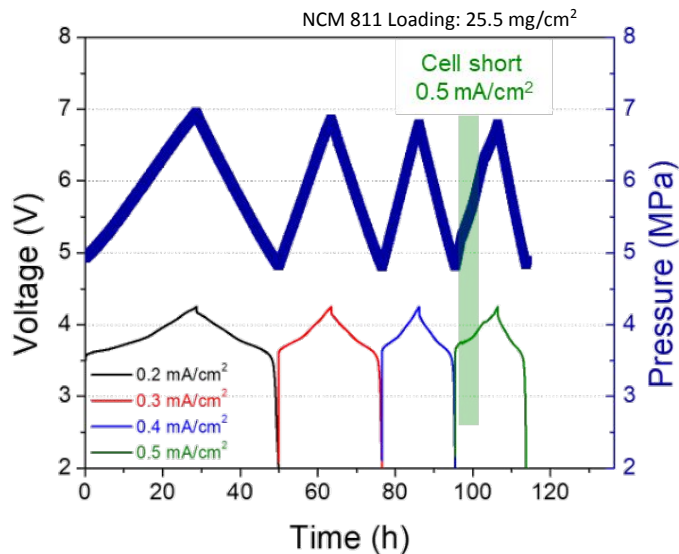
Full Cell: Fixed Gap vs. Constant Pressure



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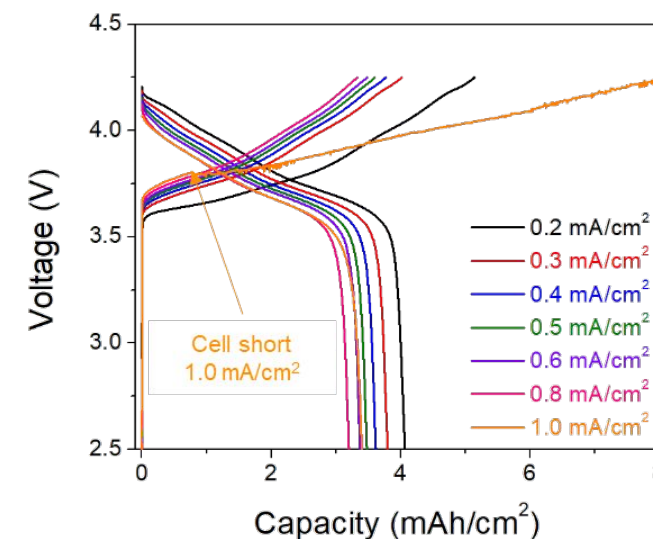
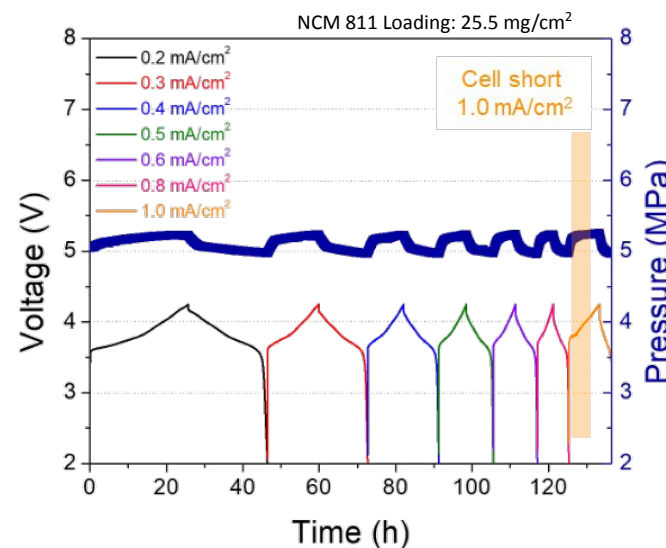
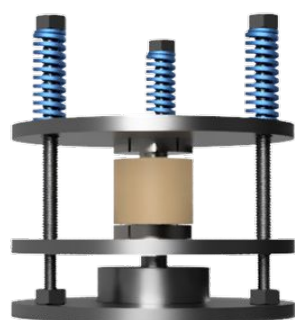
Fixed gap



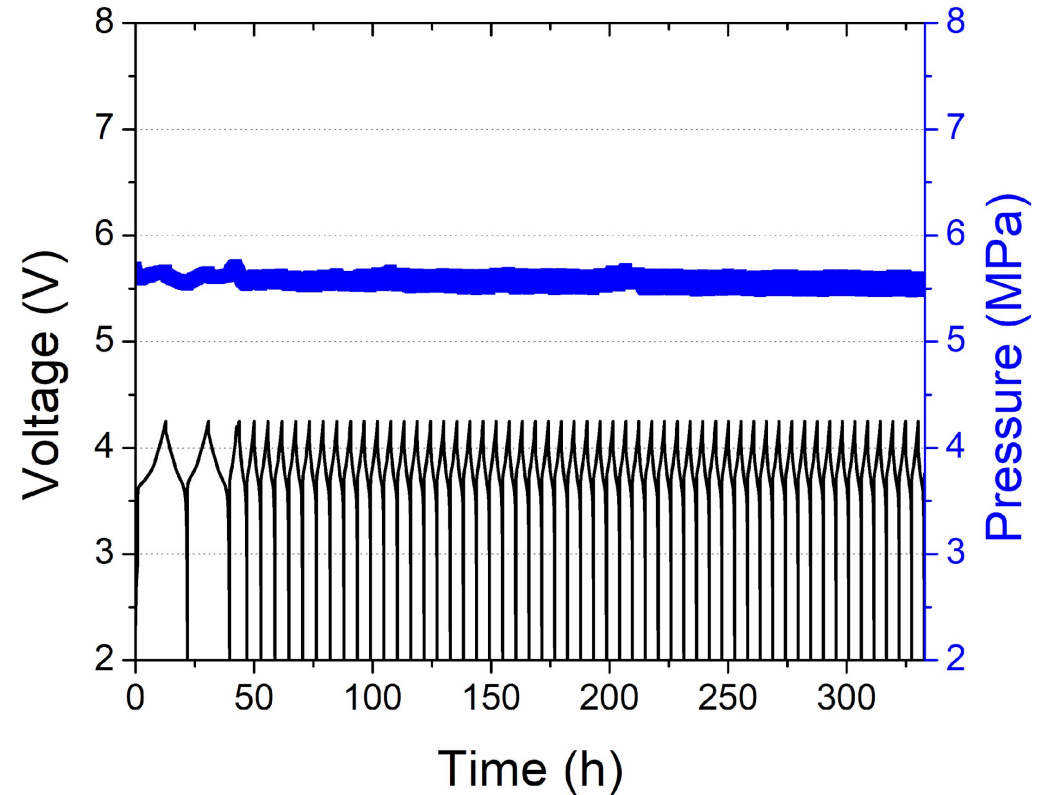
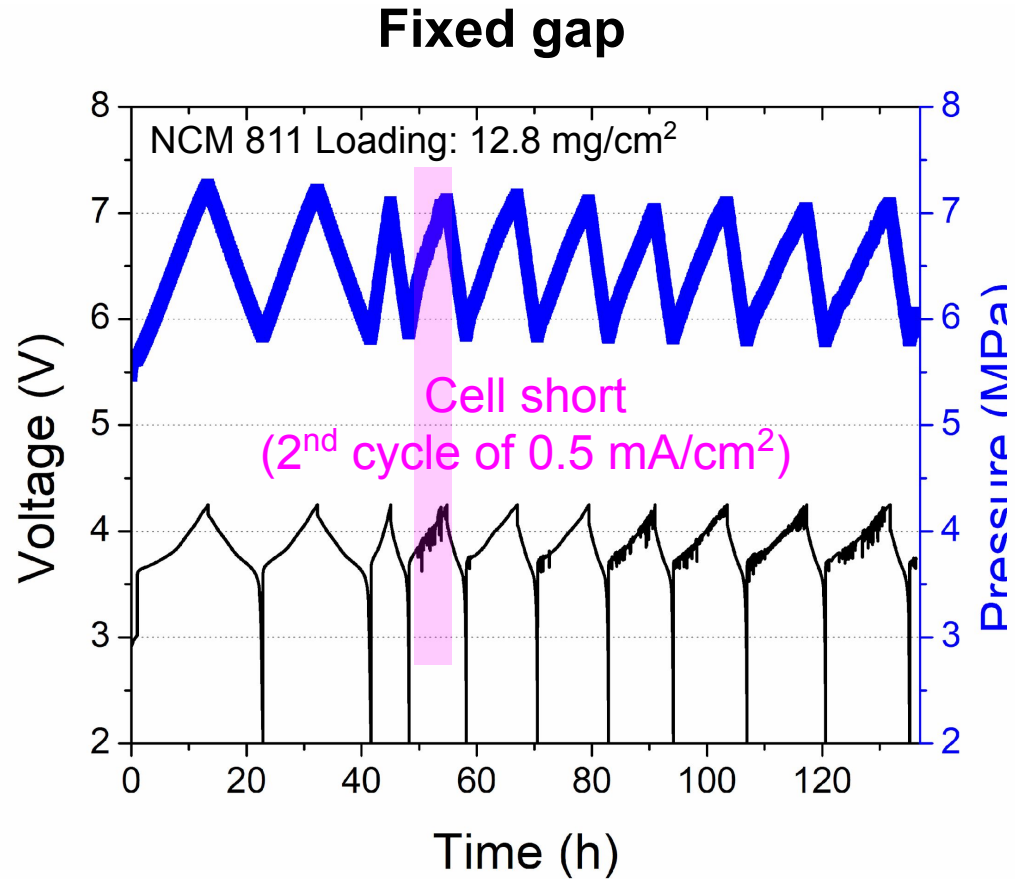
**Lower pressure change
and higher CCD
in const pressure setup**

**Constant
Pressure Set-up**
for the volume change
compensation

Constant pressure



Long-term Cycling of Constant Pressure Setup

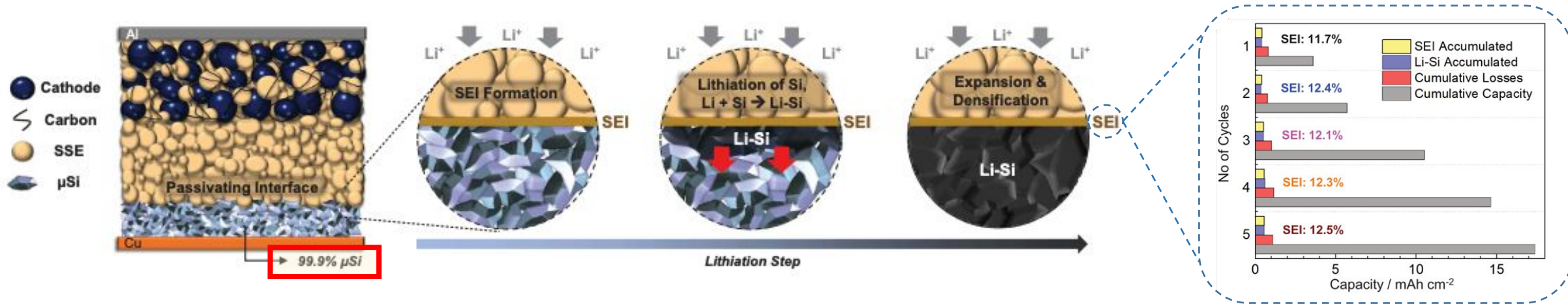


- 0.5 mA/cm^2 Long term cycling after two activation cycles
- Fixed gap: Shorted at 2nd cycle at 0.5 mA/cm^2
- Fixed gap: Cycled more than 50th cycle at 0.5 mA/cm^2

All-Solid-State-Li Metal Batteries:

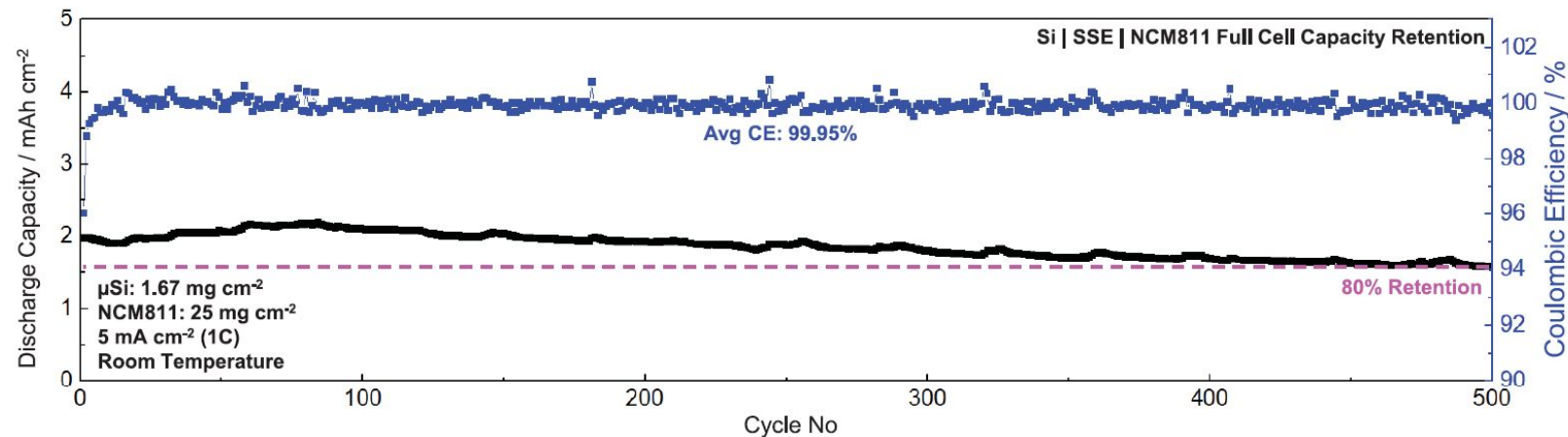
- Elucidate CCD discrepancy between Li symmetric and full cell
- Li symmetric cell
 - Better contact enabled higher CCD
- Li full cell
 - Pressure change during cycling induce cell shorting
 - Constant pressure setup: Mitigate cell shorting

Si Anode Synergy in Solid-State Batteries

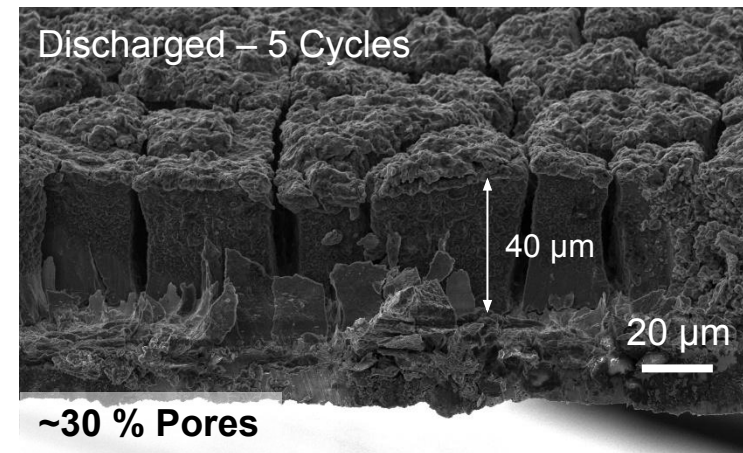
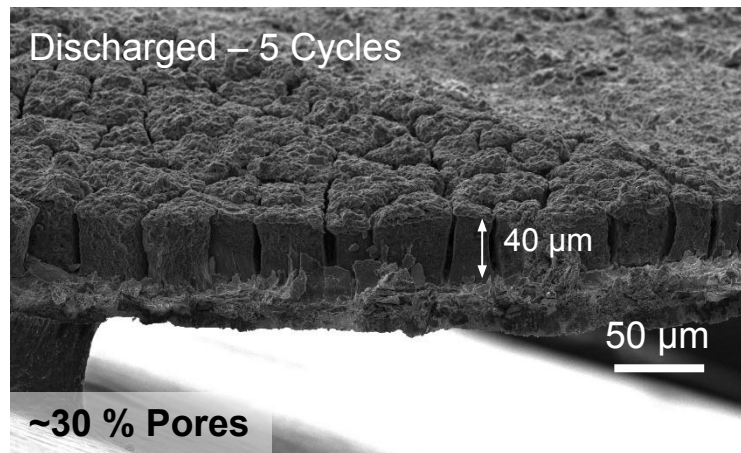
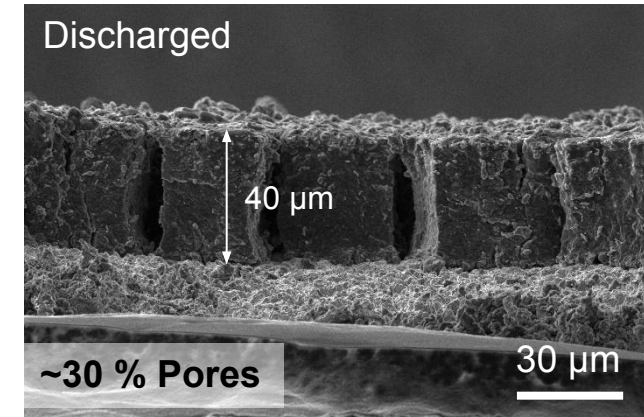
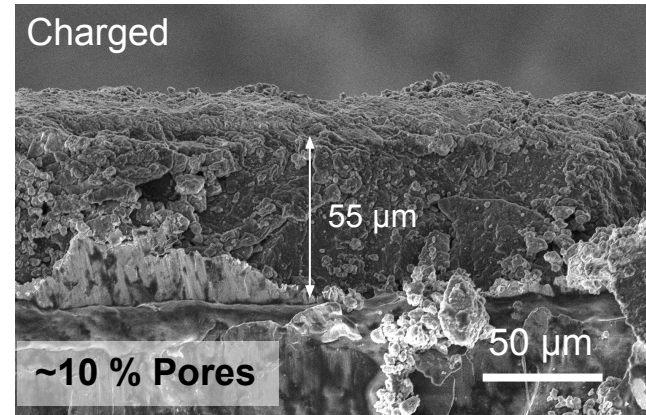
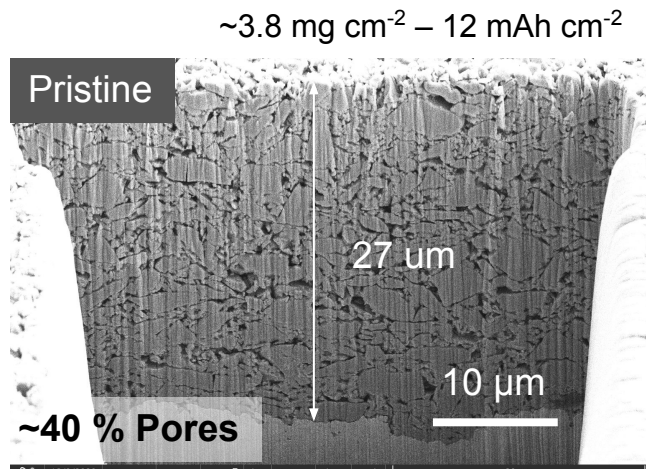


- Enable **99.9%** Si anode without carbon and solid electrolyte
- Inventory loss to the passivating SEI remained relatively constant
- Realized Si cycling >500 cycles

However, it is paramount to *improve the initial Coulombic efficiency (~76%)* to achieve high energy density all-solid-state batteries



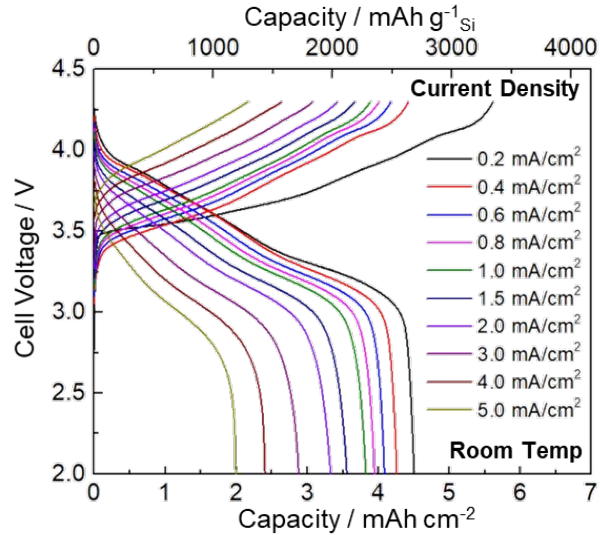
Porosity changes during cycling



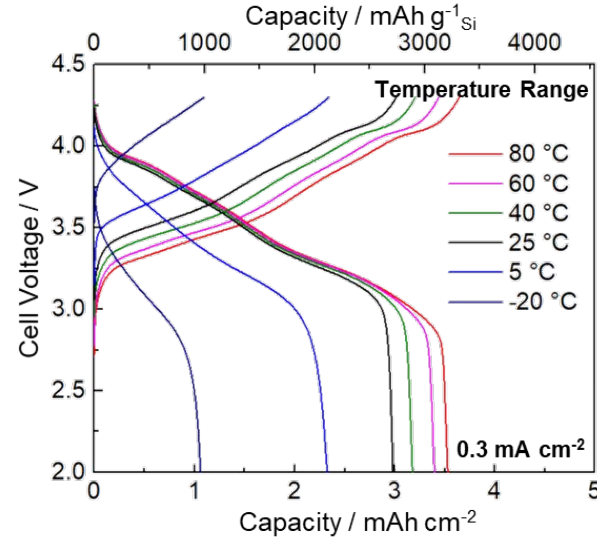
Electrochemical performance



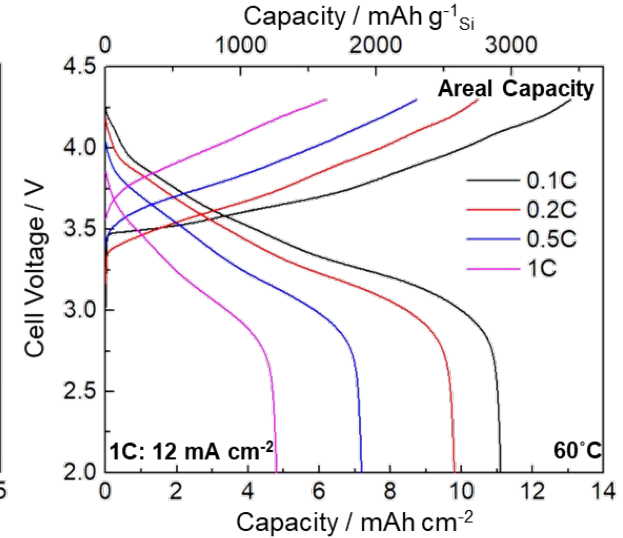
Dr. Darren Tan



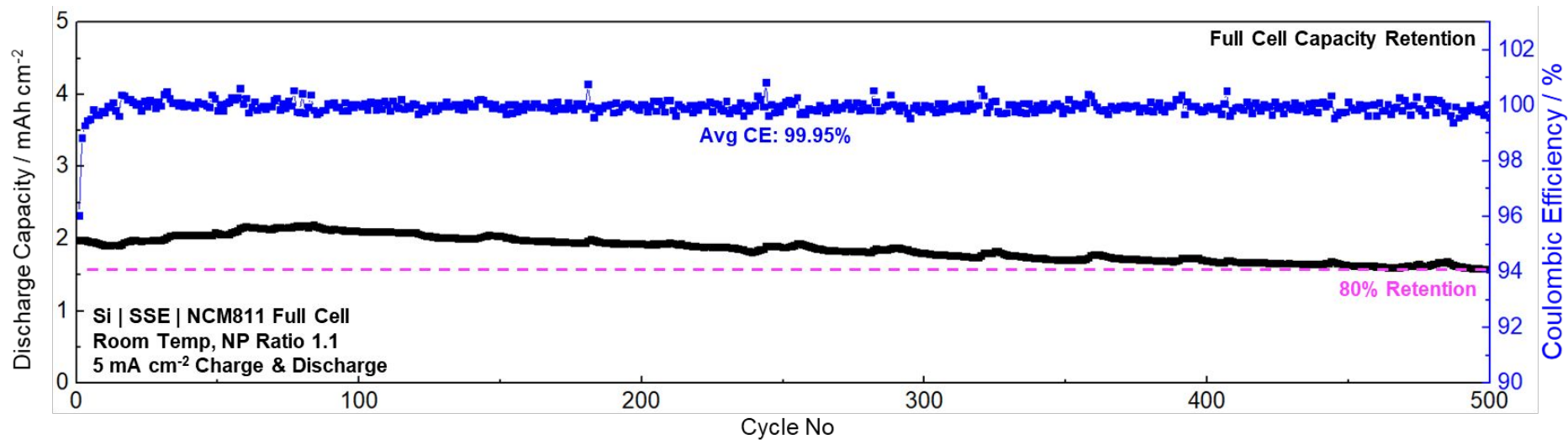
High Current Density



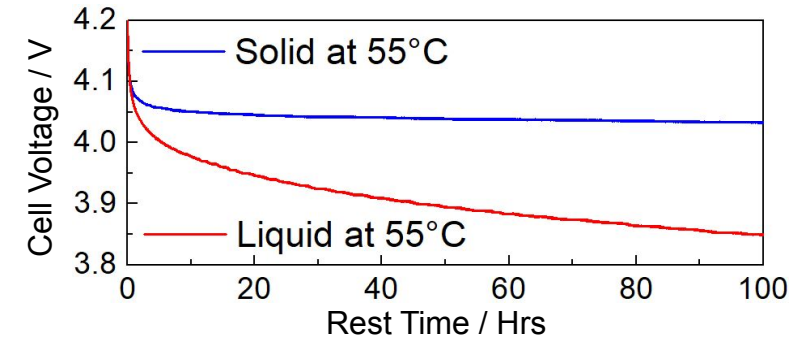
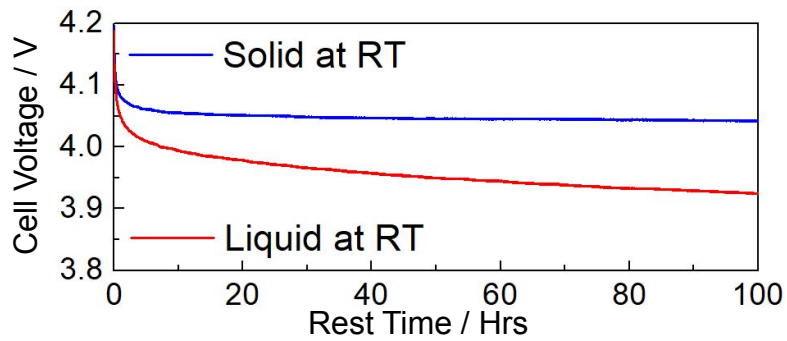
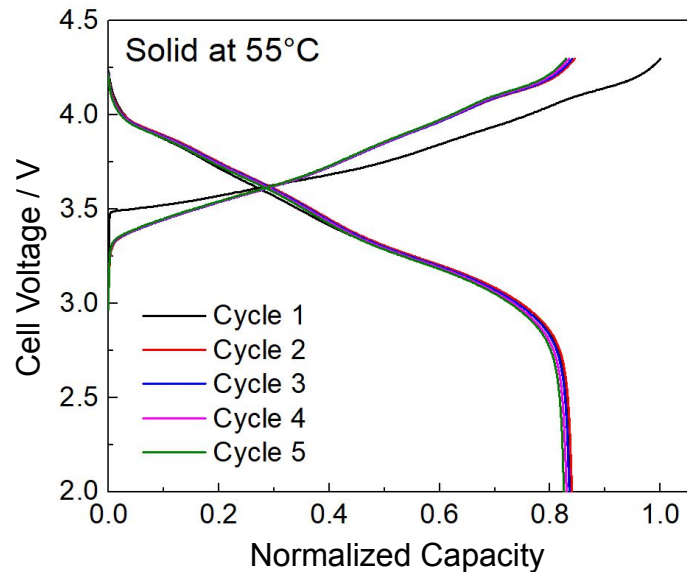
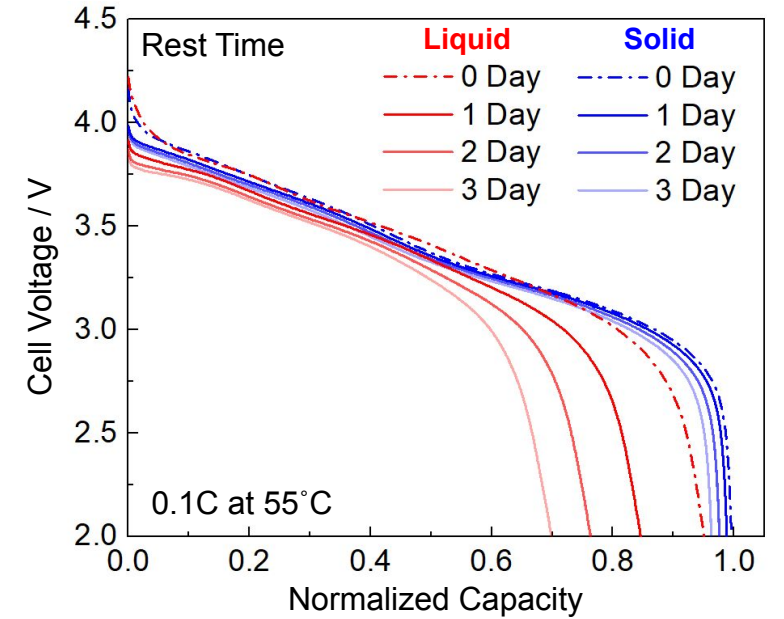
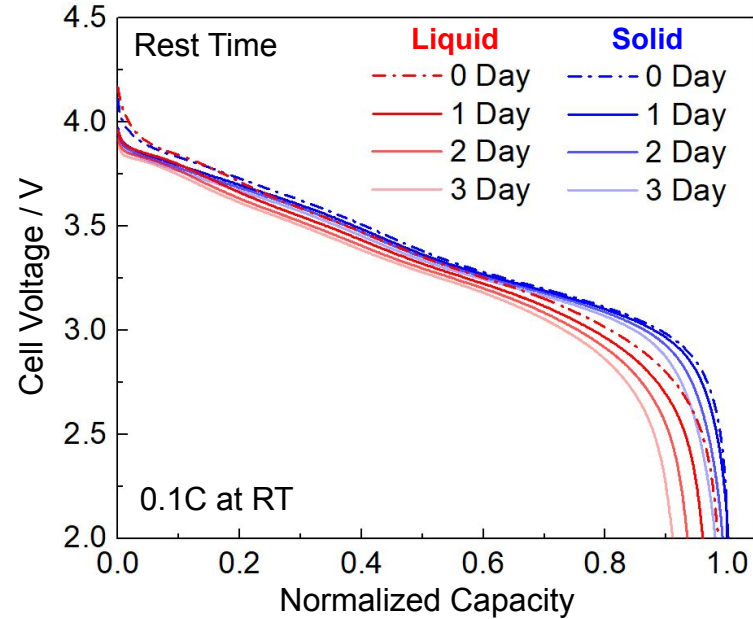
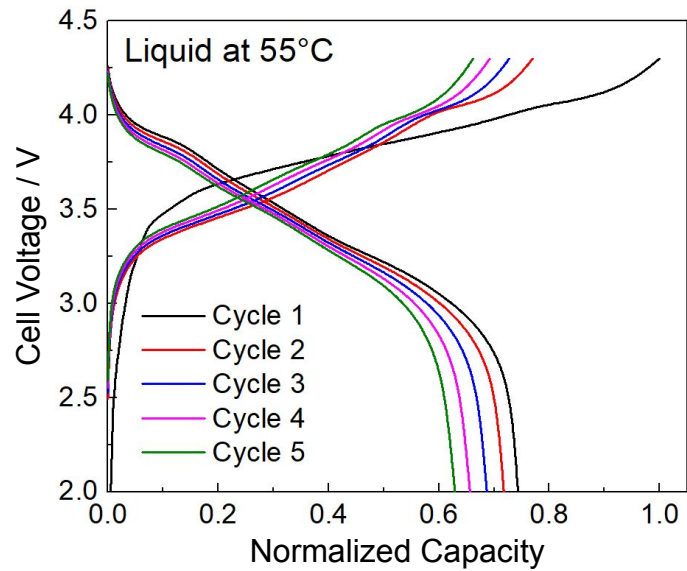
Wide Temperature Range



High Loading



Passivating Interfaces – Extremely Stable

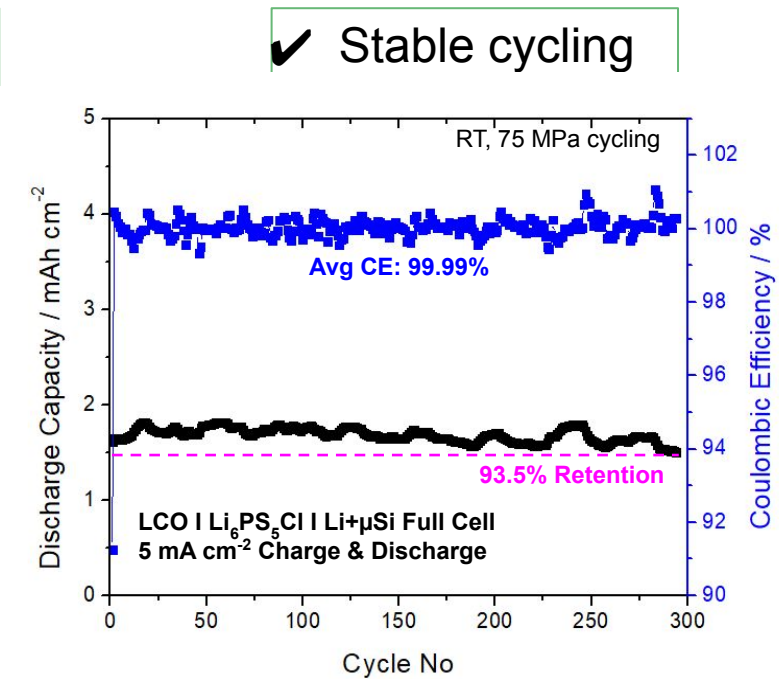
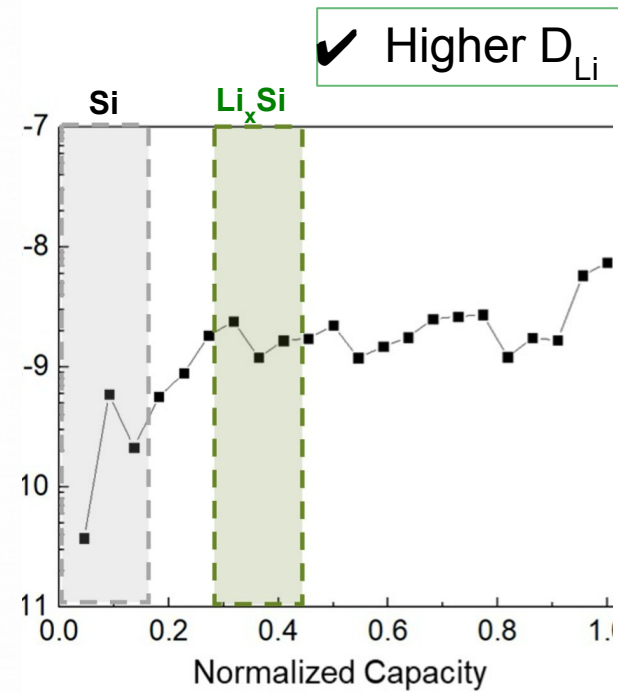
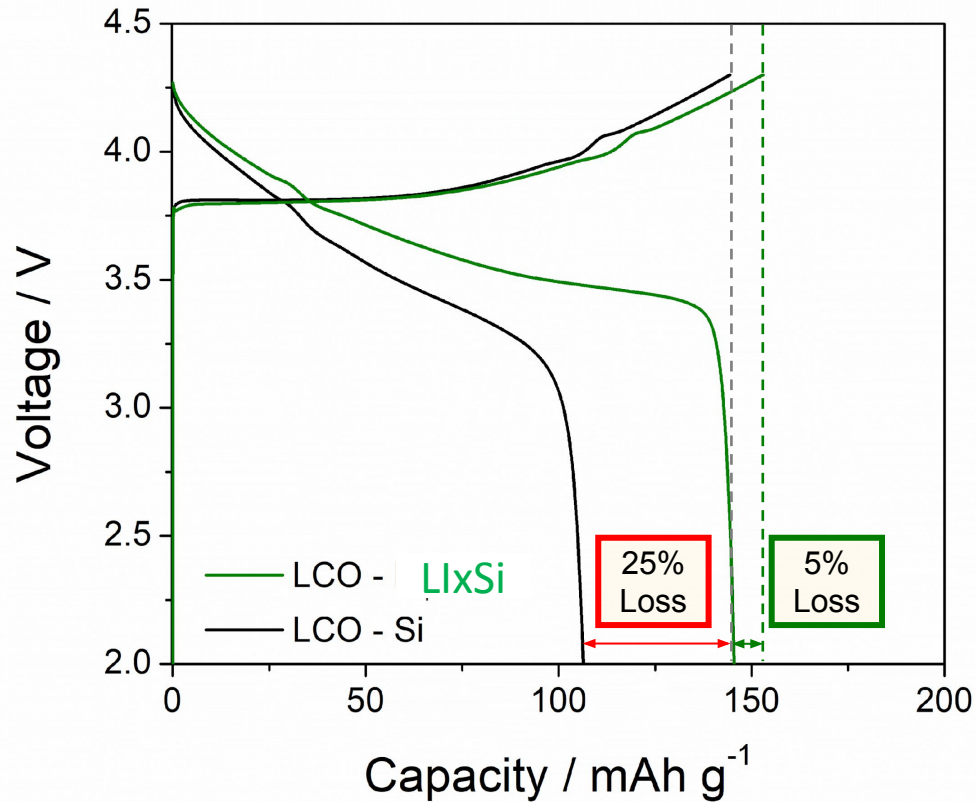


LG FRL - Anode Strategies

1st Year Achievement

- Enhanced ICE
- Higher D_{Li} and stable cycling

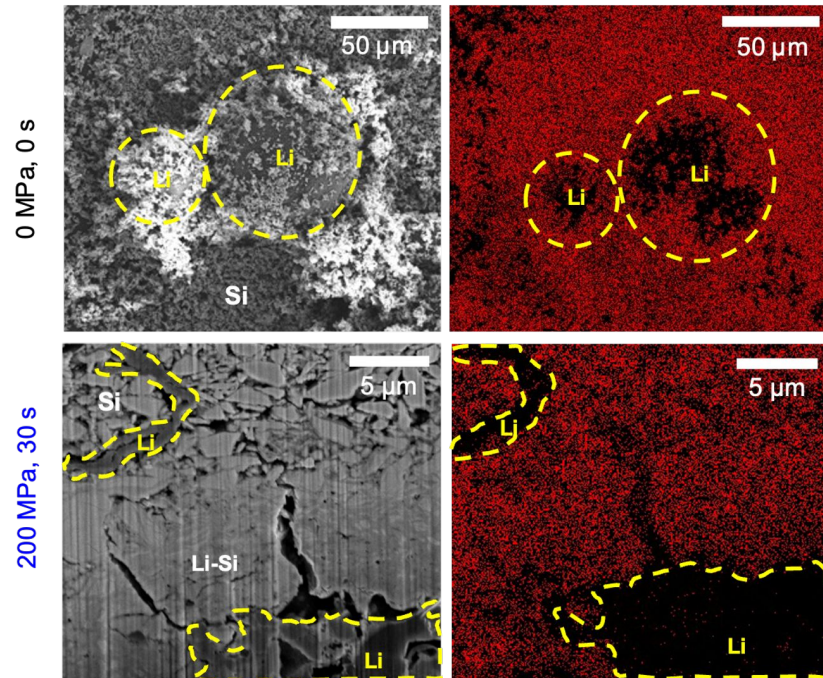
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Prelithiation of Micro-Silicon via Pressurization

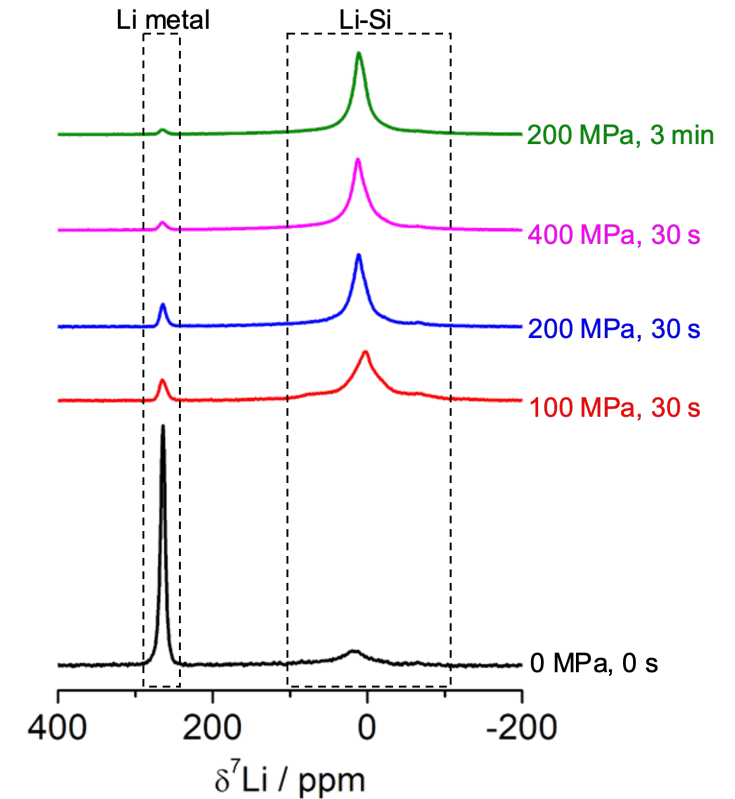
Morphology of Li_1Si before and after pressurization



- **Li-Si formed after pressurization**
- Before pressurization: Preserved morphology of sphere stabilized lithium metal powder (SLMP) and micro-Si
- After pressurization: Li metal deformed and Li-Si phase
Nature Communications, under review (2023)

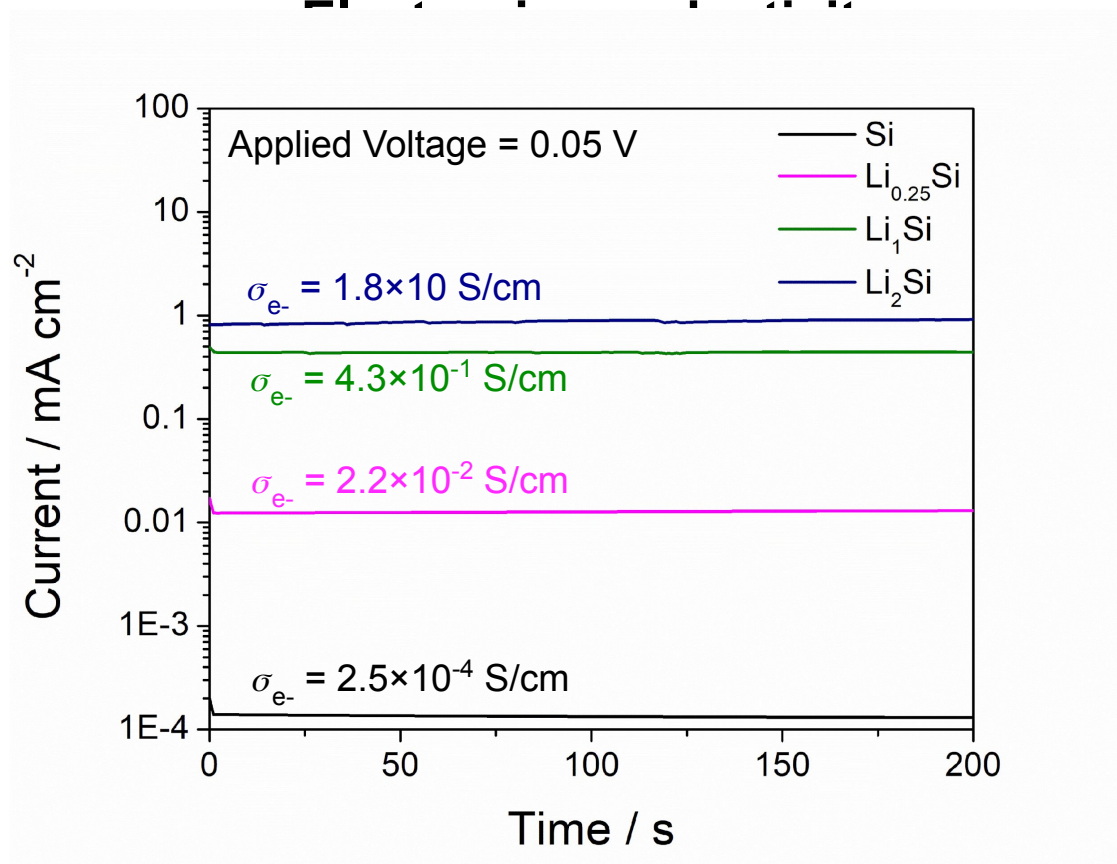
^7Li NMR of Li_1Si

before and after pressurization

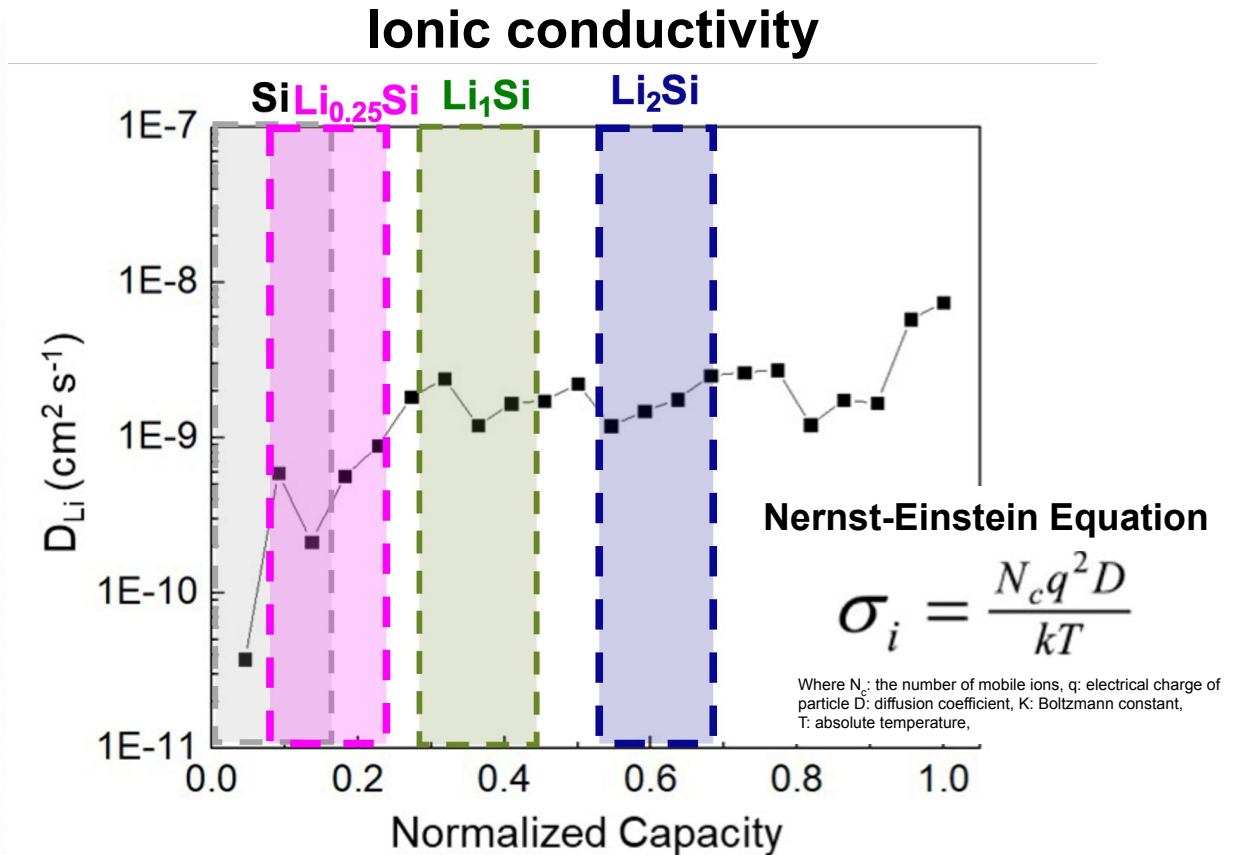


- Higher Li metal peak in unpressed (0 MPa, 0 s)
- **Longer pressurization time/higher pressure drives more Li incorporation**
- Pressurization converts nearly all Li metal to Li-Si

Physical Properties of Pure Si *and* Li_xSi



- **Higher Electronic conductivity:**
Increase by orders of magnitude with more Li content

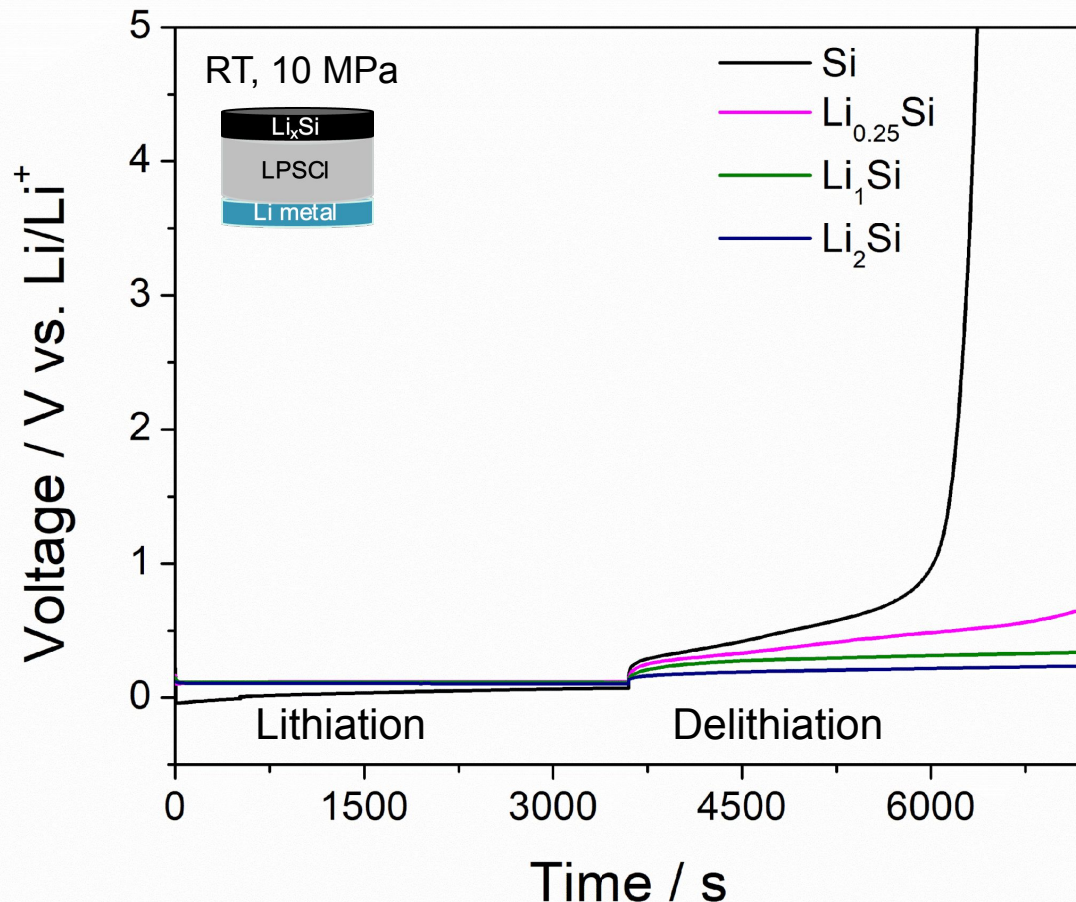


- Li^+ diffusivity in Si : Higher in higher SOC
- Prelithiation \square Faster Li^+ diffusion in Si
- **Higher ionic conductivity** increased by prelithiation

Li_xSi in Half- and Full-Cell

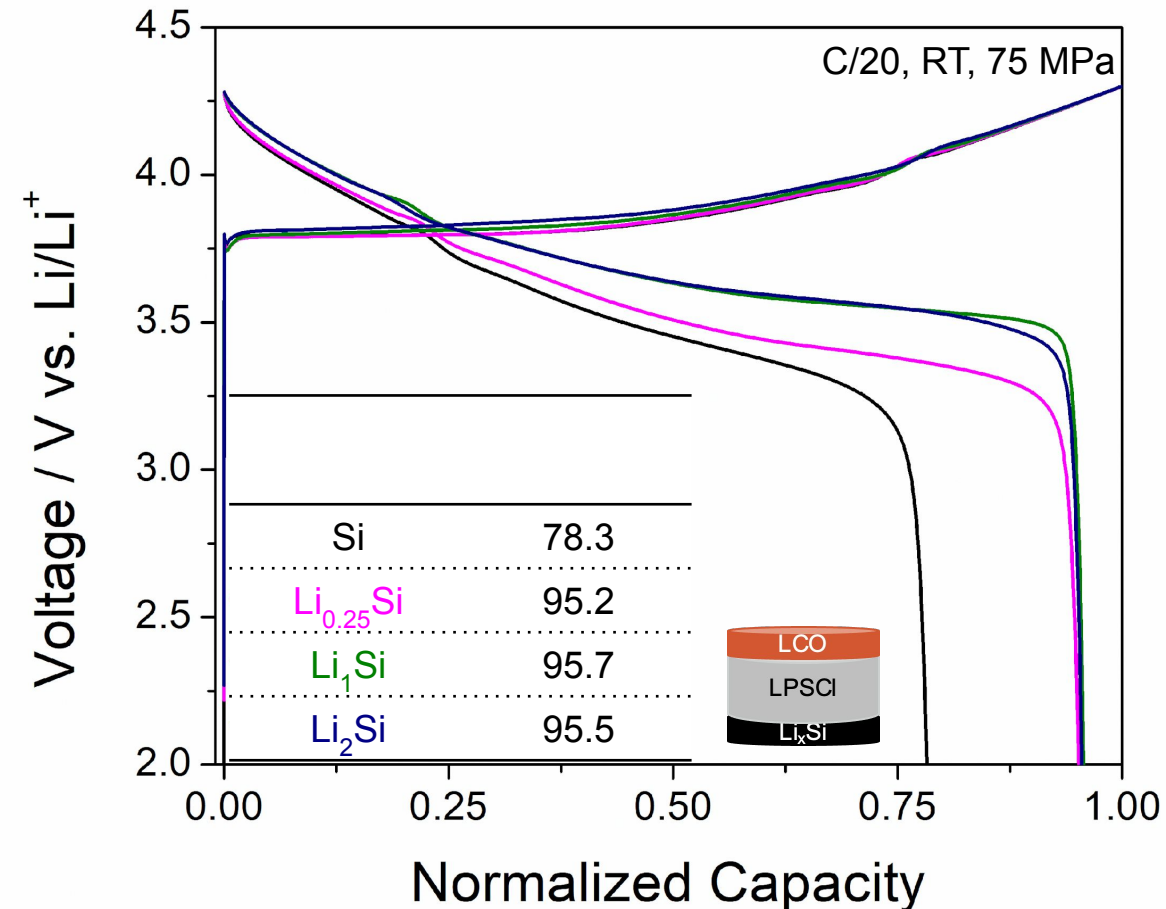
Nature Communications, under review (2023)

Half-Cell



- Lithiation: Similar overpotential
- Delithiation: Higher overpotential in less lithiated Li_xSi
- **Harder to delithiate Li_xSi depending on how much Li is in Si**

Full-Cell

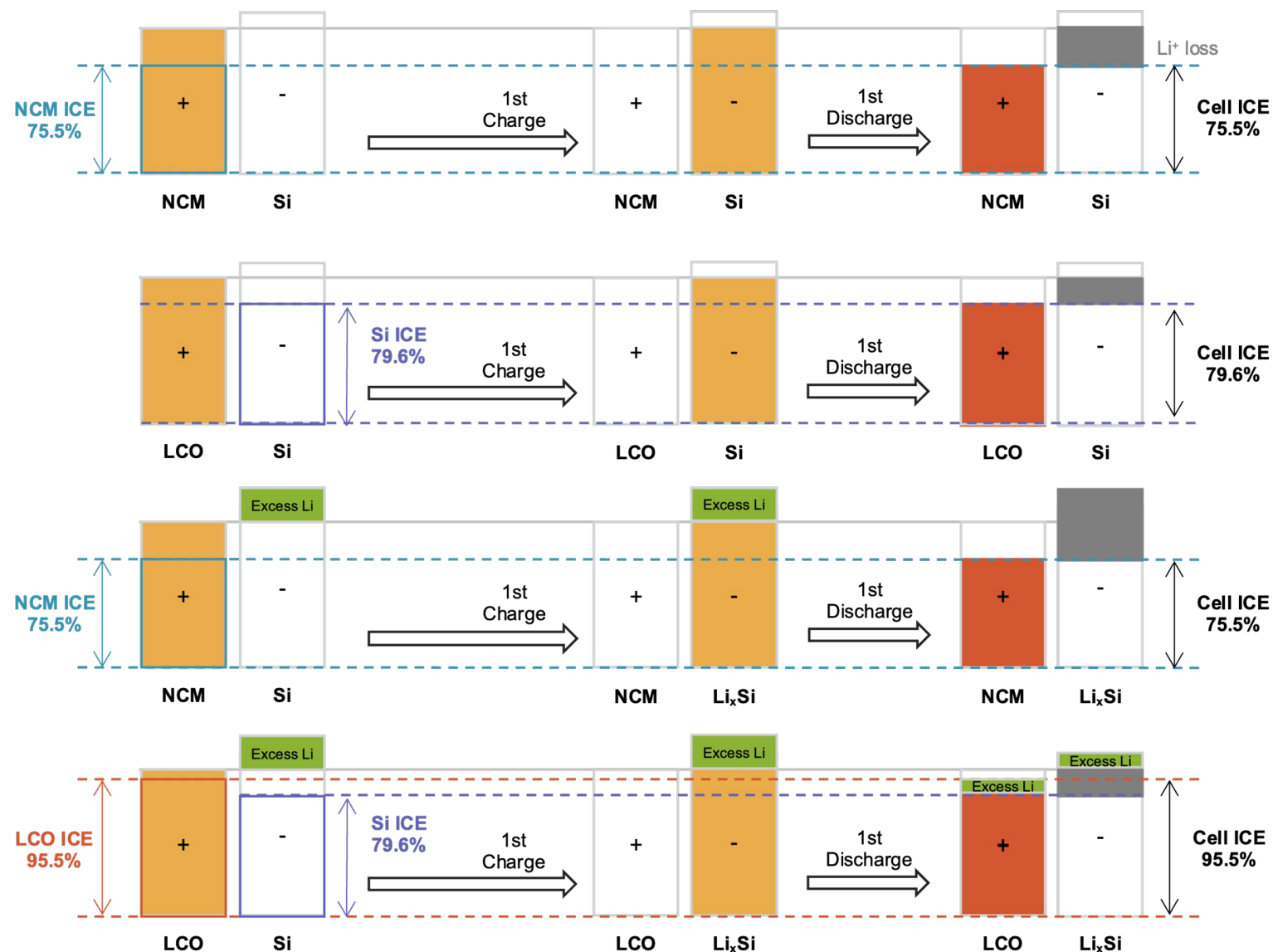


- Improved ICE in all lithiated Si, but converged to 95.5% after Li₁Si
- Li₁Si for remaining cell cycling experiments

Prelithiation Driven Improvement: NCM vs LCO

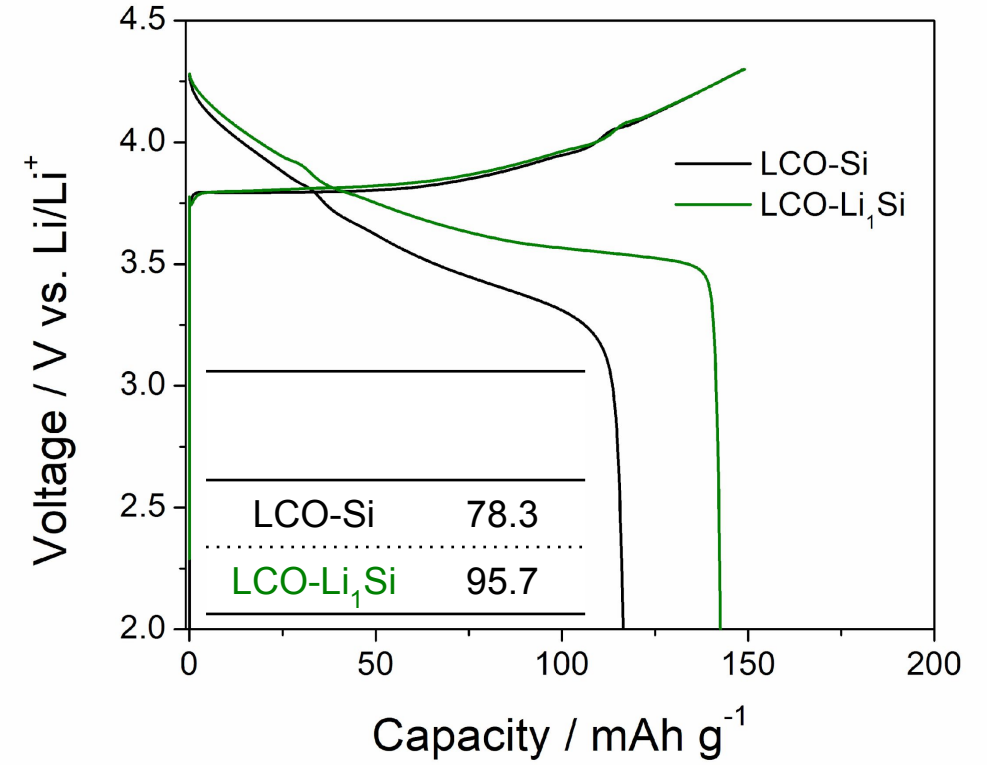
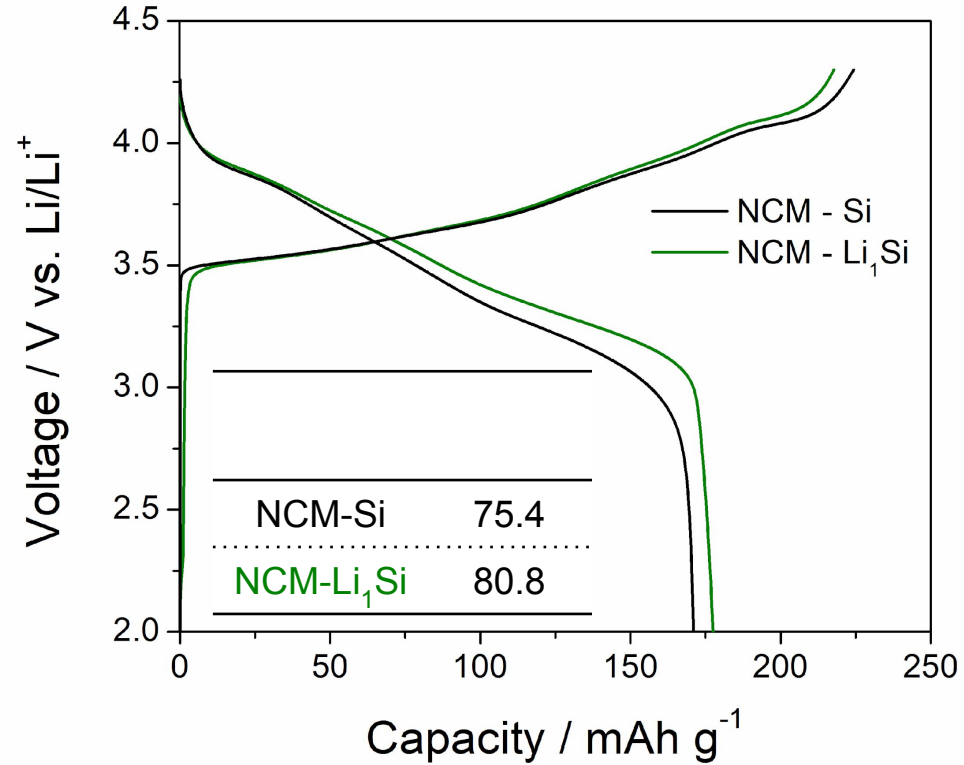


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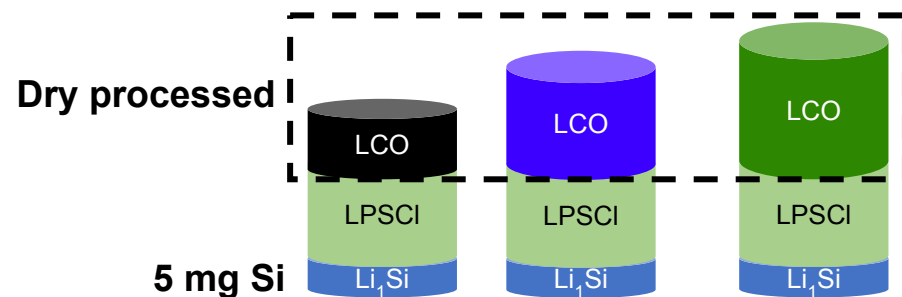
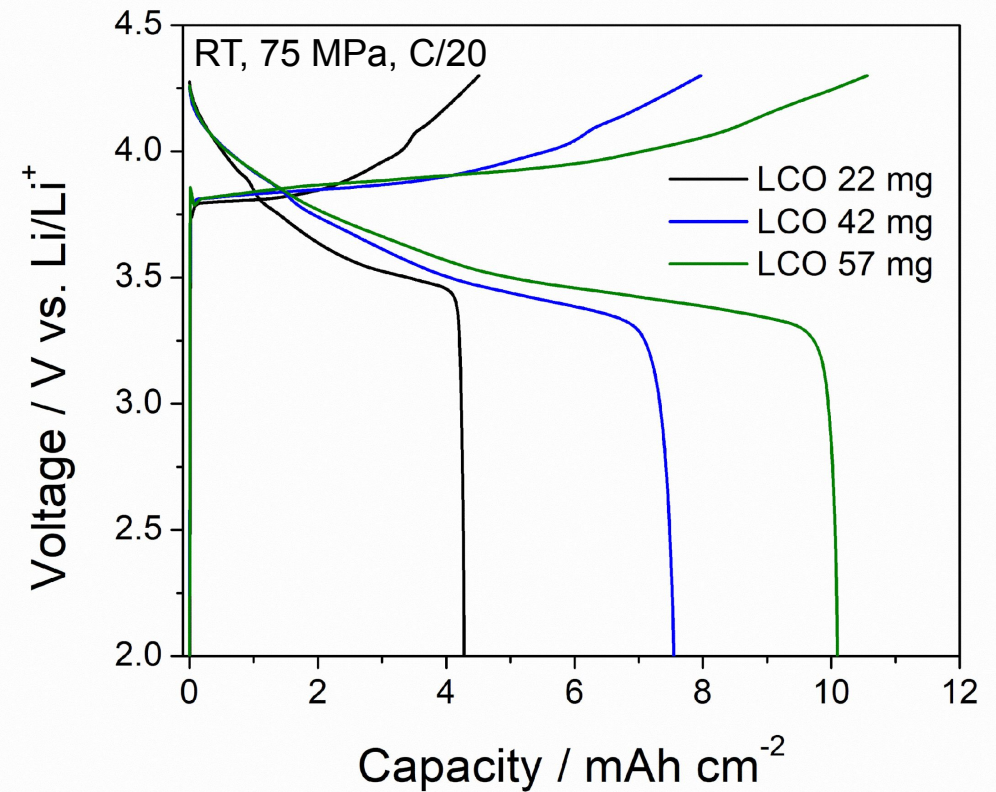
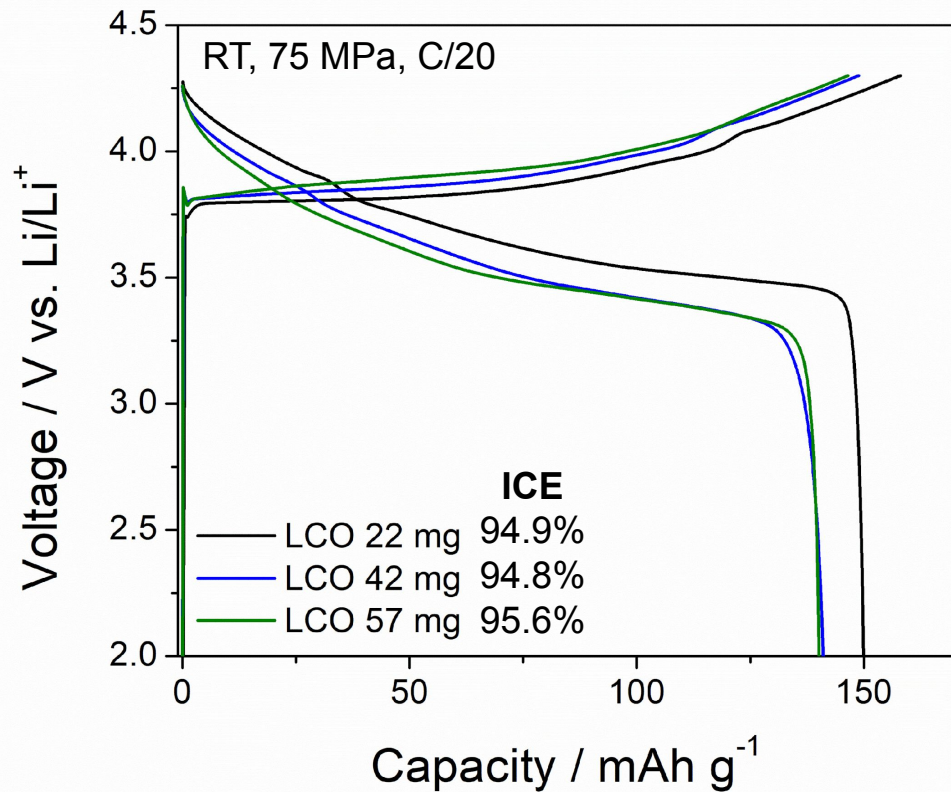
- Based on ICE of NCM, LCO, and Si (in the half-cell)
- Even with preithiation, NCM-LiSi ICE will be still limited by NCM ICE limit **“Cathode Limit”**
- LCO-LiSi case: Higher ICE expected up to LCO ICE limit **“Anode Limit”**

Prelithiation Driven Improvement: NCM vs LCO



- With prelithiation, only LCO cell showed large ICE improvement
- NCM: limited by NCM ICE □ No further improvement from prelithiation
- **Prelithiation of anode is effective for “anode-limiting”**

Dry processed LCO – Li_1Si : High Loading Viability



- Similar ICE for all loadings
- Higher polarization for high loading but still achieved 140 mAh/g discharge capacity with areal capacity of more than 10 mAh cm^{-2} discharge capacity
- **Li_1Si (5 mg Si) could accommodate 10 mAh cm^{-2} capacity**

Remaining Challenges

Precursors

Li₂S price needs to come down by 5X -10X
SSE particle size control must be done

Processibility

Dry room compatibility - yes!
Dry processing – at scale!!!

Pressure reduction from 100MPa – 50MPa – 5MPa
Making SSB structural component

Pressure

Solid State Fire ???!

Thermal Runaway Behavior of $\text{Li}_6\text{PS}_5\text{Cl}$ Solid Electrolytes for $\text{LiNi}_{0.8}\text{Co}_{0.1}\text{Mn}_{0.1}\text{O}_2$ and LiFePO_4 in All-Solid-State Batteries

Taehun Kim,[‡] Kanghyeon Kim,[‡] Seonghyun Lee, Gawon Song, Min Soo Jung, and Kyu Tae Lee*

Cite This: *Chem. Mater.* 2022, 34, 9159–9171

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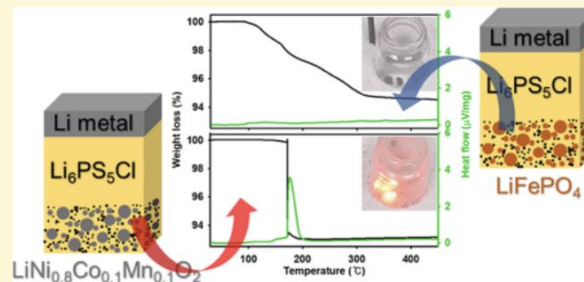
ACCESS |

Metrics & More

Article Recommendations

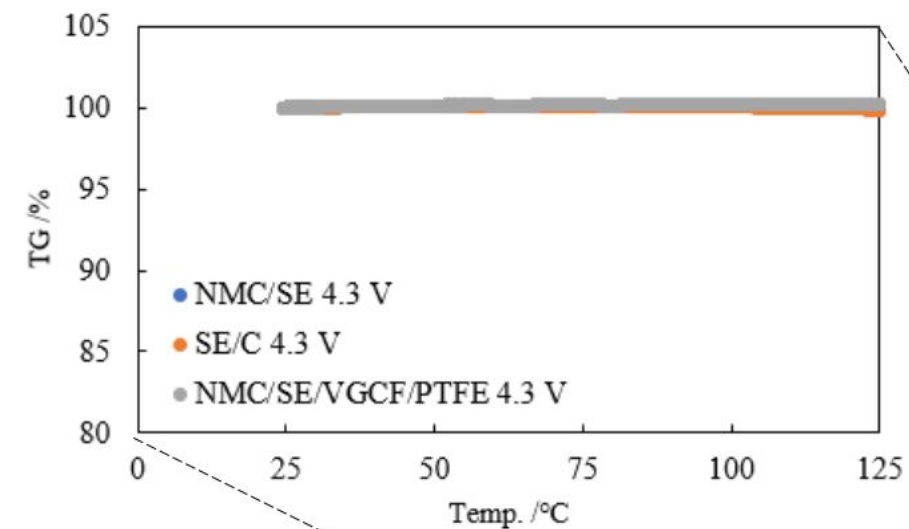
Supporting Information

ABSTRACT: All-solid-state batteries (ASSBs) have received much attention because of their high energy density and safety. However, the safety of argyrodite-type $\text{Li}_6\text{PS}_5\text{Cl}$ (LPSCI)-based ASSBs is still not assured because their thermal stability has been assessed under selected mild conditions. Herein, we introduce the poor thermal stability of LPSCI with Ni-rich layered oxide cathode materials as the trigger of thermal runaway. The charged composite cathode pellets containing $\text{Li}_{1-x}\text{Ni}_{0.8}\text{Co}_{0.1}\text{Mn}_{0.1}\text{O}_2$ and LPSCI are explosively burned at 150 °C even in Ar. Moreover, the mechanical abuse gives rise to violent burning at room temperature. This is due to vigorous exothermic chemical reactions between delithiated $\text{Li}_{1-x}\text{Ni}_{0.8}\text{Co}_{0.1}\text{Mn}_{0.1}\text{O}_2$ and LPSCI. However, LPSCI with LiFePO_4 exhibits excellent thermal stability, such as no violent exothermic reactions even at 350 °C. This is because LPSCI is metastable with delithiated $\text{Li}_{1-x}\text{FePO}_4$. Moreover, LiFePO_4 shows excellent electrochemical performance, such as a high reversible capacity of 141 mAh g⁻¹ and stable capacity retention over 1000 cycles, despite the fact that LiFePO_4 is known to be poorly electrochemically active for ASSBs. These findings provide fundamental insights to improve the thermal stability and electrochemical performance of LPSCI-based ASSBs.



As a summary, I believe this paper is creating very specific scenarios to generate a NCM fire, and misinterpreting the cause as the SSE, which is not related to the ignition in the first place.

Dr. Darren Tan – CEO of UNIGRID



Data from work UC San Diego

LGES UCSD – FRL Team Members



Ong

Kim

Shpyrko

Clement

Meng

Liu

Chen

**Computation
Modeling**

**Characterization
Novel Materials**

**Scalable
Processing**

**Devices
Prototyping**

**Recycling
Safety**



LGES – UCSD Frontier Research Laboratory

Acknowledgements First



U.S. DEPARTMENT OF
ENERGY

Office of Science

ThermoFisher
SCIENTIFIC



Battery Prototyping



 **LG Energy Solution**
LGES-UCSD Frontier Research Laboratory

DOE BES 2012-now (Dr. Jane Zhu)
LiPON SSB and Perovskite SC and
Memristive

Workflow design for battery
Next-gen Cryo EM for Energy and
Quantum materials
Falcon Camera etc.



Solid State Battery Team at my group